

**NUTRIENT ELEMENTS AND HR-LCMS ANALYSIS OF
SECONDARY METABOLITES PRESENT IN
METHANOLIC EXTRACTS OF *Colocasia esculenta* (L.)
SCHOTT (cocoyam)**



**A THESIS SUBMITTED TO THE
DEPARTMENT OF CHEMISTRY
BIRENDRA MULTIPLE CAMPUS
INSTITUTE OF SCIENCE AND TECHNOLOGY
TRIBHUVAN UNIVERSITY
NEPAL**

**FOR THE PARTIAL FULFILMENT OF THE REQUIREMENTS FOR THE
MASTER OF SCIENCE DEGREE
IN CHEMISTRY**

**BY
AJAYA MAHATO
ROLL NO: CHE 2199/077
TU REGISTRATION NO: 5-2-19-577-2011**

FEBRUARY, 2025

DECLARATION

I declare that this dissertation entitled “**Nutrient Elements and HR-LCMS Analysis of Secondary Metabolites Present in Methanolic Extracts of (*Colocasia esculenta* (L.) Schott (Cocoyam),**” are my own research work. This work has not been published or accepted and submitted for any degree award.

Plagiarism checked at Birendra Multiple Campus library also confirmed that the work is original and genuine.



.....
Mr. Ajaya Mahato

Symbol No: CHE 2199/077

T. U. Registration no: 5-2-19-577-2011

RECOMMENDATION

The dissertation entitles “Nutrient Elements and HR-LCMS Analysis of Secondary Metabolites Present in Methanolic Extracts of (*Colocasia esculenta* (L.) Schott (Cocoyam),” is submitted by Mr. Ajaya Mahato for the completion of M.Sc. degree in Chemistry at Birendra Multiple Campus. The entire work is completed under our supervision. All the reports presented here are his finding. We confidently recommend this thesis for final evaluation.



Dr. Ganga Raj Pokhrel

(Supervisor)

Assoc. Professor

Department of Chemistry

Birendra Multiple Campus, Bharatpur, Nepal

Co- supervisor

Bodh babu Bhattarai (Ph.D.)

(Assistant prof.)

Department of Chemistry

Birendra Multiple Campus, Bharatpur, Nepal

February, 2025

FOREWORD

The thesis work “Nutrient Elements and HR-LCMS Analysis of Secondary Metabolites Present in Methanolic Extracts of (*Colocasia esculenta* (L.) Schott (Cocoyam),” submitted by Ajaya Mahato as a part of M.Sc. Course work in Chemistry at Birendra Multiple Campus is carried out under my supervision. Any part of this thesis work has not been submitted for any other degree award.



Supervisor

Ganga Raj Pokharel, Ph.D.

(Associate professor)

Department of Chemistry

Birendra Multiple Campus, Bharatpur, Nepal

Date: February, 2025

LETTER OF APPROVAL

Date: 6th February, 2025

On the recommendation of **Ganga Raj Pokhrel, Ph.D. (Assoc. Prof.)** and **Bodh Babu Bhattarai, Ph.D. (Asst. Prof.)**, this M.Sc. thesis submitted by **Ajaya Mahato** entitled "**Nutrient Elements and HR-LCMS Analysis of Secondary Metabolites Present in Methanolic Extracts of (*Colocasia esculenta* (L.) Schott (Cocoyam),**" is forwarded by Department of Chemistry, Birendra Multiple Campus (BMC) to the office of Dean, IOST, T. U.



Supervisor

Ganga Raj Pokharel, Ph.D.

(Associate professor)

Department of Chemistry

Head of Department of Chemistry

Birendra Multiple Campus,

Bharatpur, Nepal

ACKNOWLEDGEMENTS

I would like to express my deepest gratitude to my supervisor, **Ganga Raj Pokhrel, Ph.D. (Assoc. Prof.)** for his extraordinary support, guidance, and encouragement throughout the course of my dissertation. His insightful feedback, patience, and unwavering dedication were instrumental in shaping this work. Without his expertise and constant encouragement, this dissertation would not have been possible. I am truly grateful for the opportunity to learn under his mentorship and for the invaluable experience this journey has provided.

Thank you, **Ganga Raj Pokhrel, Ph.D. (Assoc. Prof.)**, for your exceptional support and for believing in my abilities even when I doubted myself.

I wish to extend my heartfelt thanks to my co-supervisor, **Bodh Babu Bhattarai, Ph.D. (Asst. Prof.)**, for his extraordinary contribution to my thesis. His deep insights, constructive feedback, and constant encouragement have been invaluable to the development and completion of this work. I am immensely grateful for the time and effort he invested in guiding me, and for his unwavering support throughout this process.

I am very much indebted with Mrs. Pavitra Shrestha Head of Department, Department of Microbiology and Lab incharge Mr. Santa Prasad Dhungana, my seniors Mr. Bipin Khanal, Miss Ranjana Khanal, Miss. Purnima Banjade, in my dissertation. My academic career has been greatly aided by the department's resources, advice, and encouragement. I would like to express my deepest gratitude to Center for Research in Nanotechnology and science (CRNTS), Sophisticated Analytical Instrument Facility (SAIF), Indian Institute of Technology, Bombay, 400076, India.

Last but not least, I would like to express my heartfelt gratitude to my parents, my wife and kids for encouraging me throughout this journey.

Ajaya Mahato

Symbol No: CHE 2199/077

T.U. Registration no.: **5-2-19-577-2011**

Email: ajayamahato203@gmail.com

ABSTRACT

Nutrient Elements and HR-LCMS Analysis of Secondary Metabolites Present in Methanolic Extracts of *Colocasia esculenta* (L.) Schott (Cocoyam) is still not understood clearly. Methanolic extracts of three different organs, i.e. tuber, petiole and leaves of plants were utilized for antioxidants activities using 2,2-Diphenyl-1-picrylhydrazyl (DPPH) free radical scavenging and phosphomolybdate assay (TAA). Total phenolic (TPC), total flavonoid (TFC) and total carbohydrate contents (TCC) were analyzed using spectrophotometer. The antimicrobial susceptibility test was conducted on seven microorganisms i.e. *E. coli*, *K. pneumonia*, *P. aeruginosa*, *B. subtilis*, *S. aureus*, *A. baumannii* and *S. typhispecies*. The ZOI was evaluated as: *A. baumannii* (18.60 ± 1.15 mm), *B. subtilis* (26.30 ± 4.04 mm), *E. coli* (14.33 ± 0.57 mm), *S. aureus* (14.00 ± 1.00 mm), *K. pneumonia* (16.00 ± 1.00 mm) respectively. Nutrient elements and toxic metallic element quantification and secondary metabolites analysis of the extract of underground and above ground organs (i.e. tuber, petiole and leaves) were carried out using ICPAES and HR-LCMS respectively.

IC₅₀ value of extracts lied between 240.45-1018.44 µg/mL. The amount of TAA using phosphomolybdate assay in the samples were found 1.25 ± 0.01 to 28.29 ± 0.01 mg AAE/g DS. The analysis of TPC in different samples were evaluated between 6.38 ± 0.00 and 19.96 ± 0.01 mg GAE/g DS. The total amount of TFC was in the range of 10.46 ± 0.00 and 89.10 ± 0.00 mg QE/g DS. The overall amount of TCC in the samples remained in between 1.78 ± 0.00 and 112.00 ± 0.00 mg GE/g DS. Concentrations of nutrient elements in tuber, petiole and leaves were found in the order as: **Ca** (8910.00, 303.00, 355.00 mg/Kg), **Cr** (0.47, 0.25, 0.42 mg/Kg), **Cu** (1.22, 0.67, 0.72 mg/Kg), **Fe** (15.60, 1.24, 2.41 mg/Kg), **Mg** (96.80, 43.60, 104.00 mg/Kg), **Zn** (128.00, 0.68, 1.94 mg/Kg) and toxic elements **As** (15.70, 9.80, 65.60 mg/Kg) whereas the concentration of **Pb** was below the detection limit (1ppb).

HR-LCMS analysis of the methanolic extracts of tuber, petiole and leaf of *Colocasia esculenta* revealed large number of phytochemical constituents. The tuber extract contained 200 different molecules out of which 140 remained unidentified, while 60 were found listed in the HR-LCMS library. The identified compounds comprised 11 alkaloids, 4 phenolic compounds, 5 flavonoids, 3 glycosides, 5 steroids, and 32 other miscellaneous compounds.

Similarly, the extract of petiole also contained 200 different plant secondary metabolites, including 85 unidentified and 115 identified constituents. Among the identified compounds, 12 were alkaloids, 15 phenolic compounds, 26 flavonoids, 9 glycosides, and 53 other compounds.

Likewise, in the leaf extract, a total of 197 compounds were detected, among them 106 were unidentified and 91 were identified. The identified constituents contained 13 alkaloids, 17 flavonoids, 3 terpenoids, 1 steroid, 12 pharmaceutical drugs, and 45 other compounds.

Notably, several potentially bioactive compounds were detected across the extracts, including neuraminic acid, amprenavir, butin, catechin, cyclopamine, disopyramide, funtumine, gentamycin, jervine, kaempferol, ketamine, retapamulin, swertiamarin, syringic acid, and usambarensine, among others. These compounds are known to possess diverse pharmacological activities, highlighting the therapeutic potential of *Colocasia esculenta*.

Key words: Cocoyam, Phytochemical screening, Antioxidant activity, Antimicrobial Activity, DPPH, HR-LCMS

सारांश

कोलोकेसिया इस्कलेन्टा (एल.)स्कट (कर्कलो) को पौष्टिक तत्व र HR-LCMS विश्लेषण अभै स्पष्ट रूपमा बुझ्न सकिएको छैन । पालुङ्ग, उपरदाङ्गढी र पदमपुरबाट सर्किलित कर्कलोको तीन भागहरु (पिडालु, डाँठ र पात) को मिथानोलिक एक्स्ट्र्याक्टबाट एन्टिअक्सिडेन्ट गतिविधी, जम्मा फिनोलिक मात्रा (टि.पि.सी.) जम्मा फ्लाभोनोइड मात्रा (टि.एफ.सी) जम्मा कार्बोहाइड्रेट मात्रा (टि.सी.सी), स्फेक्ट्रोफोटोमिटर को प्रयोग गरी निकालियो । एन्माइक्रोवियल गतिविधी सात सुक्ष्म जीवहरु (इ.कोली) के., न्युमोनि, पि.एरुजिनोसा, बि. सव्टिलिस, एस. आरियस, ए.बौमानी, एस. टाइफी,) को विरुद्धमा हेरियो । पोषक तत्व र विशाक्त धातुहरुको विश्लेषण एवम् HR-ICMS विश्लेषणको लागी कर्कलोको नमुना सेन्टर फर रिसर्च इन नानोटेक्नोलोजी एण्ड साइन्स, आइ.आइ.टी.बम्बे पठाइएको थियो ।

विरुवाको एक्स्ट्र्याक्टको आइ.सी.₅₀ मात्रा २४०.४५-१०१८.४४ माइक्रोग्राम/मि.लि रहेको पाइयो/त्यसैगरी टि.पि.सी, टि.एफ.सी, टि.सी.सी: क्रमशः ६.३८ ± ०.००४ देखी १९.९६ ± ०.०१ मि.ग्रा. जि.ए.इ/ग्रा.डि.एस, १०.४६ ± ०.०००७ देखी ८९.१ ± ०.००६ मि.ग्रा. क्यु.इ/ग्रा.डि.एस, १.७८ ± ०.० देखी ११२.० ± ०.० मि.ग्रा.जि.इ/ग्रा. डि.एस रहेको पाइयो फोस्फोमोलिब्डेनम एस्सेद्वारा जम्मा एन्टिअक्सिडेन्ट गतिविधी १.२५ ± ०.०१ देखी २८.२९ ± ०.०१ को विचमा रहेको पाइयो । त्यस्तै पोषक तत्व र विशाक्त तत्वहरुको मात्रा निम्न रहेको पाइयो । क्याल्सियम (०.०७६, ०.२५२, ०.४२७, ३५५ मिग्रा/के. जि, कोमियम (०.४७६, ०.२५२, ०.४२७ मि.ग्रा/के.जि) म्याग्नेसियम (९६.८, ४३.६, १०४.० मि.ग्रा/के.जि) फलाम (१५.६, १.२४१, २.४१ मि.ग्रा/के.जि) जिङक (१२८.०, ०.६८५, १.९४२ मि.ग्रा/के.जि), आर्सेनिक (१५.७, ९.८, ६५.६ मि.ग्रा/के.जि)।

HR- LCMS विश्लेषण बाट उपरदाङ्गढीको पिडालुमा जम्मा २०० वटा यौगिकहरु रहेको पाइयो जसमध्ये १४० वटा पहिचान नभएका र ६० वटा पहिचान भएका, पहिचान भएका भएका मध्ये ११ वटा अल्कालोइड, ४ फिनोलिक, ५ फ्लाभोनोइड, ३ ग्लाइकोसाइट, ५ स्टेरोइड र ३२ अन्य यौगिकहरु थिए । डाँठमा जम्मा २०० यौगिक मध्ये ८५ पहिचान रहित ११५ पहिचान भएका, पहिचान भएका मध्ये १२ अल्कालोइड, १५ फिनोलिक, २६ फ्लाभोनोट, ९ ग्लाइकोसाइट, ५३ अन्य यौगिक रहेको पाइयो । पातको नमुनामा जम्मा १९७ यौगिक जसमध्ये १०६ पहिचानरहित, ९१ पहिचान भएका, पहिचान भएका मध्ये १३ अल्कालोइड, १६ फ्लाभोनोइड, ३ टर्पिनोइड, १ स्टेरोइड, १२ फर्मासिउटिकल ड्रग्स र ४५ अन्य यौगिक रहेको पाइयो । न्युरामिनिक एसिड, एम्प्रिनाभिर, बुटिन, क्याटेचिन, साइक्लोपामिन, डाइसोपाइरामाइड, फुन्टुमाइन, जेन्टामाइसिन, जर्भिन, क्याम्फेरोल, किटामाइन, स्वेर्टियामरिन, सिरिन्जिक एसिड, उसाम्बरिन्सिन इत्यादि औषधीको रूपमा प्रयोग गरिदै आएको छ ।

मुख्य शब्दहरु: कोकोयाम (कर्कलो), फाइटोकेमिकल स्क्रिनिङ्ग, एन्टिअक्सिडेन्ट गतिविधी एन्टिमाइक्रोवियल गतिविधी, डिपिपिएव, एचआर-एल.सि.एम.एस

LIST OF ACROMYMS AND ABBREVIATIONS

Acetyl CoA	:	Acetyl coenzyme A
ADME	:	Absorption, Distribution, Metabolism and Excretion
Cm	:	Centimeter
DMSO	:	Dimethyl Sulphoxide
DMSO	:	Dimethylsulfoxide
DNA	:	Deoxyribonucleic acid
eGI	:	estimated Glycemic Index
FCR	:	Folin-Ciocalteu Reagent
g/L	:	Gram per Liter
GAC	:	Gallic acid concentration
GAE	:	Gallic Acid Equivalent
H	:	Hours
H ₂ SO ₄	:	Sulfuric Acid
HCL	:	Hydrochloride Acid
HR-LCMS	:	High Resolution Liquid Chromatography Mass Spectroscopy
HTS	:	High Throughput Screening
Kg	:	kilogram
mg	:	Milligram
MHA	:	Muller Hinton Agar
ml	:	Milliliter
mm	:	Millimeter
Na ₃ PO ₄	:	Trisodium phosphate
nm	:	Nanometer
NPs	:	Natural products
ppm	:	Parts per million
ROS	:	Reactive Oxygen Species
RTC	:	Root and Tuber Crops
TAA	:	Total antioxidant activity
TFC	:	Total Flavonoid Content

TPC : Total Phenolic Content
UV : Ultraviolet
wt : weight
ZOI : Zone of Inhibition

LIST OF SYMBOLS

°C	:	Degree Celsius
%	:	Percentage
&	:	And
	:	Lambda
μL	:	Microliter
μg/mL	:	Microgram per milliliter

LIST OF TABLES

Table 1	% yield of samples after drying	20
Table 2	% yield of samples after Soxhlet extraction	20
Table 3	Concentration of various elements through ICP-AES analysis	31
Table 4	Identified compounds in UPT using HR-LCMS	32
Table 5	Identified compounds in UPS using HR-LCMS	34
Table 6	Identified compounds in UPL using HR-LCMS	37

LIST OF FIGURES

Figure 1	Antimicrobial susceptibility test of plant extracts against:	21
Figure 2	Calibration curve of gallic acid to determine total phenolic content	22
Figure 3	Total phenolic contents of samples	23
Figure 4	Calibration curve of Quercetin to evaluate total flavonoid content	24
Figure 5	Total flavonoid contents of samples	25
Figure 6	Calibration curve of glucose dextrose to calculate total carbohydrates content	26
Figure 7	Total carbohydrates contents of samples	26
Figure 8	Calibration curve of ascorbic acid for phosphomolybdate assay	27
Figure 9	TAC of samples	28
Figure 10	Conc vs % inhibition of DPPH by A: BHA, B: IC50 of Samples	29
Figure 11	DPPH free radical scavenging activities of samples (IC50)	30
Figure 12	Chromatogram of UPT in +ve ESI mode using HR-LCMS	38
Figure 13	Chromatogram of UPS in +ve ESI mode	39
Figure 14	Chromatogram of UPL in +ESI mode	39
Figure 15	Chromatogram of UPT in -ve ESI mode	40
Figure 16	Chromatogram of UPS in -ve ESI mode	40
Figure 17	Chromatogram of UPL in -ESI mode	41

TABLE OF CONTENTS

TITLE PAGE	i
DECLARATION	ii
RECOMMENDATION	iii
FOREWORD	iv
LETTER OF APPROVAL	v
BOARD OF EXAMINER AND CERTIFICATE OF APPROVAL	vi
ACKNOWLEDGEMENTS	vii
ABSTRACT	viii
LIST OF ACROMYMS AND ABBREVIATIONS	x
LIST OF SYMBOLS	xiii
LIST OF TABLES	xiv
LIST OF FIGURES	xv
TABLE OF CONTENTS	xvi
CHAPTER 1: INTRODUCTION	1
1.1 Natural products	2
1.2 Oxidative stress and antioxidants	3
1.3 Antimicrobial activity	4
1.4 Essential nutrient elements and toxic metals	5
1.5 Research objectives	5
1.6 Rationale	6
CHAPTER 2 : LITERATURE REVIEW	7
CHAPTER 3. MATERIALS AND METHOD	10
3.1 Chemicals and equipment	10
3.2 Collection of plant	10
3.3 Methanolic extraction of phytochemicals.	11
CHAPTER 4 : RESULT AND DISCUSSION	20
4.1. Yield of sample after drying	20
4.2 Yield of samples after Soxhlet extraction.	20
4.3 Antimicrobial susceptibility test analysis	21
4.4 Estimation of total phenolic content	22
4.5 Estimation of total flavonoids content	24
4.6 Estimation of total carbohydrates content	25

4.7	Antioxidant activity	27
4.7.1	Phosphomolybdenum assay	27
4.7.2	DPPH free radical scavenging assay	28
4.8	Quantification of nutrient elements and toxic metals.	30
4.9	HR-LCMS PROFILING	31
CHAPTER 5. CONCLUSIONS and recommednation		56
5.1	Conclusion	56
5.2	Recommendations	57
REFERENCES		58
Appendix I : HR-LCMS METHODS OF ANALYSIS		81
Appendix II : HR-LCMS profiling of compounds in UPT in positive ESI mode.		85
Appendix III : HR-LCMS profiling of compounds in UPS in positive ESI mode.		87
Appendix IV : HR-LCMS profiling of compounds in UPL in positive ESI mode.		89
Appendix V : HR-LCMS profiling of compounds in UPT in negative ESI mode		91
Appendix VI : HR-LCMS profiling of compounds in UPS in negative ESI mode		93
Appendix VII : HR-LCMS profiling of compounds in UPL in negative ESI mode		95
Appendix VIII : PREPARATIONS		97
Appendix IX : SOME PHOTOGRAPHS		100
PLAGARISM REPORT		103

CHAPTER 1: INTRODUCTION

Over 7000 different kinds of edible plants have been used by humans as their food sources. However, to meet the world's food demand researchers has focused on a small number of crops. Just three crops (rice, wheat, and maize) provide more than half of the world's protein and energy demands, and about 35 crop plant species account for 95% of all crop production(Wada et al., 2019).Cocoyam (*Colocasia esculenta* (L.) Schott), a staple food crop in many tropical and subtropical regions, including parts of Africa, the Caribbean, and Asia, has garnered increasing attention not only for its nutritional value but also as a potential source of chemically valuable compounds. While primarily cultivated for its starchy corms and cormels. various parts of the cocoyam plant, including the leaves, stems, and even the peels, have been traditionally used in folk medicine and have demonstrated potential applications in various industries. This renewed interest stems from the growing recognition of the diverse phytochemical profile of cocoyam, encompassing a range of compounds with potential health benefits and industrial applications.Cocoyamis a significant source of carbohydrates, primarily starch, making it a valuable energy source (Eke et al., 2024). Beyond carbohydrates, it also contains appreciable amounts of dietary fiber, vitamins (e.g., vitamin C, B vitamins), and minerals (e.g., potassium, calcium, iron) (Chukwu et al., 2022). However, what makes cocoyam particularly interesting from a chemical perspective is its rich content of bioactive compounds. These include polyphenols, flavonoids, alkaloids, and saponins, among others. These phytochemicals have been associated with various biological activities, including antioxidant, anti-inflammatory, antimicrobial, and even anticancer properties ((Awa et al., 2015).

The presence of these diverse chemical constituents suggests that cocoyam could be explored as a source of valuable compounds for various applications. For instance, the high starch content could be utilized in the development of novel biodegradable materials or as a source of modified starches with specific functionalities (e.g., resistant starch). The presence of antioxidant compounds suggests potential applications in the food and pharmaceutical industries as natural preservatives or therapeutic agents. Furthermore, the reported antimicrobial activity of cocoyam extracts warrants further investigation into the specific compounds responsible and their potential use in developing new antimicrobial agents.

Despite the growing body of research on the nutritional and potential health benefits of cocoyam, a comprehensive understanding of its complex chemical composition, particularly concerning the specific types and concentrations of bioactive compounds present in different parts of the plant and under varying growth conditions, remains limited. Further research is crucial to fully characterize the chemical profile of cocoyam, identify and quantify its valuable constituents, and explore their potential applications in various fields. This thesis aims to address this gap by HR-LCMS analysis of secondary metabolites of methanolic extracts of cocoyam. This study will contribute to a deeper understanding of the chemical composition of cocoyam and its potential for utilization as a valuable resource.

1.1 Natural products

Nature is fundamentally important for human kind supporting life in various aspects for living such as food, shelter, ideal climate based on molecular evolution (Szuba, 2002). Nepal is rich in biodiversity due to its diverse geographical distribution from an altitude 70m from sea level to the tallest canopy, The Mt. Everest (8848m), in south Asia, in between China and India. The biodiversity of Nepal includes 118 varieties of ecosystems harboring over 2% of the flowering plants, 3% of the pteridophytes, 6% of bryophytes, 3.9% mammals, 8.9% of birds and 3.7% of global fauna of butterflies (Poudel et al., 2011).

Biologically active compounds derived from natural sources i.e. plants, animals, microbes etc. are the natural products (NPs). Secondary metabolites are plant NPs produced as end-products, in the forms of biosynthetic intermediate (acetyl coenzyme A (acetyl-CoA), shikimic acid, mevalonic acid, 1-deoxyxylulose-5-phosphate), during environmental adaptation or predator defense mechanism beside reproduction and regular growth and development of plant [Newman, 2007]. Natural products escorts with homogenous triumph rate (50%) in the Lipinski and parallel universe due to influence of active transport mechanism [Ganesa, 2008]. Earlier clinical practices of natural products were reported to be originated from North Africa, India and China which revolutionized into new era by the victorious synthesis of quinine, cholesterol, cortisone, chlorophyll and reserpine by Robert Burns Woodward [Zhang et al., 2020]. New-born male babies of South Californian Indian tribes were faithfully cooked in hot ashes of *Salvia* to nurture as the most robustified member of the tribe and to

safeguard from any respiratory maladies throughout the life [Newman, 2007]. The evolution of natural products in modern drug discovery, subsequent to expansion of combinatorial chemistry and High Throughput Screening (HTS) technologies, comprises high affinity and manifold array of biological choices particularly membrane proteins like ion channels, receptors, and transporters [Irina et al., 2010, Zhang et al., 2020]. The 3D molecular shape, stereochemistry and ring complexity of NPs are highly complicated incorporating far-flung ADME and physiochemical properties as contemplated the drug-like space [Chen et al.,2020]. Many natural products-based drugs like morphine, vinblastine, vincristine, quinine, avermectin, artemisinin, etoposide, teniposide, paclitaxel, and the camptothecin derivatives topotecan irinotecan etc. are being used in treatment of wide range of human diseases including parasitic diseases, Alzheimer's diseases, cardiovascular diseases, cancer, diabetes, neuro-degenerative disorders and many more [Chen et al.,2020, Kinston, 2010, Silva et al.,2014, Bedekar et al., 2010, Franziska et al., 2018,Shaito et al., 2020]. The topmost recognition of pertinence of NPs was 2015 Nobel prize in physiology or medicine for discovery of avermectin and artemisinin [Chen et al., 2020].

1.2 Oxidative stress and antioxidants

Highly reactive chemical species called reactive oxygen species (ROS) are produced in living organisms because of regular cellular mechanisms and external influences which can damage proteins and nucleic acids and change their activities. The physiologically significant ROS are superoxide anion (O_2^-), Hydroxyl radical (OH), hydrogen peroxide(H_2O_2) [Birben et al., 2012] alkoxy and lipid peroxy radicals, nitric oxide and peroxynitrite [Pisoschia et al.,2015]. The human body, as a defense system against oxidative damage, produces antioxidants. The change in the ratio of oxidants and antioxidants in the favor of oxidants is called oxidative stress [Gupta et al., 2014]. Prolonged manifestation of high levels of pro-oxidant chemicals can cause structural issues with mitochondrial DNA, abnormal gene expression and alterations in enzymatic mechanisms playing a major role in pathogenesis of chronic diseases like diabetes, cancer, heart diseases [Mehdi et al., 2020] neurodegenerative diseases like Alzheimer's disease, Parkinson's disease, Huntington's disease Amyotrophic Lateral Sclerosis as well as brain and spinal cord damages following stroke and traumatic brain injury [Olufunmilayo et al., 2023] and non-alcoholic

steatohepatitis and alcoholic liver disease [Li et al., 2015], hypertension, ischemia/perfusion, diabetes, acute respiratory distress syndrome, idiopathy pulmonary fibrosis, chronic obstructive pulmonary disease and asthma [Birben et al., 2012], infertility in both male and female [Bansal et al., 2010] and hepatic parthenogenesis [Ha et al., 2010]. The enzymatic and non-enzymatic antioxidants effectively block ROS in aerobic organisms [Birben et al., 2012]. Xenobiotics induced oxidative stress cause tissue damage persuading carcinogenesis due to gene mutation [Thomson et al., 1998]. Electron paramagnetic resonance only can identify the genuine presence of ROS [Jenkins, 1993]. The compounds, mainly polyphenols, which predominantly initiate the detoxification of ROS are antioxidants [Bjørklund et al., 2017]. Oxygen plays both beneficial as well as toxic role in living body. Oxygen is essential for life while its toxic effect is being used in hyperbaric and radiation therapy. Retinopathy of prematurity (ROP) was reported in new-born infants due to high oxygen concentration in newly launched incubators [Kohen et al., 2004]. The compounds which restrict the production and propagation of ROS are termed as antioxidants. A specific group of compounds (polyphenols, vitamins, trace elements etc.) or a system (enzymatic or non-enzymatic systems) scavenge the ROS inhibiting its ramifications. Polyphenols (flavonoids and phenolic acids) are influential inhibitor of ROS suppressing oxidative stress and lipid peroxidation by chelating effect [Al-Gubory et al., 2010].

This research aims to evaluate the concentration of total polyphenols i.e total phenic acids and total flavonoids present in differently spotted samples of cocoyam.

1.3 Antimicrobial activity

Varying species of microorganisms cause a wide array of health maladies in plants and animals including human beings. People from ancient time used different raw plant extracts to treat such maladies [Khan et al., 2019]. Incompatible varieties of molecules from plants functions as antibiotics, analgesic, antipyretic, anti-inflammatory, antiviral, antitumor etc. [Ramawat et al., 2009]. A revolution in the discovery and development of antibiotics has reached to the fifth generation after discovery of first antibiotic; Penicillin, by Alexander Fleming in August 1928 [Ligon et al., 2004]. Garlic and clove have antimicrobial, antiseptic and anesthetic properties [Lambert, 2004]. Very low concentration of some non-essential metals (Ag, Hg, Te) has demonstrated their ability to resist/kill bacteria and microbes [Russell et al., 2005].

Isolated phytochemicals (alkaloids, flavonoids, sesquiterpenes, diterpenes, lactones, triterpenes, naphthoquinones) from various medicinal plants are efficient against pathogens [Alanis, 2005]. Some enzyme inhibiting compounds have unassuming activities against bacteria [Fair et al., 2014]. Escalated use, misuse of antibiotics and the natal aptness of bacteria to resist against have made over 70% of antibiotics ineffective towards at least one or more bacteria [Lu et al., 2011, Odonkor et al., 2011]. It's highly substantial for researchers and pharmaceuticals to seek new and effective compounds to counteract high-resistive bacteria. This research focuses on probing the antimicrobial efficacy of cocoyam.

1.4 Essential nutrient elements and toxic metals

Many biological, chemical, molecular life processes need numerous metal/metalloids like cobalt, iron, manganese, copper, zinc in trace amounts while other metals (arsenic, lead, mercury, cadmium play noxious effect in mammals body including humans [Domingo et al., 2021]. Copper (II) ion is essential in metabolism of neural energy and antioxidants. Remarkable effects of low molecular weight chromium are studied in enhancing efficacy of insulin influencing metabolisms of carbohydrates, lipid and proteins. Zinc is associated with boosting of immune system and reduces sternness of diseases [Pokhrel et al., 2024]. Lead and arsenic have carcinogenic, neurotoxic and hematopoietic effects and affects reproductive system, kidney [Pokhrel et al., 2024, Assi et al., 2016]. The aim of this research is to enumerate the quantities of some nutrient elements as well as the toxic lead and arsenic using Inductively coupled plasma atomic emission spectroscopy (ICP-AES).

1.5 Research objectives

The broad objective of this research is to conduct nutrient elements and HR-LCMS analysis of secondary metabolites present in methanolic extracts of (*Colocasia esculenta* (L.) Schott (Cocoyam).

The specific object of this research are as follows:

- To measure total phenolic (TPC) & total flavonoids content (TFC).
- To evaluate total antioxidant activity (TAA).
- To quantify the carbohydrate contents(TCC).
- To determine the amounts of nutrient elements i.e. iron, zinc, copper, chromium, Calcium and Magnesium.

- To analyze the toxic heavy metals like arsenic and lead.
- To resolve secondary metabolites using HR-LCMS.

1.6 Rationale

Cocoyam, represents a vital, yet often underutilized, staple crop, particularly in developing countries across Africa, Asia, and the Pacific. This research aims to address critical gaps in our understanding of nutrient elements contents and chemical compositions of tuber, petiole and leaf through HR-LCMS and contribute to enhancing its contribution to food security and livelihoods. Cocoyam is a valuable source of carbohydrates, dietary fiber, vitamins (especially Vitamin C and B vitamins), and minerals (including potassium, iron, and calcium) (Okonkwo, 2013). Its low glycemic index makes it a suitable food for managing diabetes (Eleazu et al., 2016). However, the nutritional composition can vary significantly depending on the variety, growing conditions, and processing methods [Okechukwu, 2017, Adefega,2017, Ndabikunze et al., 2011]. Cocoyam plays a crucial role in food security, particularly in resource-limited communities (FAO, 2019). It is often cultivated as a subsistence crop and serves as a staple food, especially during lean seasons. Furthermore, cocoyam can contribute to income generation through local markets and value-added products [Oyenka, 2014]. Despite its nutritional and economic importance, cocoyam remains underutilized compared to other staple crops. There is a significant potential to expand its cultivation and utilization through improved agronomic practices, development of improved varieties, and diversification of its food applications [Okechukwu, 2017, Adefega,2017, Oyenka, 2014].

Existing research on cocoyam limited in nutrient, toxic element and HR-LCMS analysis of entire plant parts. This research aims to fill this gap by ICP-AES analysis of nutrient elements and HR-LCMS analysis of secondary metabolites present in methanolic extracts of cocoyam. The findings of this study will contribute to process cocoyam in food and pharmaceutical industries.

CHAPTER 2 : LITERATURE REVIEW

Many ethnic groups of people from Nepal uses different locally available plants and their parts as a cure of wide range of health discomforts and diseases. This indicates the prosperity of Nepal in medicinal plant biodiversity [Shrestha et al.,2016]. People from all around world feed mainly in cereals, tuber crops, beans, sugarcane, vegetable plants, fruits. The utility of tuber crops is increasing day by day due to their capability of being grown in varieties of climatic conditions facilitated by genetic diversity [Tay, 2013]. Cocoyam commonly known as taro (English), *karkalo* (Nepali) is one of the mostly used root and tuber crop (RTC) in the world. It is world's six most important root and tuber crop (Boakye et al., 2018). Mainly two species of cocoyam; *Colocasia esculenta* and *Xanthosoma sagittifolium* are cultivated in more than 65 countries across the world, mostly widespread in tropical regions (Wang, 1983) for their tasty, starchy root (Ce et al., 2017). Nutrient analysis of reveals that Cocoyam tuber is rich in carbohydrates, protein, fats, fiber [Ukom et al., 2018]. Oxalate content in cocoyam made it as food of under choice because it causes itching sensation in mouth and tongue (Okechukwu, 2017). Phytochemical screening indicates the presence of alkaloids, flavonoids, glycosides, phenols, steroids and tannins [Ogukwe et al., 2017]. Consumption of the parts of cocoyam may lower the risk of colon cancer due to its potential antioxidant effect (Aovi et al., 2018). Rich of antioxidant and nutrient in cocoyam have drawn special attention for the identification of biologically active components to reduce the risk of diseases (Okechukwu, 2017).

Qualitative and quantitative study of phytochemicals and nutrient element analysis were carried out in flowers and stems of two cocoyam varieties: *Xanthosoma sagittifolium* and, *Colocasia esculenta* from Nigeria (Oguke et al., 2017). Alkaloids, Flavonoids, Glycosides, Phenols, Saponins, Steroids, Tannins along with carbohydrates, crude proteins, crude fats, fiber are present in the organs of cocoyam. Proximate, Minerals and anti-nutrient contents in Cocoyam (*Xanthosoma sagittifolium* (L.) Schott) is also reported from Ethiopia (Ezeonu et al., 2016). Phytochemical constituents, anti-nutrients along with different elements are also reported in the study. The study reveals that antioxidant potential of cocoyam inflorescence decreases on boiling (Okechukwu, 2017) in Uturu. The nutrient elements Zn, Fe, Cu and vitamins E, B₂ & C is also reported in its mass. Phytochemical investigations and pharmacological screening of *Xanthosoma sagittifolium* (L.) leaf extract (Aovi et al.,

2018) in Bangladesh ensured the presence of different phytochemicals, antioxidant activity and mild antimicrobial activity against *Bacillus subtilis*, *Staphylococcus aureus*, *Escherichia coli*, *Salmonella enterica*, *Vibrio cholera* and *Shigella dysenteriae*. Raw form of Cocoyam contains considerable amount of proximates, phytochemicals, minerals, vitamins, amylose, amylopectin and antioxidants (Awa, 2015). Cocoyam and unripe banana incorporated feed has ameliorating potentials on renal and liver growth of diabetic rats, induced with 55 and 65mg/kg (Eleazu et al., 2013). Significant decrease in estimated glycemic index (eGI) and sugar content with pasting of cocoyam flour was observed. Paste of flour has shown significantly higher α -amylase and α -glucosidase inhibitory activities (Adefega, 2017). Cocoyam has significant scope on processing of food products with enhanced nutrition and potential to promote its utilization because of good contents of calcium, magnesium, copper, iron, sodium, zinc, manganese, and potassium (Ndabikunze et al., 2011). The dietetic value of cocoyam flour enriched with cowpea flour was reported to be higher which can be used as dietary supplement to treat malnutrition and chronic diseases caused due to deficiency of protein (Olayiwola et al., 2013). Cocoyam starches exhibited lower water absorption capacity and swelling power, paste clarity and viscosity but higher solubility, gelatinization temperatures and retrogradation tendencies than cassava and corn starches (Mweta et al., 2010). Apical section of Cocoyam tuber was found to rich on protein while distal section rich in ash, fiber and minerals. Potassium was the most abundant mineral found in cocoyam. The content of oxalate reduces with drum drying process (Dedeh, 2002). *Colocasia esculenta* possesses strong antioxidant properties and other compounds that justify its medicinal properties as used in ethno-medicine (Eleazu, 2016). The uses of cocoyam are limited because of high content of anti-nutrients like cyanides, oxalates which varies according to the habitat (Olatunde, 2018). 35 compounds were detected through GC-MS analysis (Rabiu et al., 2024). HPLC-DAD analysis reported 34 phenolic compounds (Ferrerres et al., 2012). No any literature about the HPLC-MS profiling, nutritional value and antimicrobial activity of Cocoyam grown at different altitudes of Nepal is reported till the date. This research will bridge the gap about the nutritional value of Cocoyam of Nepalese origin and other world.

General overview of *Colocasia esculenta*

Cocoyam (*Colocasia esculenta*) is herbaceous plant grown extensively in tropical regions. It is mainly used as a source of starch and animal feed. Twenty different species of cocoyam are found in Nepal. Cocoyam is found as cultivated as well as wild vegetation in the marginal lands and harsh environment. It is widely cultivated in home garden and fields, along with other plants such as turmeric, ginger (Pandey, 2001). It consists of three parts: leaves, which have a high protein and fiber content; stem, composed of lignocellulose; and a corm, with starch content similar to that of cassava and potatoes (Sebastian et al., 2018). It is herbaceous perennial crops of the family *Araceae* growing to a height of 1.5 to 2 m. The rhizome produces leaves that can grow up to 40 - 25 centimeter in diameter (Ubalua et al., 2016). Leaves are dark green color upward and light green beneath. The tips of the basal lobes are rounded or sub-rounded, and they have a mucronate, triangular-ovate shape at the apex. The petiole has a height of 0.8–1.2 meters. The spadix has flowering components that can reach a diameter of 8. This rhizome crop can withstand severe weather conditions and a longer time of storage than other root and tuber crops (Ubalua et al., 2016).

Systematic classification of cocoyam

Kingdom : Plantae
Subkingdom : Tracheobionta
Super division : Spermatophytes
Division : Magnoliophyta
Class : Liliopsida
Subclass : Arecidae
Order : Arales
Family : Araceae
Genus : *Colocasia* Schott
Species : *esculenta* (L.)
Common name: Taro

Synonyms: *Alocasia dussil* Dammer
Alocasia illustris W. Bull

Source: United States Department of Agriculture
Natural resources conservation service

<https://plants.usda.gov/home/classification/16337>

CHAPTER 3. MATERIALS AND METHOD

3.1 Chemicals and equipment

All chemicals i.e., standard and reagents used in this work were of analytical grade with high purity and distilled water (DW) was collected from small scale glass distillation set present in the chemistry laboratory of Birendra Multiple campus. DPPH (Sigma-Aldrich's. St. Louis, USA), DMSO (Merck Life Science Mumbai India), ascorbic acid (Merck Mumbai India), absolute methanol (Merck Mumbai India), Sodium carbonate (Merck Mumbai India), Sodium Hydroxide (Merck Mumbai India), Sodium Hydroxide (Merck Mumbai India), Ferric chloride hexahydrate (Merck Mumbai India), Sodium acetate (Merck Mumbai India), Potassium acetate (Merck Mumbai India), Hydrochloric acid (Thermo-Fisher Scientific India, Pvt. Ltd.), Sulfuric Acid (Thermo-Fisher Scientific India), Ethanol (Thermo-Fisher Scientific India), aluminum trichloride (Thermo-Fisher Scientific India) and double distilled water was brought from local vendor. Besides these, chemicals like Gallic acid (Loba Chemie Pvt. Ltd., Mumbai India), quercetin (Loba Chemie, Mumbai India), Nutrient Agar (Hi-media Pvt. Ltd, Mumbai India), Hinton Agar (Hi-media, Mumbai India) and other analytical reagents were used without further purification.

Software:

MS Power Point, MS word and MS Excel were used to interpret and analyses all information regarding this research. The software Origin Lab Origin pro was used for the analysis of the data and construction of graphs, curves.

3.2 Collection of plant

Colocasia esculenta(cocoyam) plant samples were gathered at three different elevations in Central Nepal: Padampur (208m), Upperdangadhi (1275m), and Palung (Makawanpur, 2310m). The aboveground plant portions were collected during the blooming season (September 2023) while the underground tubers were taken during the harvest season (January 2024). After being carefully cleaned with tap water and then distilled water, the collected plant pieces were put in a fresh zippered bag.

Identification of plant

The plant was sent to National Herbarium & Plant Laboratories (NHPL), Godawari-5, Lalitpur, for its systematic identification and authentication.

Drying, grinding and storage of plant

The collected plant parts were properly washed under tap water to remove any coarse materials and then with distilled water. The wet weight of sample was taken and recorded. The washed plant parts were chopped into thin slices using a clean stainless-steel knife. The chopped slices of samples were kept under shade drying at room temperature (24-30⁰C) in a well-ventilated room for about a month. During drying it was made cautious to avoid any fungal growth by continuous reversing in each 2 days until constant weight of samples were obtained. After that the dried sample parts were ground into fine powder using clean stainless-steel bladed grinder. The powdered samples were sieved using 60mm mesh to obtain homogenous powder and kept in a clean polythene zipper bag and kept in a desiccator till phytochemicals extraction to avoid moisture.

3.3 Methanolic extraction of phytochemicals.

Certain weights of the desiccated powder samples were taken using digital balance and packed into 25mm×100mm sized thimble and phytochemicals were extracted by Soxhlet extractor using polar solvent (300ml methanol). The extraction was continued for 10 hours by setting the Soxhlet temperature to the boiling temperature of solvent as per the indication of polarity index table.

Filtration of crude extract

A simple filtration method was carried out to filter the crude extract using Whatman No. 1 (45 µm) to remove unwanted solid particles.

Evaporation of solvent

The solvent from the filtered extract was evaporated in a hot water bath until gummy solid mass was obtained.

Storage of dried samples

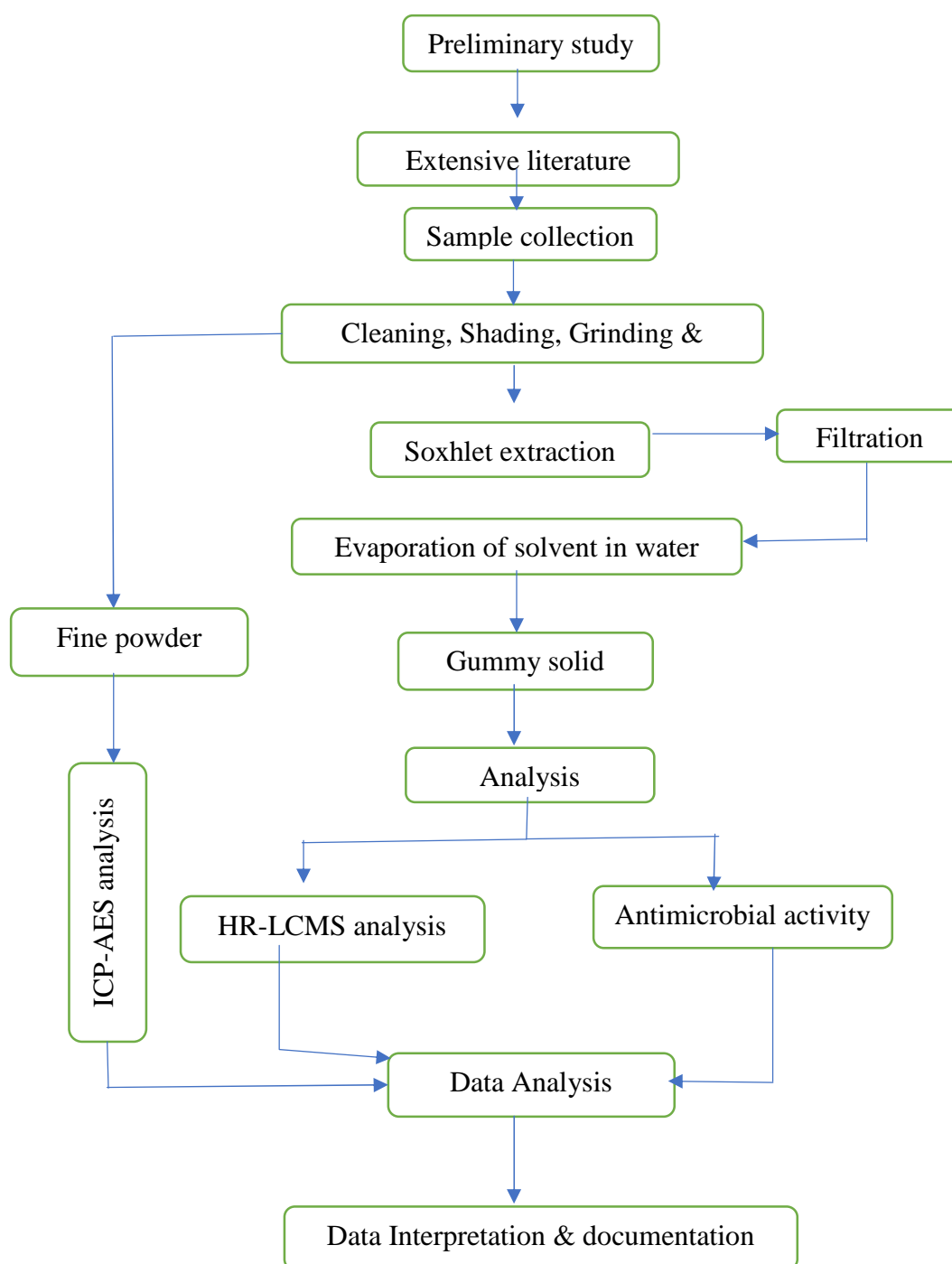
The dried gummy mass of extract was stored in an amber colored glass bottle, air tightly, wrapped with thin parafilm. A portion of samples were sent for HRLCMS profiling to Center for Research in Nanotechnology and science (CRNTS), Sophisticated Analytical Instrument Facility (SAIF), Indian Institute of Technology, Bombay, 400076, India.

Percentage yields of methanolic extracts

The % yield of methanolic extract was calculated as per the formula given below.

$$\% \text{ yield} = \frac{\text{Weight of extract}}{\text{Weight of the plant material}} \times 100$$

Schematic diagram showing the methodology of research



[Scheme 1: Flow chart showing entire methodology of research]

Antimicrobial susceptibility test of cocoyam

The antimicrobial susceptibility test of extracts of different parts of cocoyam was carried out by agar well diffusion protocol. The potency of plant extracts against bacterial activity was ascertained zone of inhibition. (Kuper et al., 2012). Clinical

laboratories standard institute 2018 guidelines were followed to conduct the antimicrobial susceptibility test.

Collection of test organisms

The strains of standard pathogenic bacteria *S. aureus*, *K. pneumoniae*, *S. typhi*, *P. aeruginosa*, *E. coli*, *A. baumannii* and *B. subtilis* were collected from Bharatpur Hospital, Chitwan, Nepal.

Preparation of plant extract stock.

A stock of 100ppm of each sample was prepared in Dimethyl sulfoxide (DMSO) and stored in a closed vial at -4°C in a refrigerator for further use.

Mueller Hinton Agar Preparation

45.6g of Mueller Hinton Agar was dissolved completely in 1200ml Distilled water and stirred with autoclaved glass rod. Then the mixture was transferred equally into three 500ml conical flasks. The agar mixture was then autoclaved at 121°C for 20 minutes. The pressure of autoclave was 15atm. Thus, prepared agar solution was left for few minutes under cooling. The agar plates of 90mm diameter were incubated at 180°C and left under laminar flow to avoid cross contamination. The agar solution was poured into the agar plates (approx. 20ml/plate) and covered with lid. The lid-covered plates were left for 45 minutes to cool. Thus, cooled agar plates were left under refrigeration for 24 hrs. The refrigerated agar plates were left under laminar flow for few minutes. Mueller Hinton Agar (MHA) plates, previously prepared were dried to eliminate excess water on the agar surface. The agar plates were labelled with date of experiment and organisms. The turbid solution of micro-organism was prepared using normal saline purchased from a local pharmacy, in different test-tubes for different micro-organisms. The turbidity of bacterial solution was finalized by comparing with 0.5 McFarland standard. A sterile cotton swab was immersed in the prepared inoculum and any surplus inoculum was removed by gently pressing and rotating the swab against the upper inner wall of test-tube, just above the liquid level. After each swabbing, the plates were rotated at 60-degree and final swabbing was done along the periphery of agar surface. The cross-contamination was avoided by leaving the inoculated plates to dry inside the laminar flow with covered lids. A sterile cork borer having 7mm of diameter was employed to punch the wall on the dried agar plate and marked accurately. 80 μl plant stock solution was gently loaded into respective wells

with the help of micropipette. The solvent (10% DMSO) was used as negative control while different antibiotics were placed as positive control. The agar plates with controls, samples were covered with lid and left about 45 minutes to diffuse the extracts and controls. The plates were uprightly placed in an incubator at 37°C for 24 hours for appropriate incubation. The clear zone of inhibition around each well in the agar plates were examined and measured using ruler. The average value of zone of inhibition was presented to disclose the strength of plant extract against pathogenic bacteria.

Quantification of total phenolic content

General protocol for estimation of total phenolic content

Estimation of total phenolic content was carried out by Folin-Ciocalteu method with slight modification by Lister and Wilson (2001) (Zahin et al., 2009). Briefly, to 0.5 ml of each sample (in triplicate) in graduated glass vial, 2.5 ml of 1/10 dilution of Folin-Ciocalteu reagent and 2 ml of 7.5% Na₂CO₃ (w/v) were added, shaken vigorously and incubated at 45°C for 15 minutes under light protected environment in hot water bath. After incubation, the absorbance having max at 765 nm was determined using UV-VIS spectrophotometer (T80+, PG Instrument, UK). The total phenolic content was expressed in milligrams of gallic acid equivalent (mg GAE/g DS) on the reference of calibration curve achieved with gallic acid as standard.

Measurement of total phenolic content (TPC)

Standard calibration curve of gallic acid was constructed and concentration of total phenolic content was determined by using the equation of straight line. The total phenolic content was calculated using formula;

$$\text{TPC (C)} = \frac{C \times v}{m} \dots\dots\dots (1)$$

Where, C = Total phenolic content (mg GAE/g DS)

c = concentration of gallic acid from curve (mg/mL)

v = Volume of extract (mL)

m = weight of plant extract (g)

Standard calibration curve of gallic acid was constructed and total phenolic content was determined from the equation of straight line.

Statistical analysis

The average absorbance from triplicate measurement of each concentration was evaluated which was used to determine linear coefficient and establishing the regression equation;

$$y = mx + c \dots \dots \dots (2)$$

Where, y = absorbance of extract

m = slope of curve

x = concentration extract

c = intercept

The total phenolic content of extract in gm GAE/g, was determined from the regression equation.

Quantification of total flavonoid content

General protocol for estimation of total flavonoid content

The total flavonoid content was determined by aluminium-chloride assay using quercetin as standard (Sagar et al.,2020). Stock solution of quercetin was prepared by dissolving 10mg quercetin in 10ml of 80% methanol. Varying concentration of quercetin were prepared from stock solution and used for calibration. 1.5ml of 80% methanol was added to standards and samples (0.5ml) followed by addition of 100µl AlCl₃, 100µl Potassium acetate and 2.8ml distilled water with vigorous shaking and left for incubation at 30⁰C for 30 minutes protected under light free condition. After incubation the absorbance was measured at 410nm using UV-VIS spectrophotometer (T80+, PG Instrument, UK). The total flavonoid content was expressed in milligrams of quercetin equivalent (mgQE/gDS). The measurement of TFC was done in the similar way done for TPC.

Quantification of total carbohydrates content (TCC):

General protocol:

The total carbohydrates of various plant extracts of *Colocasia esculenta* was quantified using modified sulfuric acid-phenol method using glucose dextrose as standard (Masuko et al., 2005). In brief, 500 μ L of standard and samples were taken in cleaned with distilled water and oven dried test-tubes. 1500 μ L of conc. H₂SO₄ was added to each test-tubes and shaken up to 30 minutes followed by addition of 300 μ L 5% phenol. The entire mixture was heated at 90⁰C for 5 minutes in a static water bath then cooled to room temperature (25⁰C) in another water bath for 5 minutes. The absorbance was measured and recorded at 490nm using UV-VIS spectrophotometer. An equation of regression line from standard calibration was implemented to quantify the total carbohydrates (TCC) and expressed as milligrams of glucose equivalent/g of dry sample (mgGE/g DS)

Measurement of total carbohydrates content (TCC):

Standard calibration curve of glucose dextrose was constructed and concentration of total carbohydrates content was determined in the similar way done for TFC.

The total antioxidant activity (TAA): Phosphomolybdenum assay

General protocol:

The phosphomolybdate method was implemented for estimation of total antioxidant capacity of samples using ascorbic acid as standard (Jie et al., 2011). In brief, 1ml of extract and standard were separately mixed with 4ml of phosphomolybdenum reagent. The phosphomolybdenum reagent was prepared by mixing equal volumes of 0.6M H₂SO₄, 28mM Na₃PO₄ and 4mM ammonium molybdate. The mixture of extract and phosphomolybdenum reagent was incubated at 95⁰C for 90 minutes and cooled to room temperature. After cooling, the absorbance having max at 695nm using spectrophotometer (T80+, PG Instrument, UK). The total antioxidant capacity of sample extract was expressed as mg AAE/g DS in comparison to the standard ascorbic acid calibration curve. A blank of 1ml reagent and 4ml methanol was implemented for the estimation of TAA.

Blank = Reagent + Solvent

DPPH free radical scavenging assay:

General protocol

The total antioxidant capacity of different plant extracts of cocoyam was carried out by DPPH free radical scavenging assay (Islam et. al., 2015). 1ml of varying concentrations of ascorbic acid standard and sample extracts were mixed with 2ml of 0.004% of methanolic DPPH and incubated at room temperature for 30 minutes in dark. After incubation, the absorbance of standard and samples were measured at 517nm using UVVIS spectrometer. A blank of 1ml methanol and 2ml 0.004% methanolic DPPH was used as control.

Measurement of DPPH free radical scavenging activity:

The following formula was implemented to estimate the free radical scavenging activity of samples.

$$\text{Percentage scavenging (\%RSA)} = \frac{A_{\text{control}} - A_{\text{sample}}}{A_{\text{control}}} \times 100$$

Solutions of varying concentrations of ascorbic acid as standard was used to evaluate the 50% of radical scavenging capacity (**IC₅₀**) and graph was constructed to correlate the %RSA of varying concentrations of samples.

Quantification of nutrient elements and toxic metals

Three plant samples (UPT, UPS and UPL) powder was sent to Center for Research in Nanotechnology and science (CRNTS), Sophisticated Analytical Instrument Facility (SAIF), Indian Institute of Technology, Bombay, 400076, India.

0.1g of sample was subjected to microwave digestion (Model: Anton Paar microwave go) using a mixture of 2ml conc. HCl, 1ml conc. HNO₃ and 2ml HF [Pokhrel et al., 2019]. The system was programmed to operate as 190°C for 25 min (15 min ramp time) and cooled to room temperature and the volume was made up to 25ml with distilled water. Elements were detected by ICP-AES.

HRLCMS profiling

An Agilent LC instrument column composition (model: G1316C) and a quadrupole time of flight mass spectrometer (QTOF-MS, Agilent technology) with a diode array detector (DAD) (model: G226A) through electro spraying ionization (ESI), was used for chromatographic separation. The chromatographic apparatus was fitted with a binary pump (model: G220B) and hip sampler (model: G4226A). Chromatographic separation was performed in a column compartment that was set at 40 °C. The Hip auto sampler was used with a 5.00 μ L injection volume, a binary pump with a post time of 35 minutes, a flow rate of 0.3 μ L/min, a maximum flow gradient of 100mL/min, and a maximum pressure of 1200.00 bar. The mass spectrometer's acquisition method with an MS scan rate (spectra/sec) of 100, was configured for a dual ESI ion source with a range of (m/z) 120–1200. The scan segment was conducted in both positive (+ve) and negative (-ve) modes. The drying gas temperature was set at 250 C with gas flow at 13 L/min, nebulizer gas at 35 psig, capillary voltage at 3500V, fragment voltage at 175V, skimmer voltage at 65V, nozzle voltage at 1000V, and octopole RF peak at 750V. Acetonitrile (B) and water: formic acid (100:0.1) (A) make up the mobile phase. The multistep linear gradient 0-1 min 5% B, 25min 100% B, 30min 100% B, 31min 5% B, and 35min 5% B comprised the chromatographic technique.

The column over temperature was kept at 40°C. Data acquisition and processing were performed using mass hunter workstation software and the mass analyses were done by both positive and negative mode ion modes respectively. The initial scan was done by 190nm to 640nm with m/z 120-1200. The parent ion and source –induced dissociation fragment could provide the accurate mass information. The HR-LC-MS data processing were used for the identification of constituents and ESI chromatogram is obtain to calculate mass to charge ratio (m/z) of the selected peaks

CHAPTER 4: RESULT AND DISCUSSION

4.1. Yield of sample after drying

The yield of samples of *Colocasia esculenta(L.) Schott.* after shade drying under wellventilated condition is tabulated below.

Table 1 : % yield of samples after drying

SN	SAMPLES CODES	WET WEIGHTS	DRY WEIGHTS	% YIELDS
1.	PDS	106.00gm	20.22gm	19.07%
2.	PDL	93.00gm	19.99gm	21.50%
3.	PDT	440.00gm	91.53gm	20.80%
4.	UPS	100.00gm	13.00gm	13.00%
5.	UPL	92.00gm	22.95gm	24.95%
6.	UPT	690.00gm	184.82gm	26.78%
7.	PS	220.00gm	37.00gm	16.81%
8.	PL	80.00gm	19.71 gm	24.64%
9.	PT	795.00gm	129.85gm	16.33%

The highest yield after drying was obtained from tuber sample from Upperdangadhi(26.78%) and the least yield from leaf sample from same spot (13%).

4.2 Yield of samples after Soxhlet extraction.

The yields of sample extracts obtained from dry weights of samples of *Colocasia esculenta (L.) Schott.* using Soxhlet apparatus and methanol as solvent is tabulated below.

Table 2 : % yield of samples after Soxhlet extraction

SN	SAMPLES	WEIGHT TAKEN(g)	YIELD(g)	% YIELD
1.	PDS	10.02gm	0.94gm	9.45%
2.	PDL	10.00gm	1.12gm	11.27%
3.	PDT	25.03gm	3.51gm	14.02%
4.	UPS	10.02gm	1.18gm	11.83%
5.	UPL	10.01gm	10.01gm	6.29%
6.	UPT	45.01gm	2.42gm	5.37%
7.	PS	15.16gm	4.38gm	28.93%
8.	PL	10.01gm	1.93gm	19.34%
9.	PT	30.02gm	3.05gm	10.16%

The Soxhlet extraction of sample of stem from Palung yielded the highest (28.93%) and the yield of tuber sample from Upperdangadhi was found to be the least (5.37%).

4.3 Antimicrobial susceptibility test analysis

100ppm (80mL) solutions of methanolic extract of samples in 10% DMSO was used to carry out antimicrobial susceptibility test against both gram +ve and gram -ve bacteria. Same volume of 10% DMSO was used as positive control. The entire antimicrobial activities of samples are shown in the graph below:

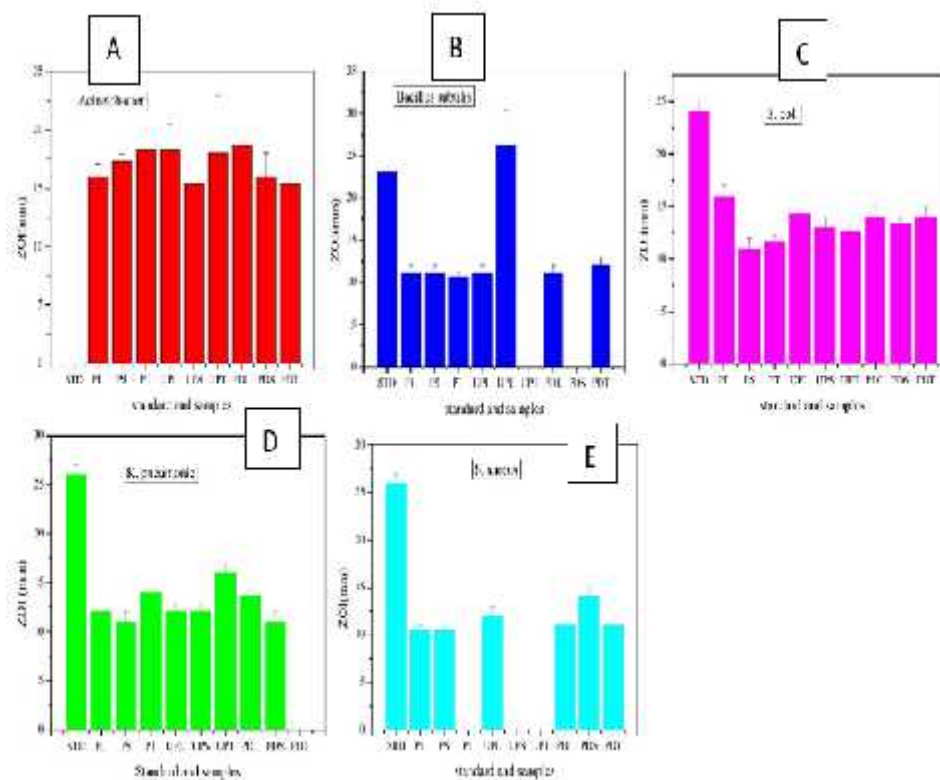


Figure 1 :Antimicrobial susceptibility test of plant extracts against: **A:** Acinetobacter using varieties of antibiotics (Amikacin 30mcg, Ciprofloxacin 5mcg, Ampicillin 10mcg, Gentamycin 10mcg, Ceftriaxone 30mcg, Ceftazidime 30mcg, Ceftazidime-avibactam 30mcg, Imipenem 10mcg, and Ceftazidime/ clavulanic acid 30/10mcg) as standards, **B:** Bacillus subtilis, Amikacin 30mcg as standard, **C:** Escherichia coli, Amikacin 30mcg as standard, **D:**Klebsiella pneumoniae, Amikacin 30mcg as standard, **E:**S. aureus, Amikacin 30mcg as standard]

The antimicrobial susceptibility test of sample extracts has shown a significant potential to inhibit the examined organisms. All sample extracts are efficient against *A.baumannii* even when a large number of standard antibiotics available in the market show no inhibition of the organism, indicating presence of more potential compound in the plant than the standard resistant *A. baumannii*. The plant extract samples were

seen not to be effective in inhibiting the *Salmonella typhi* and *Pseudomonas aeruginosa*.

4.4 Estimation of total phenolic content

Total phenolic content of sample extracts was estimated using gallic acid standard curve. Varying concentrations of gallic acid (5ppm,10ppm,20ppm,40ppm,80ppm and 100ppm) were used to calibrate the curve. The calibration curve of instrument was constructed between various concentration of gallic acid and absorbance at 765nm wavelength which is expressed below:

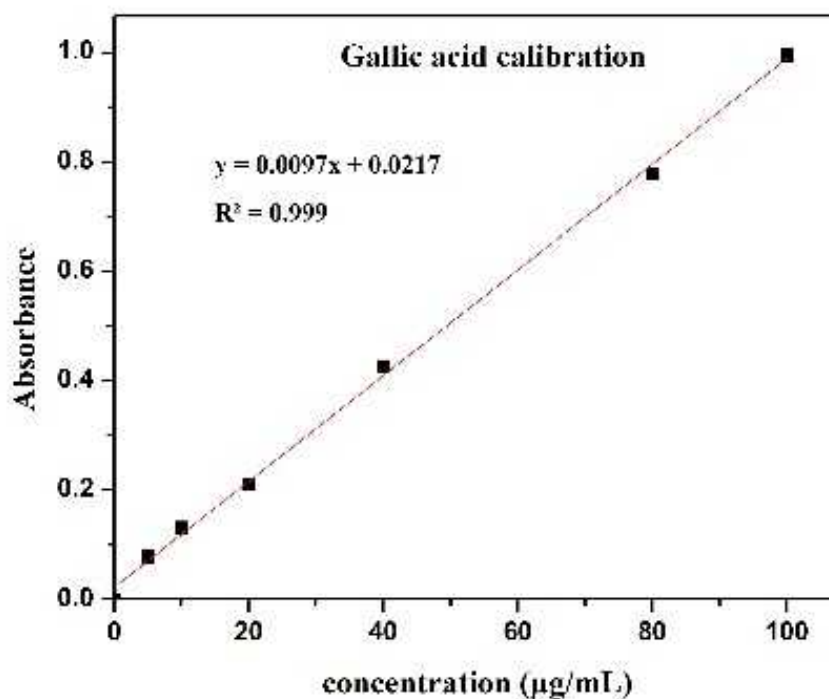


Figure 2 :Calibration curve of gallic acid to determine total phenolic content]

Calculation of total phenolic content of plant samples

The regression equation $y = 0.0097x + 0.0217$, $R^2 = 0.999$, obtained from Excel software sheet was employed to calculate the total phenolic contents of sample extracts.

Here, 'x' is concentration of gallic acid in microgram per milliliter ($\mu\text{g/mL}$)

'y' is absorbance

Slope of the curve (m) = 0.0097 and 'y' intercept (c) = 0.0217

ZA simple equation, $\text{TPC} = x (V/m)$ is used to calculate total phenolic contents of plant samples and is expressed in mg GAE/g DS.

From assay for determination of total phenolic content, the leaf extract from Padampur was found to be enriched with maximum phenolic compounds whereas the petiole extract from Padampur was contained with lower concentrations of phenolic compounds.

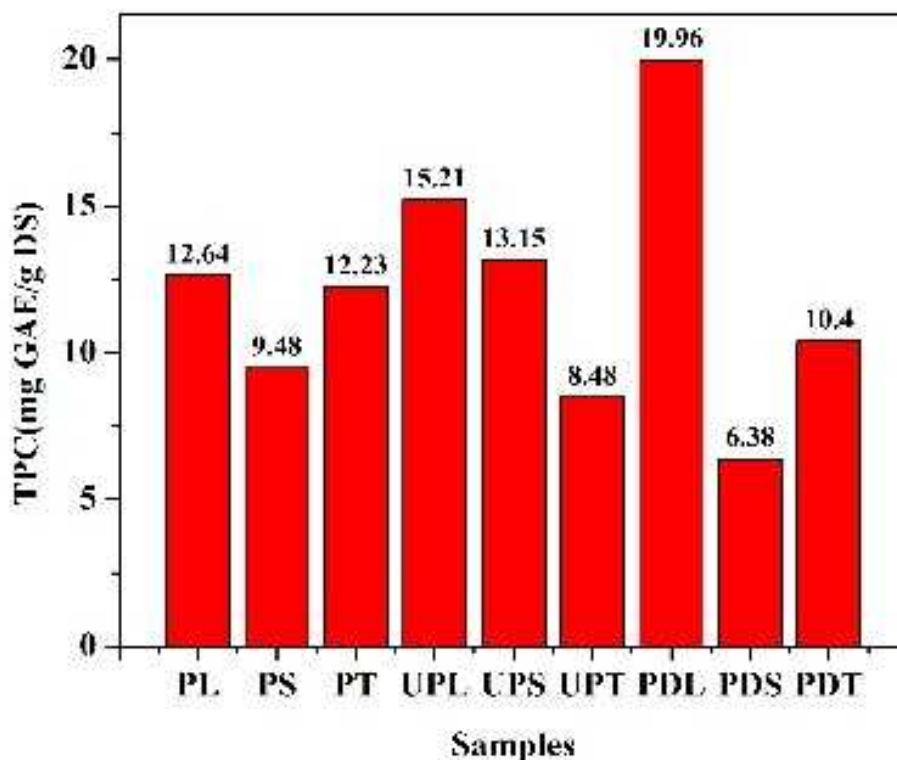


Figure 3 : Total phenolic contents of samples

4.5 Estimation of total flavonoids content

Estimation of total flavonoid content of samples was carried out by using Aluminium chloride colorimetric method. Varying concentrations of quercetin (10ppm,20ppm,40ppm and 80ppm) were used to calibrate the instrument. A calibration curve obtained from Excel software sheet by plotting concentration Vs absorbance at 410nm wavelength, with regression equation, $y = 0.0015x + 0.0048$ and $R^2 = 0.992$ was employed to estimate total flavonoid contents. The TFC was calculated in the similar manner as that for TPC and expressed as mg QE/g DS.

As similar to total phenolic compounds, good contents of flavonoids were found in the leaves of plant samples. The tuber and petiole were poorly enriched with the flavonoids.

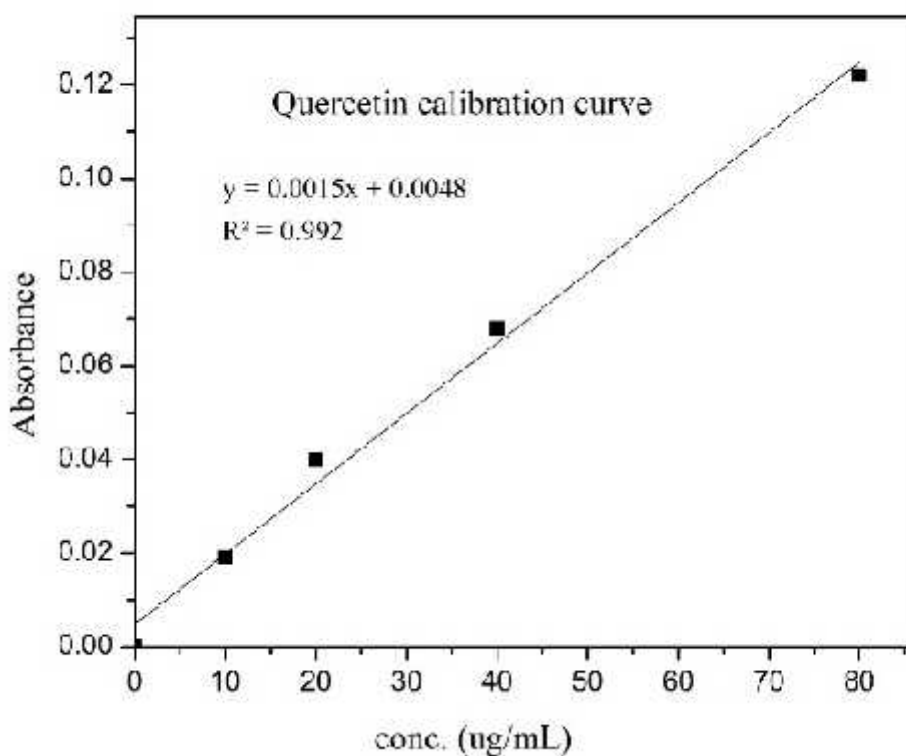


Figure 4 :Calibration curve of Quercetin to evaluate total flavonoid content

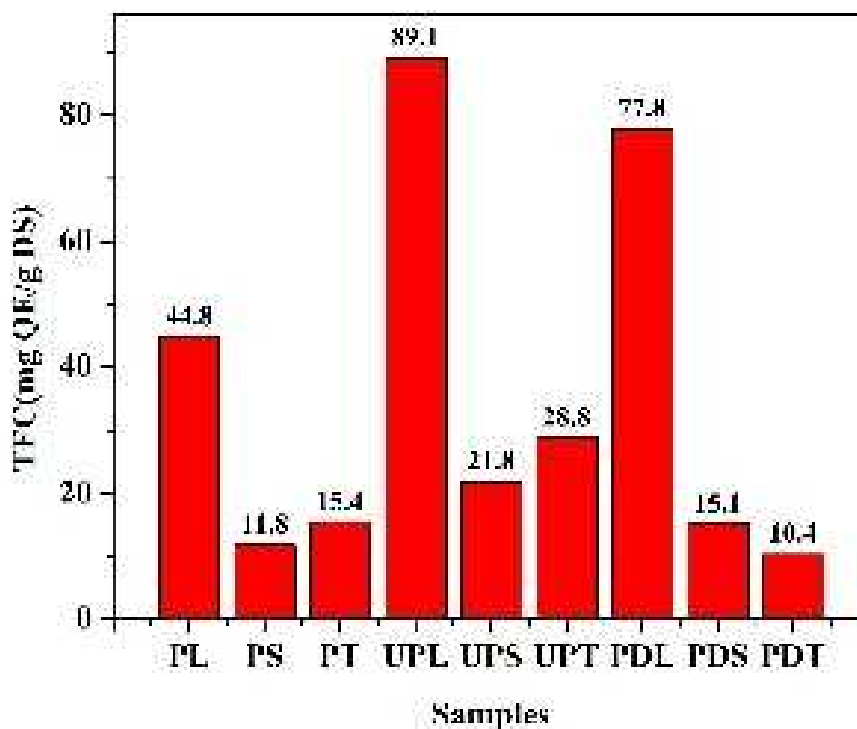


Figure 5 : Total flavonoid contents of samples

4.6 Estimation of total carbohydrates content

Modified sulfuric acid-phenol method was implied to quantify total carbohydrates contents (TCC) of samples using glucose dextrose as standard (Masukoetat., 2005). Different concentrations of glucose dextrose (30ppm,60ppm,90ppm,120ppm,150ppm) were used to calibrate the spectrophotometer. A calibration curve with regression equation $y = 0.0023x + 0.0149$, $R^2 = 0.986$, obtained from concentration vs absorbance, at 490nm wavelength, plot in Excel software was employed to calculate TCC in mg GE/g DS and the results are shown in table below. The calculation was made in the similar way as for the calculation of TFC.

The tubers of plants contain maximum carbohydrates followed by petioles and the leaves are poorly contained with carbohydrates.

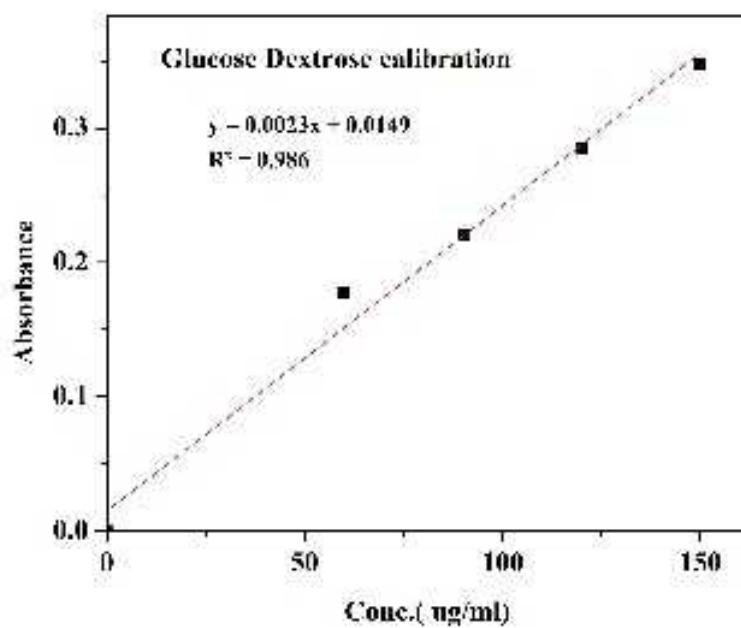


Figure 6 :Calibration curve of glucose dextrose to calculate total carbohydrates content

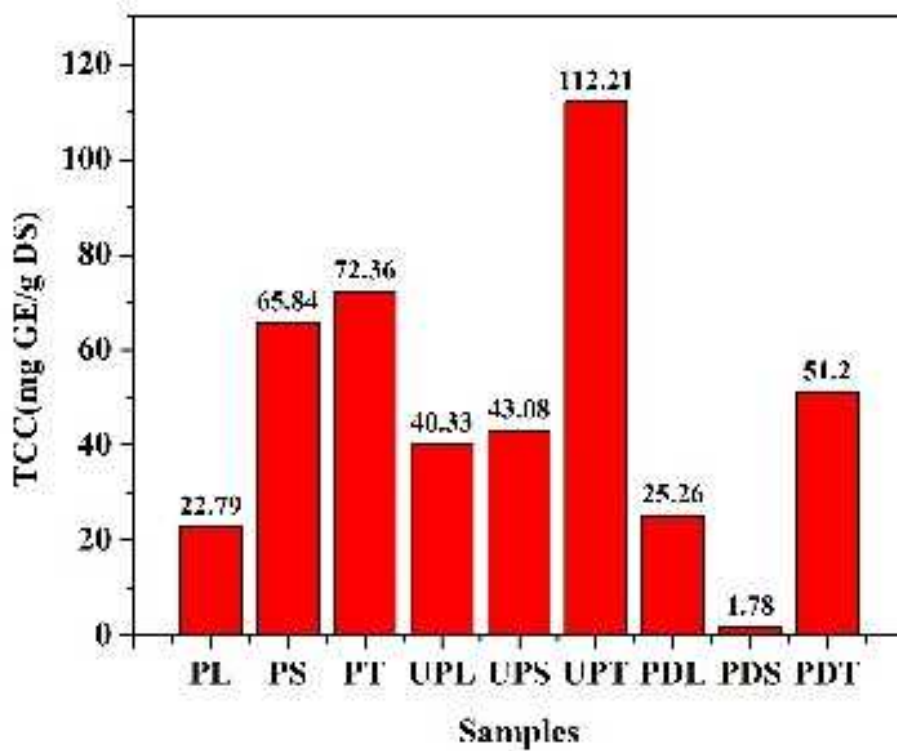


Figure 7 :Total carbohydrates contents of samples

4.7 Antioxidant activity

4.7.1 Phosphomolybdenum assay

The total antioxidant capacity (TAC) of samples was estimated following the protocols of phosphomolybdate assay (Jie et al.,2011) using ascorbic acid as standard. Different concentrations of ascorbic acid (5ppm, 10ppm, 20ppm, 40ppm, 80ppm, 150ppm) were used for calibration at 695nm wavelength. The regression equation, $y = 0.0071x + 0.0021$, $R^2 = 0.996$, obtained from a plot between concentrations and absorbance from Excel software sheet was used to calculate total antioxidant capacity in the similar way as that for the calculation of TCC and expressed as mg AAE/g DS which is given in the table below:

Varying antioxidant activities of plant extracts were revealed through phosphomolybdate assay. The highest antioxidant capacity was shown by leaf of plant extracts from Padampur followed by petiole of plant sample from Upperdangadhi. The petiole of plant from Palung have poor antioxidant properties.

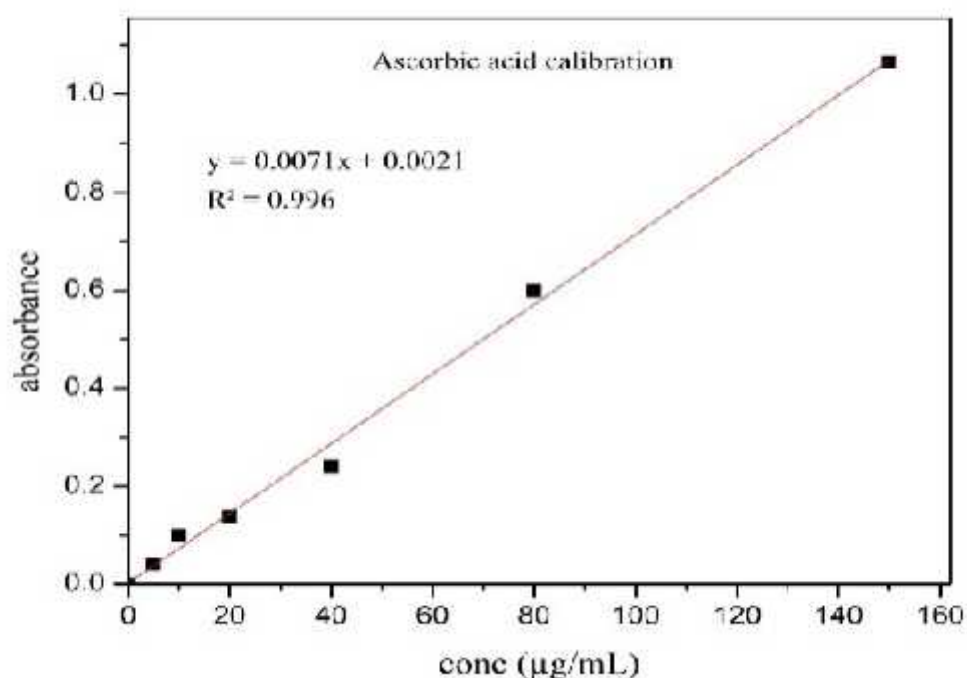


Figure 8 :Calibration curve of ascorbic acid for phosphomolybdate assay

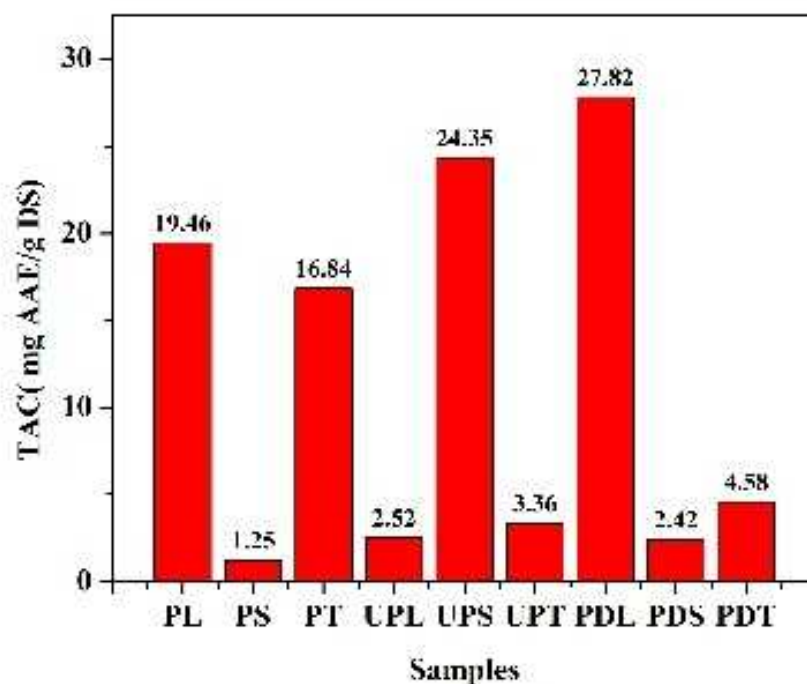
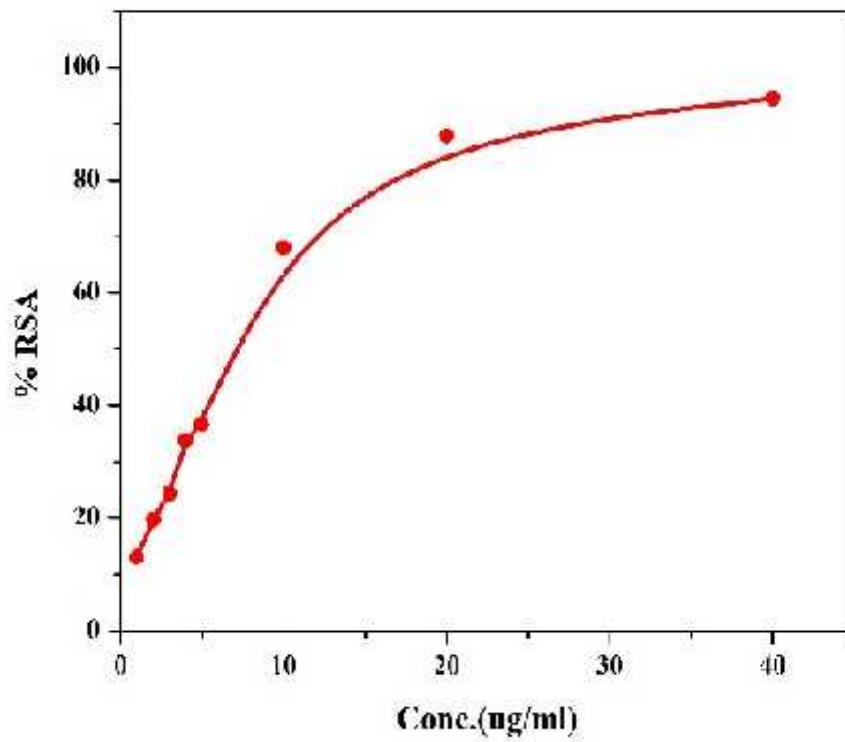


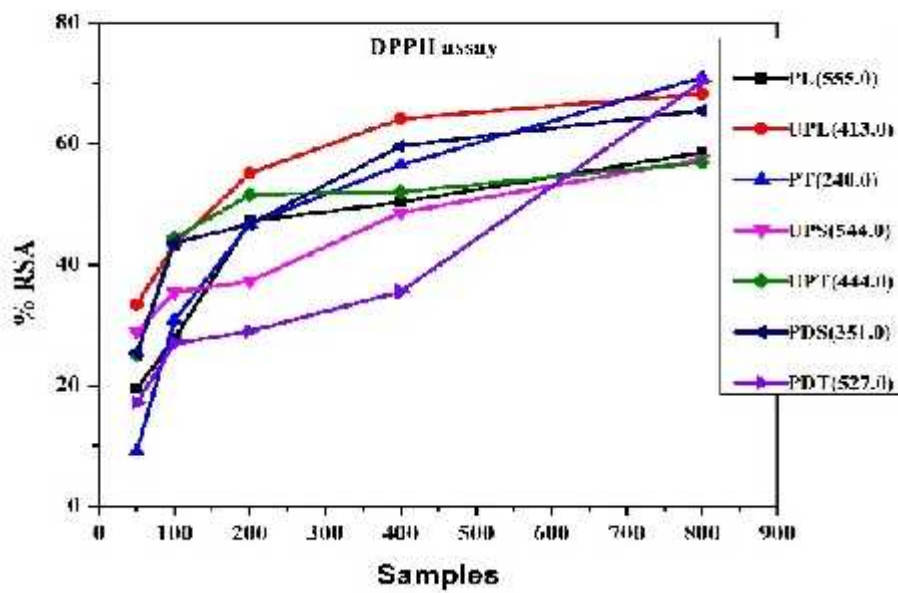
Figure 9 : TAC of samples

4.7.2 DPPH free radical scavenging assay

Total antioxidant capacity (TAC) of samples was estimated by DPPH free radical scavenging assay by butylated hydroxy anisole (BHA) as standard (Hasan et. at, 2015). The absorbance was measured at 517nm wavelength. The IC₅₀ value of both BHA and samples were calculated from regression equation $y = 6.0779x + 7.2854$, $R^2 = 0.996$, obtained from Excel by a plot between % inhibition and concentration which is shown in the table below:



(A)



(B)

Figure 10 :Conc vs % inhibition of DPPH by A: BHA, B: IC₅₀ of Samples

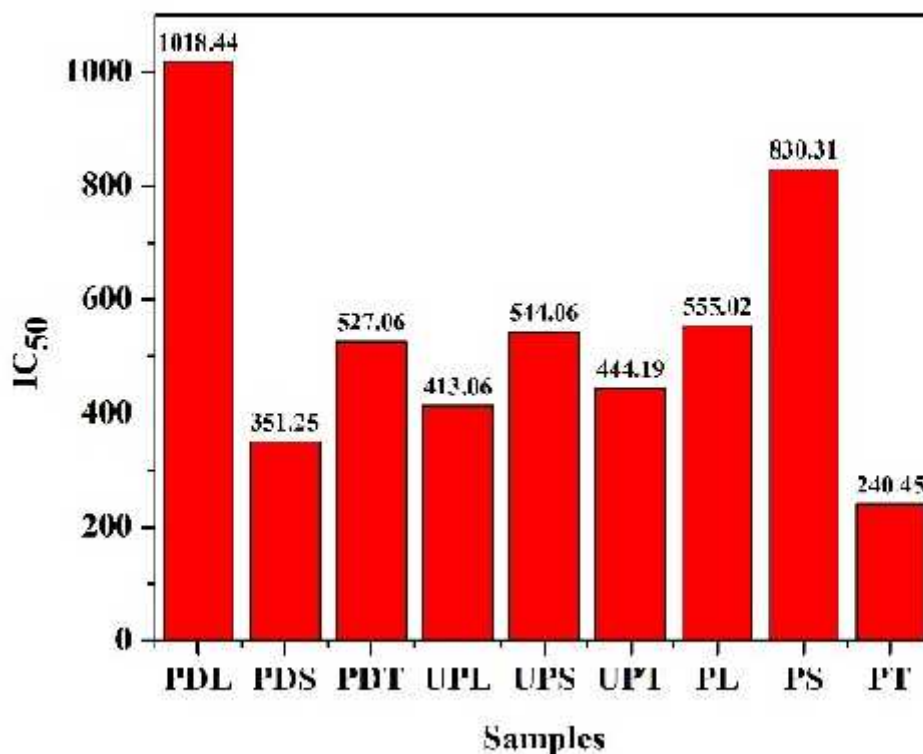


Figure 11 :DPPH free radical scavenging activities of samples (IC₅₀)

The DPPH free radical scavenging assay for estimation of total antioxidant activities of plant extracts using BHA as standard revealed varying potentials of samples to scavenge the free radicals. The minimum IC₅₀ value was for tuber of plant from Palung and the maximum IC₅₀ for leaf of plant extract from Padampur.

4.8 Quantification of nutrient elements and toxic metals.

The quantities of nutrient elements and toxic metals analyzed by ICP-AES at Center for Research in Nanotechnology and science (CRNTS), Sophisticated Analytical Instrument Facility (SAIF), Indian Institute of Technology, Bombay, 400076, India, is shown below:

Table 3 : Concentration of various elements through ICP-AES analysis

Name	Ca	Cr	Cu	Fe	Mg	Zn	As
	mg/Kg	mg/Kg	mg/Kg	mg/Kg	mg/Kg	mg/Kg	mg/Kg
UPT	8910.00	0.47	1.22	15.60	96.80	128.00	15.70
UPS	303.00	0.25	0.67	1.24	43.60	0.68	9.82
UPL	355.00	0.42	0.72	2.41	104.00	1.94	65.60

4.9 HR-LCMS PROFILING

The methanolic extracts of tuber, petiole and leaf *Colocasia esculents* was subjected to HR-LCMS analysis.

HR-LCMS analysis is conducted in both encompassing both positive and negative modes which revealed total 200 compounds present in UPT, among which 60 are known, 140 are unknown. Among known, 11 alkaloids, 4 phenolics, 5 flavonoids, 3 glycosides, 5 steroids and 32 other compounds are present. In the similar way UPS sample contains total 200 compounds, 115 known and 85 are unknown. Out of known, 12 are alkaloids, 15 phenolic, 26 flavonoids, 9 glycosides and 55 other class of compounds. The UPL sample screened total 197 compounds, 89 are known while 116 are unknown. The known compounds include 13 alkaloids, 17 flavonoids, 3 terpenoids, 1 steroid and 57 other compounds.

Table 4 :Identified compounds in UPT using HR-LCMS

S.N	Sample	Mode	Putative compounds	Retention Time(Min)	Mass	Molecular Formula	m/z	Chemical Class	Clinical uses
1			Diasarone 2	5.139	416.2199	C ₂₄ H ₃₂ O ₆	439.2095	phenolic	Anti dengue viral (Yao et, 2017)
2			Trenbolone	5.839	270.1643	C ₁₈ H ₂₂ O ₂	293.154	steroid	
3			Gentamicin	8.129	477.3206	C ₂₁ H ₄₃ N ₅ O ₇	478.3281	glyco side	Antibiotics (Chen et al., 2013)
4		+ve ESI mode	Dicyclomine	10.854	309.2627	C ₁₉ H ₃₅ NO ₂	332.252	alkaloid	Anti-tumor, anti-inflammatory (Lei et al., 2020)
5			Jervine	12.243	425.2937	C ₂₇ H ₃₉ NO ₃	448.2831	alkaloid	
6			Glutamyl-lysine	12.515	275.1479	C ₁₁ H ₂₁ N ₃ O ₅	298.1371	enzyme	
7	PDT		funtumine	13.781	317.2675	C ₂₁ H ₃₇ NO	318.2748	Steroidal alkaloid	Anti-tumor (Badmus et al., 2020)
8			Mangalkanyl glucoside	-6.82	386.2695	C ₂₁ H ₃₈ O ₆	373.2144	glucoside	Antioxidant, anti-inflammatory (Jiang et al., 2022)
9			Vanillic acid	6.435	168.0423	C ₈ H ₈ O ₄	167.0349	phenolic	anticancer, antiobesity, antidiabetic, antibacterial, anti-inflammatory, and antioxidant (Kaur et al., 2022)
10		-ve ESI mode	Catechin	7.128	290.0812	C ₁₅ H ₁₄ O ₆	289.0722	flavonoid	Anti-inflammatory, anti-tumor (Musial et al., 2020)

S.N	Sample	Mode	Putative compounds	Retention Time(Min)	Mass	Molecular Formula	m/z	Chemical Class	Clinical uses
11			Usambarensine	11.53	450.2845	C ₃₀ H ₃₄ N ₄	449.2773	alkaloid	anti-amoebic and Anti-plasmodial Activities, anti-tumor (Passemar et al., 2011)
12			Corchorifatty acid F	11.63	328.2259	C ₁₈ H ₃₂ O ₅	327.2185	lipid	antioxidant and anti-inflammatory properties (Yoshikawa et al.,1998)
13			Momordicoside G	11.726	632.429	C ₃₇ H ₆₀ O ₈	677.4275	triterpene	Anticancer ((Du et al., 2019)
14			Macrocarpal B	21.138	472.2839	C ₂₈ H ₄₀ O ₆	471.2767	sesquiterpenoid	Anti-bacterial (Nagata et al.,2006)
15			Diepomuricanin A	18.42	546.4667	C ₃₅ H ₆₂ O ₄	591.465	Acetogenins	antibacterial, anticancer, antidiabetic and anti-inflammatory properties (Al Kazman et al., 2022)
16			Canrenone	23.98	340.2079	C ₂₂ H ₂₈ O ₃	339.201	steroid	Anti-diabeticAntiarrhythmic (Armanini et al., 2014, D browski et al.,2020)
17			Etidocaine	24.409	276.2223	C ₁₇ H ₂₈ N ₂ O	335.2363	Amino acid	Anesthetic (Hameroff rt al, 1984)
18			3-keto stearic acid	22.575	298.2518	C ₁₈ H ₃₄ O ₃	297.2446	lipid	anti-inflammatory and anti-cancer (Tiebe et al., 2018)

Table 5 :Identified compounds in UPS using HR-LCMS

S. N	Sample	Mode	Putative compounds	Retention Time(Min)	Mass	Molecular Formula	m/z	Chemical Class	Clinical uses
1		+ve ESI mode	Neuraminic acid	3.662	267.0818	C ₉ H ₁₇ NO ₈	268.1007	glycoprotein	anti-inflammatory, anti-viral, anti-tumor, anti-hypertensive and skin whitening properties [Mingli et al., 2023].
3	Biperiden		4.548	311.226	C ₂₁ H ₂₉ NO	334.2154	alkaloid	Parkinson's disease (Kostelnik et al., 2017)	
4	Dolasetron		5.075	324.1481	C ₁₉ H ₂₀ N ₂ O ₃	347.1376	Indole carboxylic acids (alkaloid)	Anti-nausea, anti-vomitting (Meyer et al., 2005)	
5	Ketamine		5.291	237.0925	C ₁₃ H ₁₆ ClNO	238.0998	cyclohexanone	ulcerative cystitis, neurocognitive impairment, deficits in working and episodic memory [Celia et al., 2011].	
6	Mianserin		5.316	264.1645	C ₁₈ H ₂₀ N ₂	287.1539	dibenzoazepine	Antidepressant (Wakeling, 1983)	
7	Thienamycin		6.416	272.0836	C ₁₁ H ₁₆ N ₂ O ₄ S	295.0728	Non ribosomal peptides	Antibiotic (Reyes et al., 1985)	
8	Sciadopitysin		6.516	580.1313	C ₃₃ H ₂₄ O ₁₀	581.1387	phenolics	anti-inflammatory, anti-oxidant and anti-apoptotic (Ijaz et al.,2023)	
9	Naltrindole		15.774	414.1953	C ₂₆ H ₂₆ N ₂ O ₃	415.2025	alkaloid	anticonvulsant and immunosuppressant properties (Chenet et al., 1997)	

S. N	Sample	Mode	Putative compounds	Retention Time(Min)	Mass	Molecular Formula	m/z	Chemical Class	Clinical uses
10		-ve ESI mode	Syringic acid	4.056	198.0524	C ₉ H ₁₀ O ₅	257.0668	phenolic	anti-inflammatory, hepatoprotective, cardio protective, neuroprotective, antibacterial, antidiabetic, and antiendotoxic effects [Shimsa et al., 2024] and colorectal cancer [Mihanfar et al., 2020].
11			Caffeic acid	8.336	180.0418	C ₉ H ₈ O ₄	179.0345	polyphenol	antioxidant, anti-inflammatory, and anticancer (Tajner-Czopek, 2020)
12			Chlorogenic acid	9.444	354.0954	C ₁₆ H ₁₈ O ₉	353.088	phenolic	anti-inflammatory, anti-oxidant, antibacterial, anti-tumor (Wang et al., 2022)
13			Butin	8.702	272.0692	C ₁₅ H ₁₂ O ₅	317.0679	flavonoid	Antioxidant, anti-tumor (Wu et al., 2022)
14			Sorbose	1.495	180.063	C ₆ H ₁₂ O ₆	179.0557	monosaccharide	Anti-diabetic (Oku et al., 2014)
15			Eudistomin N	1.655	245.9781	C ₁₁ H ₇ BrN ₂	304.9917	alkaloid	Anti-cancer (Yang et al., 2023)
16			Swertiamarin	6.431	374.1189	C ₁₆ H ₂₂ O ₁₀	373.1118	glycoside	hepatoprotective, analgesic, anti-inflammatory, antiarthritis, antidiabetic, antioxidant, neuroprotective and gastroprotective activities (Fadjil et al., 2021)
17			Vanillin	6.987	152.0464	C ₈ H ₈ O ₃	151.0393	phenolic	Antioxidant (Xu et al., 2024)

S. N	Sample	Mode	Putative compounds	Retention Time(Min)	Mass	Molecular Formula	m/z	Chemical Class	Clinical uses
18			Quercetin 3,7-dirhamnoside	8.616	594.161	C ₂₇ H ₃₀ O ₁₁	593.1536	flavonol	anti-inflammatory (He et al., 2023)
19			L-Malic acid	1.49	134.0212	C ₄ H ₆ O ₅	133.0139	Alpha hydroxy acid	antioxidants, disinfectants (Chen et al., 2017)

Table 6 : Identified compounds in UPL using HR-LCMS

S.N	Sample	Mode	Putative compounds	Retention Time(Min)	Mass	Molecular Formula	DB Diff (ppm)	m/z	Chemical Class	Clinical uses	
1			Sultamicillin	7.662	594.1467	C ₂₅ H ₃₀ N ₄ O ₉ S ₂	-2.2	595.1539	Beta lactamase inhibitor antibiotic	Antibiotic (Friedel et al., 1989)	
2			Cyclopamine	7.667	411.3122	C ₂₇ H ₄₁ NO ₂	3.64	434.3013	Alkaloid	anti-cancer (Wilson et al., 2010)	
3			Ginkgolide C	8.69	440.1258	C ₂₀ H ₂₄ O ₁₁	13.82	463.1145	Diterpenoid	strong anti-inflammatory and neuroprotective properties (Hebert et al., 2022)	
4		+ve ESI mode	Disopyramide	10.669	339.2354	C ₂₁ H ₂₉ N ₃ O	-12.82	362.2247	Alkaloid	obstructive hypertrophic cardiomyopathy (Massera et al., 2025)	
5			Spiperone	11.149	395.2013	C ₂₃ H ₂₆ FN ₃ O ₂	-0.96	396.2084	alkaloid	Schizophrenia (Henning et al., 1999)	
6			Tiropamide	11.488	467.3169	C ₂₈ H ₄₁ N ₃ O ₃	-4.59	490.3062	Phenylalanine	Antispasmodic (Takayanagi et al., 1989)	
7			Tubulosine	11.983	475.2831	C ₂₉ H ₃₇ N ₃ O ₃	0.84	476.2904	Alkaloids	Anti breast cancer (Kim et al., 2019)	
8			Retapamulin	13.232	517.3292	C ₃₀ H ₄₇ NO ₄ S	-12.85	518.3366	Terpenes	Anti-bacterial (Tanus et al., 2014)	
9			Somniferine	23.6	608.2513	C ₃₆ H ₃₆ N ₂ O ₇	1.49	609.2587	Alkaloid	Antiviral (Shree et al., 2022)	
10			Pederin	12.716	503.3139	C ₂₅ H ₄₅ NO ₉	-8.79	504.3214	Alkaloid	hemolymph toxin (Kellner and Dettner, 1996)	
11			Kaempferol	11.11	286.048	C ₁₅ H ₁₀ O ₆	-1.04	285.0409	Flavonoid	anti-tumorigenic, antiproliferative, and apoptotic effects (Kaur et al., 2024)	
12	UPL			Luteolin	11.11	286.048	C ₁₅ H ₁₀ O ₆	-1.04	285.0409	flavonoid	anti-inflammatory, anti-proliferative, and antioxidant properties (Fikry et al., 2025)
13				Morindone	12.137	270.0522	C ₁₅ H ₁₀ O ₅	2.16	269.045	phenolic	Anti colorectal cancer (Chee et al., 2024)
14			Anatibant	8.827	710.188	C ₃₄ H ₃₆ Cl ₂ N ₆ O ₅ S	-4.99	709.1803	Amino acid	traumatic brain injury (Shakur et el, 2009)	
15		-ve ESI mode	Tectorigenin	12.416	300.0634	C ₁₆ H ₁₂ O ₆	-0.16	299.0562	Phenolic	hepatoprotective, estrogenic, hypoglycemic and anti-inflammatory activities (Wang et al., 2013)	
16			Hexazinone	18.737	252.155	C ₁₂ H ₂₀ N ₄ O ₂	14.3	297.1535	Triazine heterocyclic compound	Herbicide (Jasemizad and Padhye, 2022)	
17			Amprenavir	23.363	505.226	C ₂₅ H ₃₅ N ₃ O ₆ S	-2.65	564.2396	polyphenolic	HIV-1 protease inhibitor (Arvieux and Tribut, 2005)	
18			Luteolin-4'-O-glucoside	8.742	448.1018	C ₂₁ H ₂₀ O ₁₁	-2.76	447.0943	flavonoid	hyperuricemia and gout (Lin et al., 2018)	
19			Ursodeoxycholic acid 3-sulfate	9.154	472.2545	C ₂₄ H ₄₀ O ₇ S	-10.56	471.2471	steroid	hepatobiliary diseases (Kobayashi et al., 2000)	

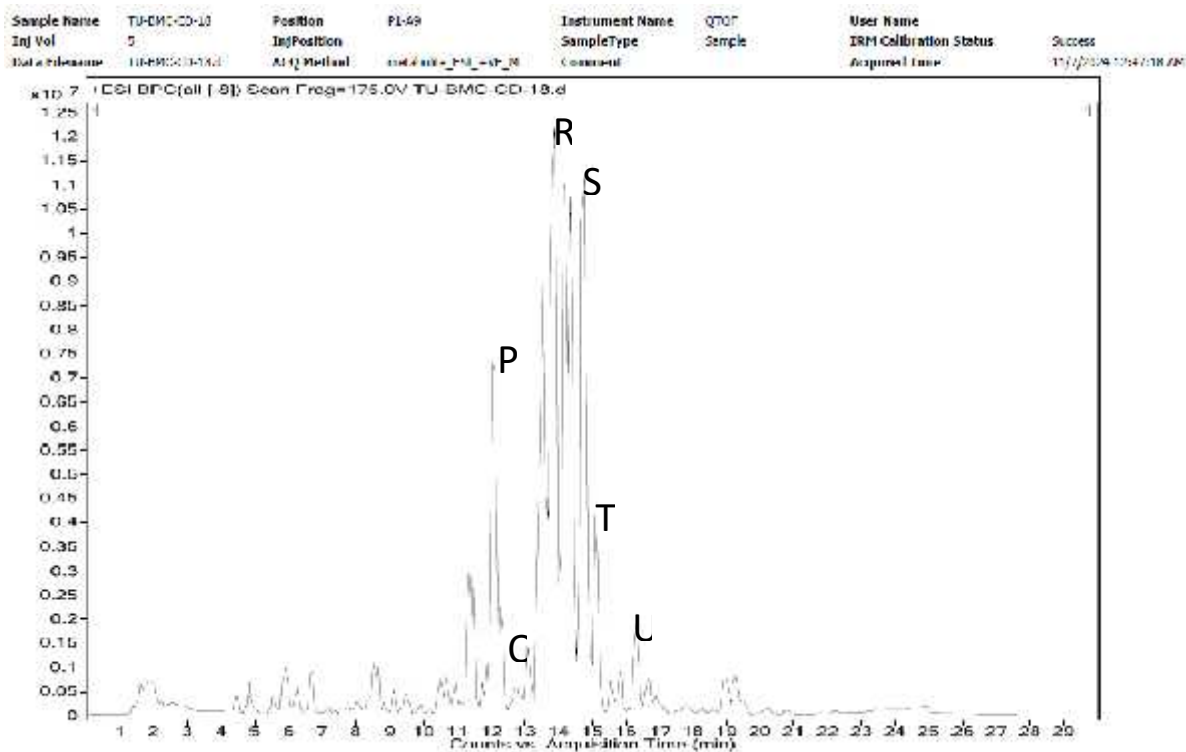


Figure 12 :Chromatogram of UPT in +ve ESI mode using HR-LCMS]

P: Jervine

Q: Funtumine

R: Unknown compound

S: Unknown compound

T: Unknown compound

U: Septentriodine

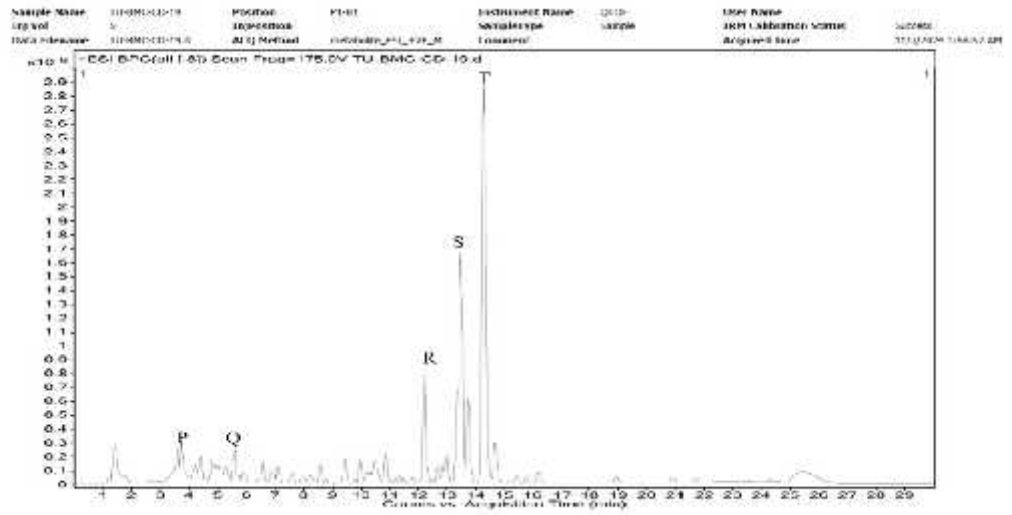


Figure 13 :Chromatogram of UPS in +ve ESI mode]

P: Neuraminic acid

Q: Mianserin

R, S and T: Unknown compounds

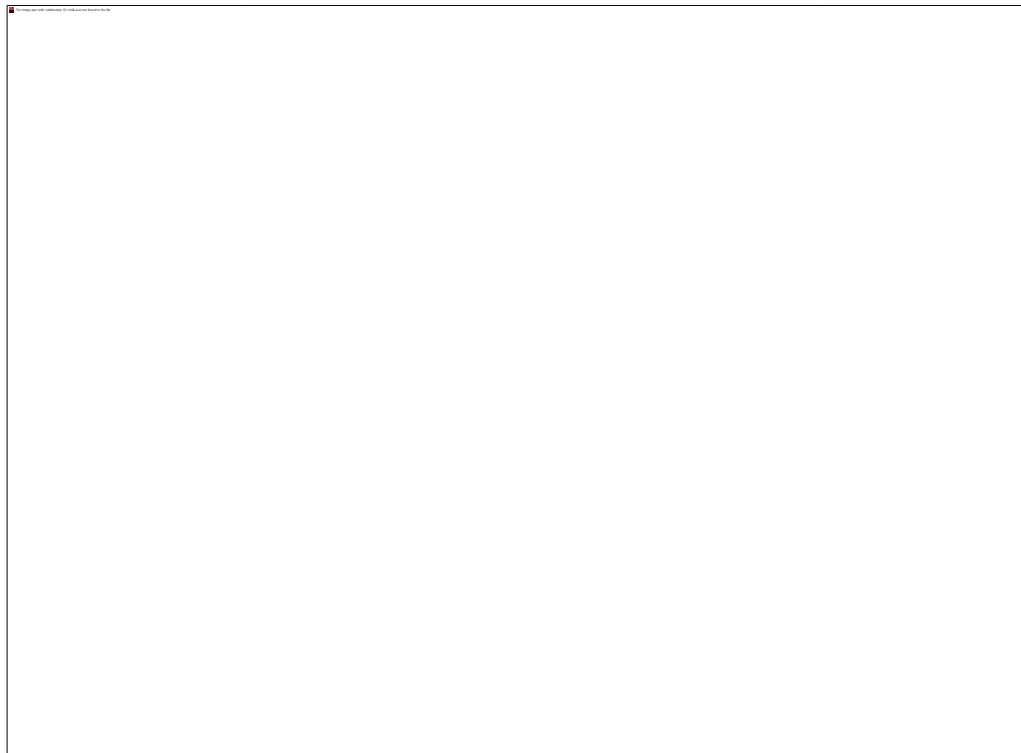


Figure 14 :Chromatogram of UPL in +ESI mode]

P: Abscisic acid glucose ester

Q: 8-pentanoylneosalaniol

R: Unknown compound

S: Unknown compound

T: Jubanine C

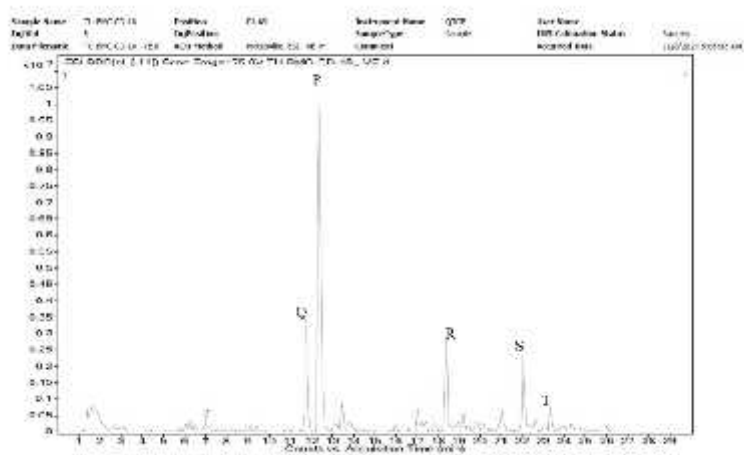


Figure 15 : Chromatogram of UPT in -ve ESI mode]

P: Corchorifatty acid F

Q: 3-Hydroxyquinine

R: Unknown compound

S: Unknown compound

T: Linalyl carprylate

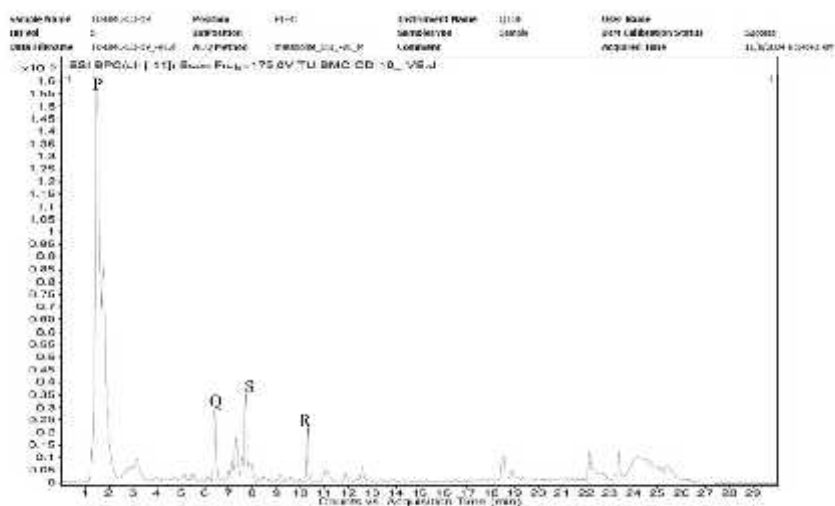


Figure 16 :Chromatogram of UPS in -ve ESI mode

P: Eudistomin N

Q: Swertiamarin

R: Gibberlin A43

S: Luteolin-4'-O-glucoside

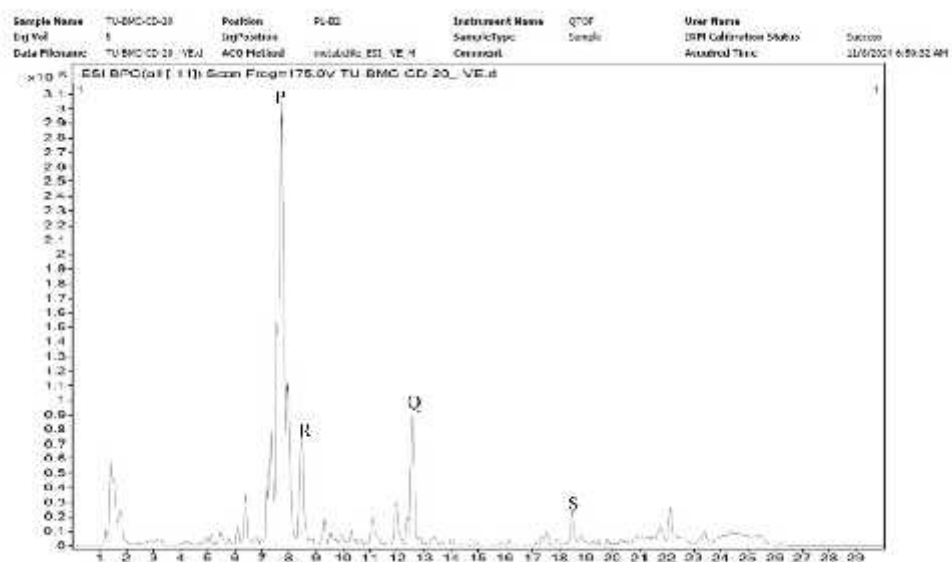


Figure 17 :Chromatogram of UPL in -ESI mode

P: Diosmetin 7-O-beta-D-glucuronopyranoside

Q: 9-chloro-10-hydroxy-hexadecanoic acid

R: Quercetin 3,7-dirhamnoside

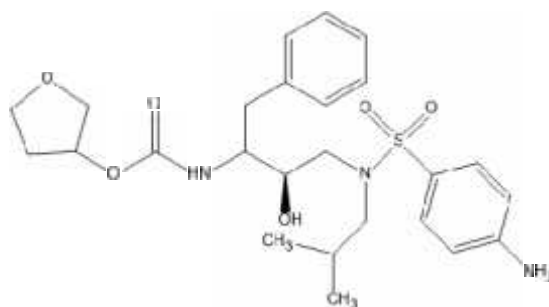
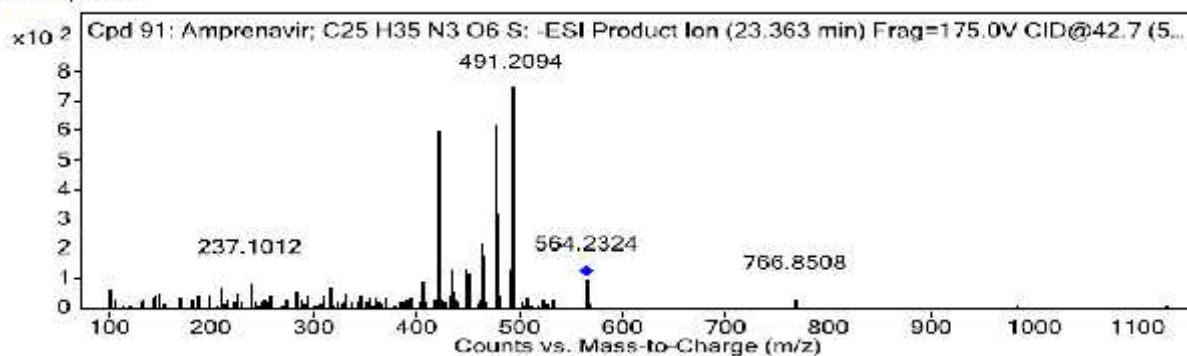
S: Hexazinone

Amprenavir

Amprenavir is a nucleoside reverse transcriptase inhibitors(NRTIs), widely used in amprenavir monotherapy to treat HIV infection which was patented in 1992 and approved by the Food and Drug Administration on April 15, 1999, for clinical use[Stuart et al., 2000].

Compound Label	Name	m/z	RT	Algorithm	Mass
Cpd 91: Amprenavir; C25 H35 N3 O6 S	Amprenavir	564.2396	23.363	Auto MS/MS	505.226

MSMS Spectrum



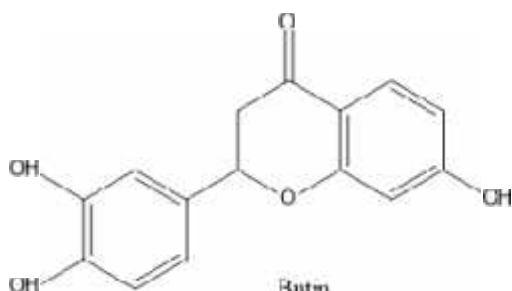
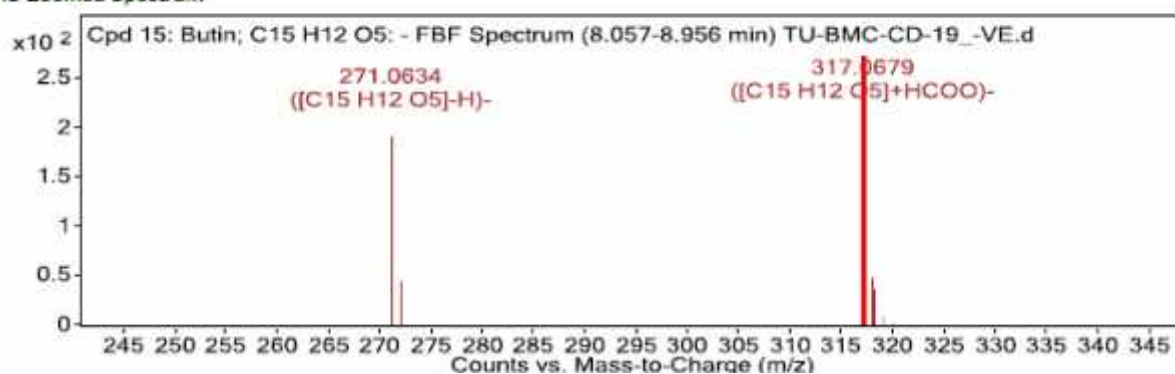
Amprenavir
Chemical Formula: C₂₅H₃₅N₃O₆S
Molecular Weight: 505.63

Butin

Butin is a flavonoid that increases intracellular ROS and DPPH radical scavenging capabilities, giving it antioxidant qualities and a cytoprotective impact against oxidative stress. In cells exposed with H₂O₂, it prevents cellular DNA damage and membrane lipid peroxidation [Zhang et al., 2008]. It can be produced from liquiritigenin through *Bm*TYR-catalyzed hydroxylation [Wu et al., 2022].

Compound Label	Name	<i>m/z</i>	RT	Algorithm	Mass
Cpd 15: Butin; C ₁₅ H ₁₂ O ₅	Butin	317.0679	8.702	Find By Formula	272.0692

MS Zoomed Spectrum



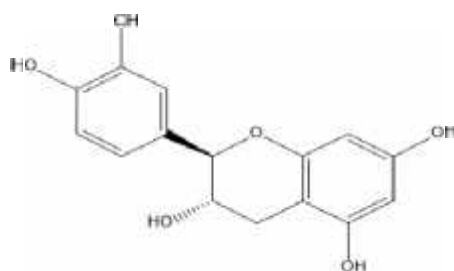
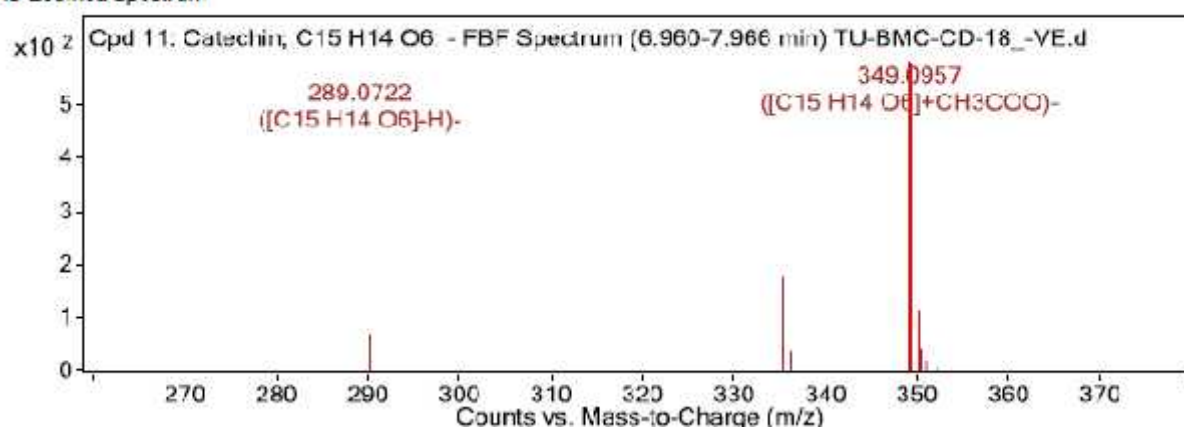
Butin
Chemical Formula: C₁₅H₁₂O₅
Molecular Weight: 272.25

Catechin

T.F.L. Nees von Esenbeck, 1832, successfully isolated crystalline catechin from *Uncaria gambir*. Catechin is well-known for its anti-inflammatory and anti-cancer effects. One powerful characteristic of catechin is its ability to neutralize reactive oxygen and nitrogen species. The prevention of lung, breast, esophagus, stomach, liver, and prostate gland cancers is extremely effective with it [Musial et al., 2020].

Compound Label	Name	m/z	RT	Algorithm	Mass
Cpd 11: Catechin; C15 H14 O6	Catechin	289.0722	7.128	Find By Formula	290.0812

MS Zoomed Spectrum



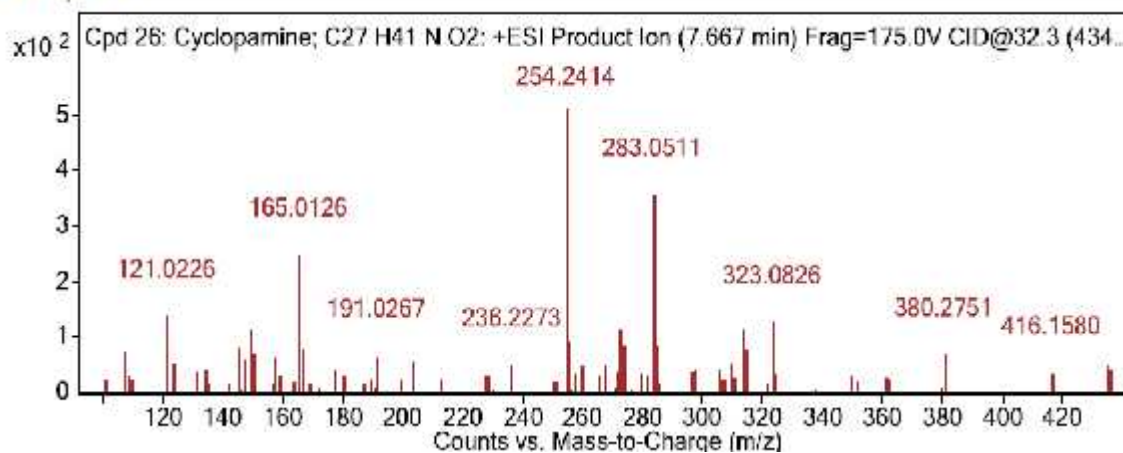
catechin
Chemical Formula: C₁₅H₁₄O₆
Molecular Weight: 290.27

Cyclopamine

The cyclopamine is a terpene alkaloid, isolated from *V. californicum* in 1968. Cyclopamine plays a significant role in treatment of several cancers such as basal cell carcinoma, medullablastoma, and rhabdomyosarcoma by inhibiting the Shh signaling pathway for lethal cancers [Lee et al., 2014]

Compound Label	Name	<i>m/z</i>	RT	Algorithm	Mass
Cpd 26: Cyclopamine; C27 H41 N O2	Cyclopamine	434.3013	7.667	Auto MS/MS	411.3122

MS/MS Spectrum

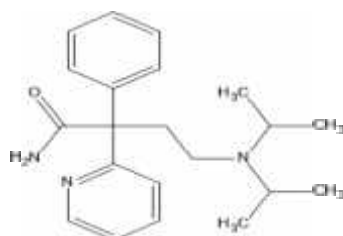
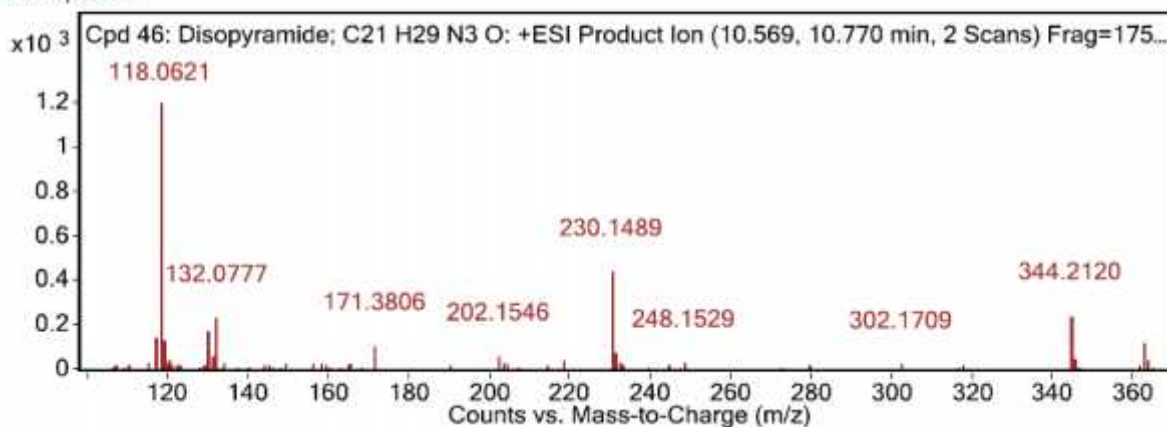


Disopyramide

A sodium channel blocker named disopyramide has an adverse inotropic impact on the ventricular myocardium, which considerably reduces the ability of the heart muscles to contract. It is effective in treating ventricular tachycardia because it inhibits the rise in sodium permeability of cardiac myocyte during Phase 0 of the cardiac action potential with lowering the inward sodium current [Rizos et al., 1987, Kim et al., 1990, Mathur et al., 1972].

Compound Label	Name	m/z	RT	Algorithm	Mass
Cpd 46: Disopyramide; C ₂₁ H ₂₉ N ₃ O	Disopyramide	362.2247	10.669	Auto MS/MS	339.2354

MSMS Spectrum

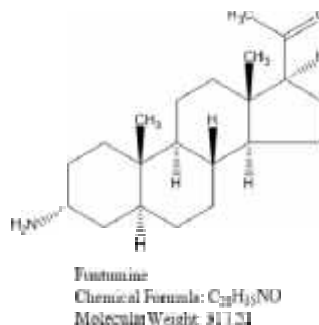
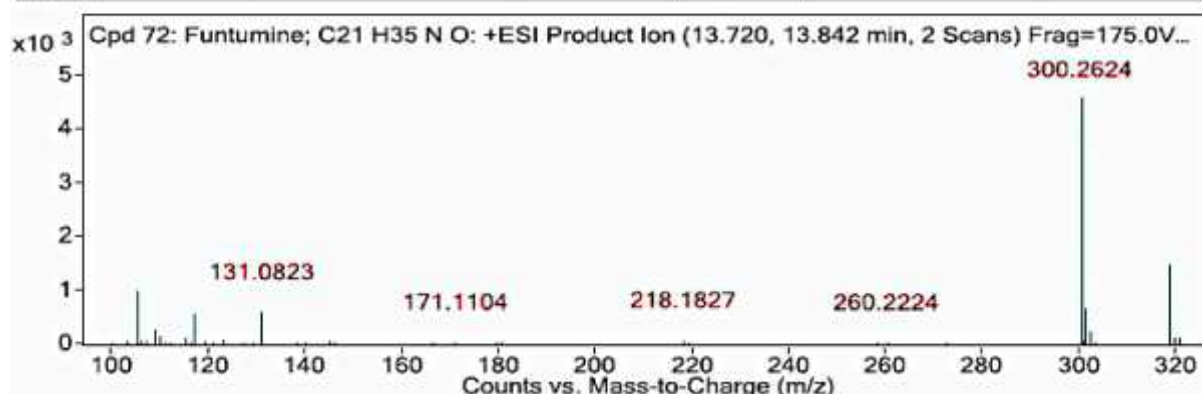


Disopyramide
Chemical Formula: C₂₁H₂₉N₃O
Molecular Weight: 339.47

Funtumine

Funtumine is a steroidal alkaloid which has anti-proliferative mechanism of action on cancer cell lines (HT-29, MCF-7 and HeLa) by exploring the mitochondrial depolarization effects, reactive oxygen species (ROS) induction, apoptosis, F-actin perturbation, and inhibition of topoisomerase-I. [Badmus et al., 2020)

Compound Label	Name	m/z	RT	Algorithm	Mass
Cpd 72: Funtumine; C21 H35 N O	Funtumine	318.2748	13.781	Auto MS/MS	317.2675

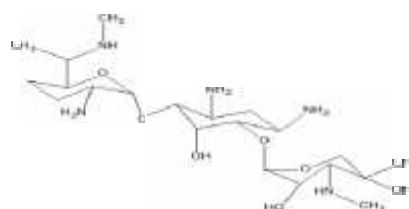
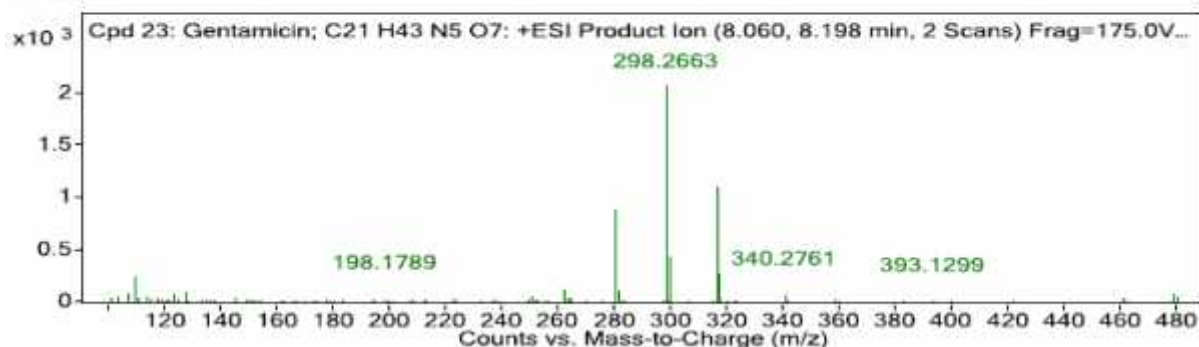


Gentamycin

Weinstein et al., at Schering Corporation in Bloomfield, New Jersey, discovered the aminoglycoside gentamycin from *Micromonospora* in 1963. It is used to treat a number of infections peritonitis from gastrointestinal tract infections, respiratory tract infections, urinary tract infectious diseases caused by *Klebsiella pneumoniae*, *Escherichia coli*, *Serratia marcescens*, *Citrobacter* spp., *Enterobacteriaceae* spp., and *Pseudomonas* spp. Gentamycin may be used as an adjuvant for febrile neutropenia, female genital infection, uterine infection, postnatal infection, necrotizing enterocolitis in fetuses or newborns, osteomyelitis, pelvic inflammatory disease, plague, gonorrhea, tularemia, prophylaxis of post-cholecystectomy infection,

transrectal prostate biopsy, and posttympanostomy-related infection, malignant otitis externa, and intratympanically or transtympanically for Meniere's disease [Chen et al., 2013].

Compound Label	Name	<i>m/z</i>	RT	Algorithm	Mass
Cpd 23: Gentamicin; C21 H43 N5 O7	Gentamicin	478.3281	8.129	Auto MS/MS	477.3206



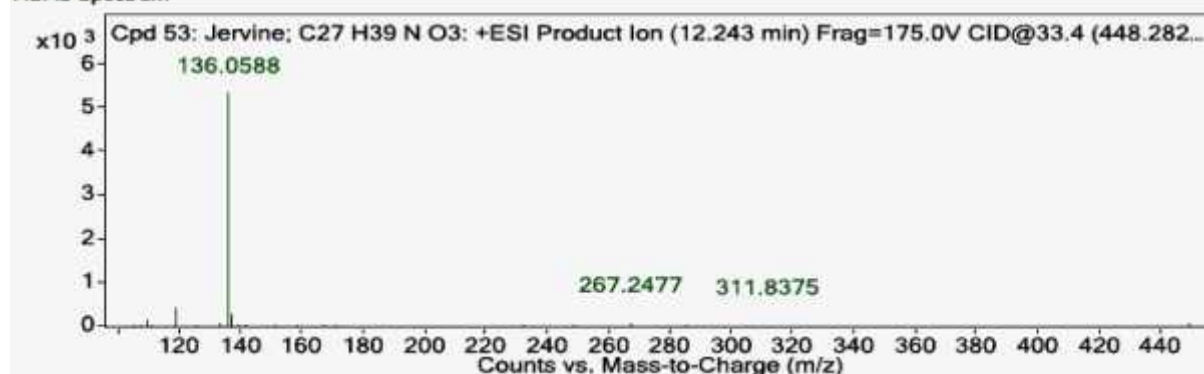
Gentamicin
Chemical Formula: C₂₁H₄₃N₅O₇
Molecular Weight: 477.631

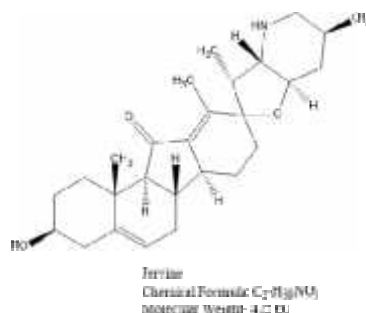
Jervine

Jervine is a steroidal alkaloid identified by Russell Molyneux and James. Jervine is very potent anti-inflammatory agent [Dumlu et al., 2018]. It plays a protective role against radiation-induced gastrointestinal toxicity [Yakan et al., 2019]. Jervine has demonstrated anti-tumor efficacy in non-small cells lung cancer (NSCLC) by reducing proliferation ability of NSCLC cells along with colony formation capacity inhibition [Lei et al., 2020].

Compound Label	Name	<i>m/z</i>	RT	Algorithm	Mass
Cpd 53: Jervine; C27 H39 N O3	Jervine	448.2831	12.243	Auto MS/MS	425.2937

MSMS Spectrum



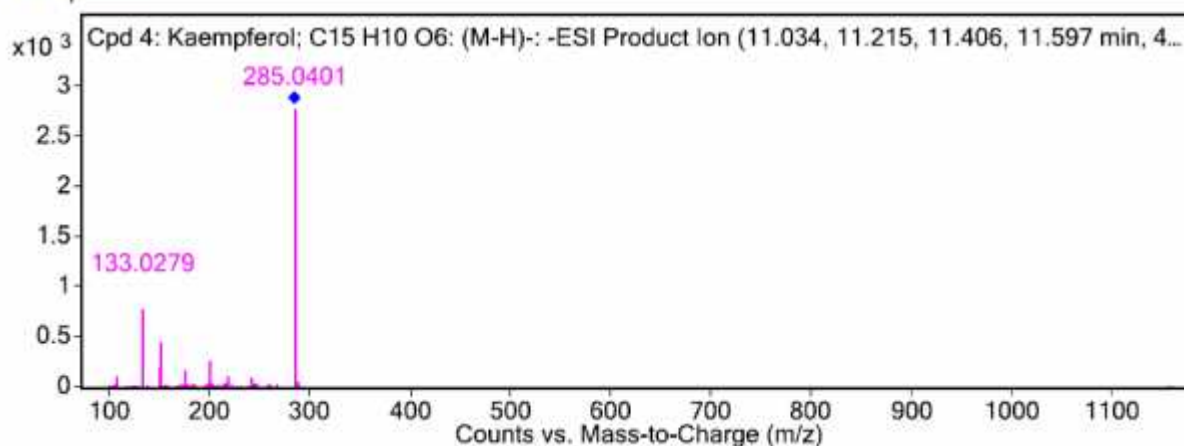


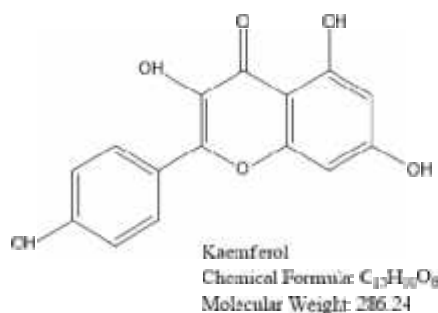
Kaempferol

Kaempferol is a flavon, contains a diphenylpropane structure which make it hydrophobic. Kaempferol is abundantly distributed in plant species of genera Delphinium, Camellia, Berberis, Citrus, Brassica, Allium, Malus, etc. Kaempferol serves as anti-inflammatory agent by modulating pro-inflammatory enzyme activities and gene expression involved in inflammation. It also inhibits transcription factors and is a potent anti-oxidant agent [Devi et al., 2015].

Compound Label	Name	m/z	RT	Algorithm	Mass
Cpd 4: Kaempferol; C15 H10 O6	Kaempferol	285.0409	11.11	Find By Formula	286.048

MSMS Spectrum



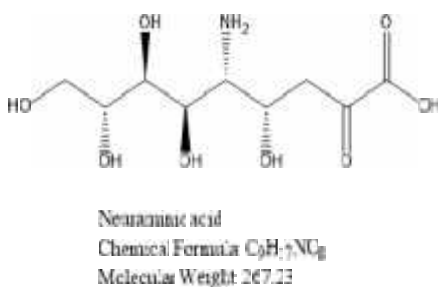
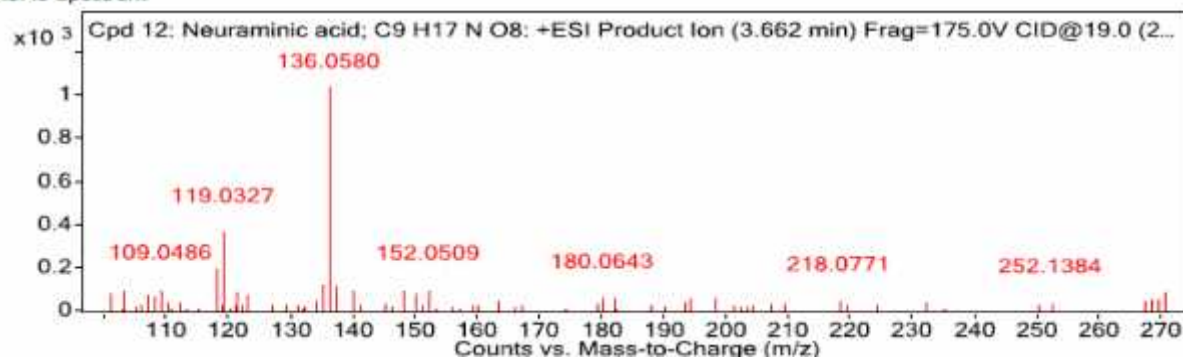


Neuraminic acid

Neuraminic acid is a 9-carbon atom containing monosaccharide which is beneficial in boosting immune, nervous system, gastrointestinal health, brain development. It also has anti-inflammatory, anti-viral, anti-tumor, anti-hypertensive and skin whitening properties [Mingli et al., 2023].

Compound Label	Name	<i>m/z</i>	RT	Algorithm	Mass
Cpd 12: Neuraminic acid; C9 H17 N O8	Neuraminic acid	268.1007	3.662	Auto MS/MS	267.0818

MSMS Spectrum

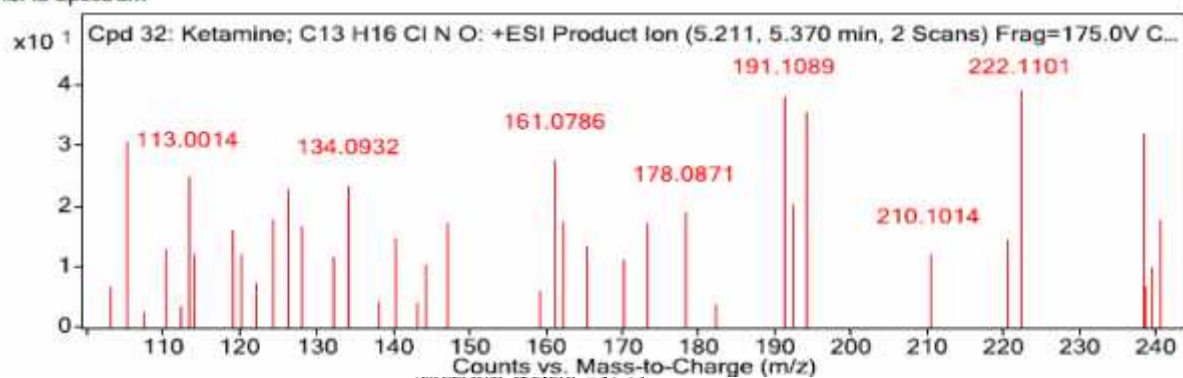


Ketamine

Ketamine, a cyclohexanone, used as anesthesia and pain reliever. The repeated misuse of this molecule has exhibited harmful physical and physiological consequences which includes ulcerative cystitis, neurocognitive impairment, deficits in working and episodic memory [Celia et al., 2011].

Cpd 32: Ketamine; C13 H16 Cl N O	Ketamine	238.0998	5.291	Auto MS/MS	237.0925
-------------------------------------	----------	----------	-------	------------	----------

MSMS Spectrum

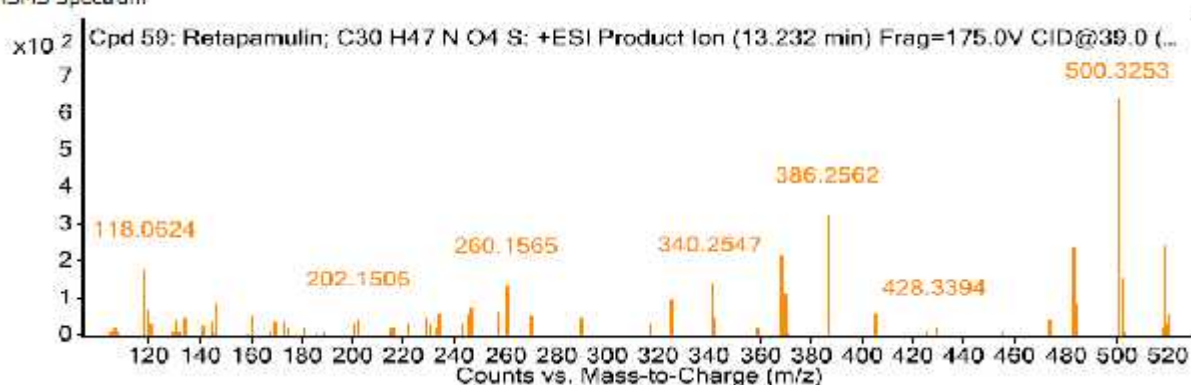


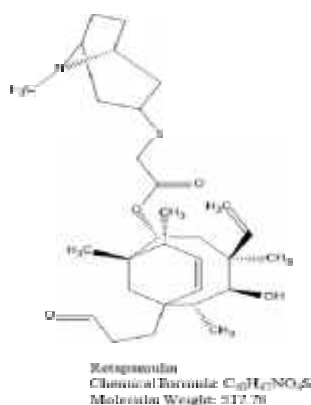
Retapamulin

Retapamulin, a semisynthetic derivative of pleuromutilin, exhibits superior *in vitro* anti-*S. aureus* and anti-*S. pyogenes* action. By attaching to a location on the 50S subunit of the bacterial ribosome, in a manner distinct from that of other ribosomally targeted antibiotics such as macrolides, it specifically prevents the production of proteins by bacteria. In April 2007, the US FDA authorized its topical use for the treatment of bacterial skin infections, such as impetigo, and infected minor cuts, scrapes, or sutured wounds.

Compound Label	Name	<i>m/z</i>	RT	Algorithm	Mass
Cpd 59: Retapamulin; C30 H47 N O4 S	Retapamulin	518.3366	13.232	Auto MS/MS	517.3292

MSMS Spectrum

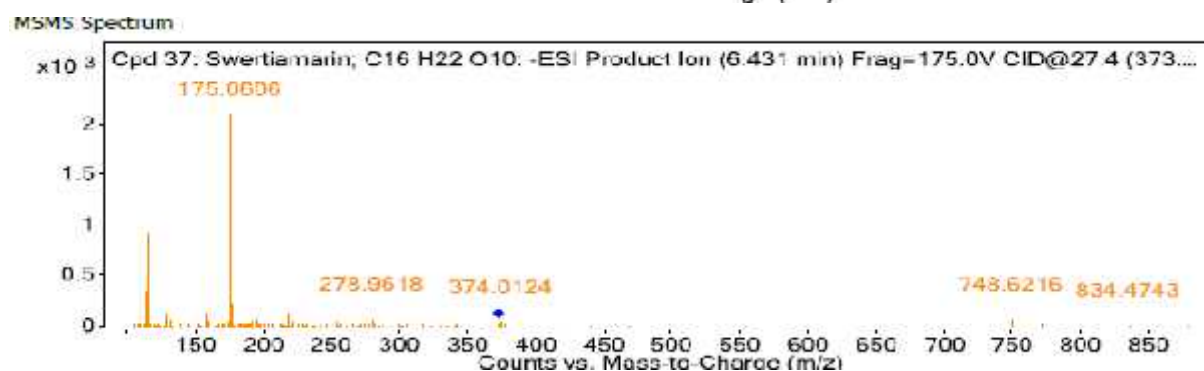
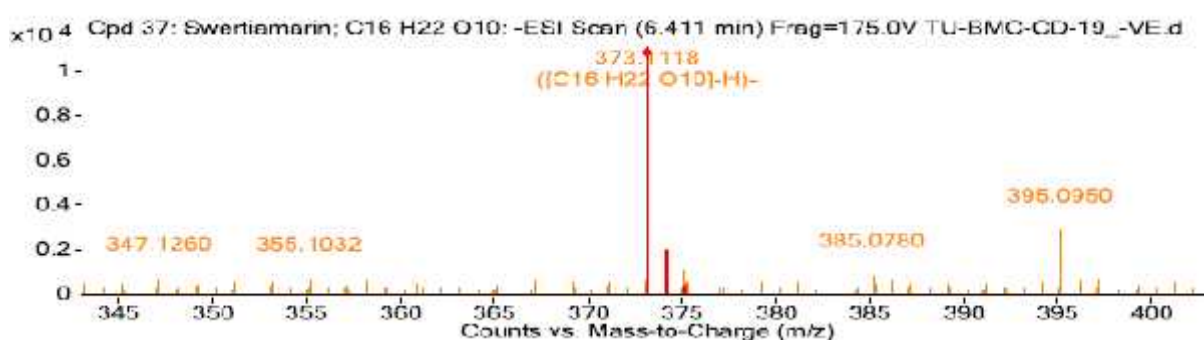


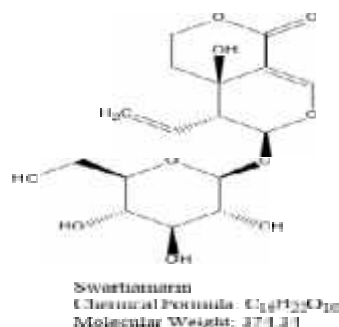


Swertiamarin

A secoiridoid glycoside, swertiamarin, has a variety of biological properties, including anti-inflammatory, antidiabetic, anti-atherosclerotic, and antioxidant actions. Its impact on a number of signaling pathways linked to cardiac remodeling events, apoptosis, inflammatory and lipid peroxidation markers, and the activation of antioxidant enzymes was the primary cause of its activities [Leong et al., 2016].

Compound Label	Name	<i>m/z</i>	RT	Algorithm	Mass
Cpd 37: Swertiamarin; C16 H22 O10	Swertiamarin	373.1118	6.411	Auto MS/MS	374.1189



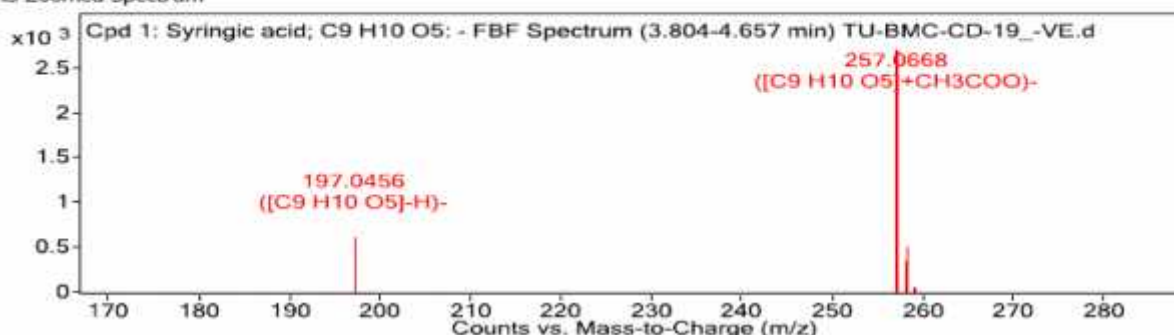


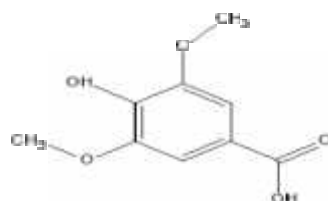
Syringic acid

Syringic acid is one of the most prevalent phenolic acids, member of the hydroxybenzoic acid. Olives, dates, grapes, walnuts, radishes, and pumpkins are the main foods that contain it. The chemical structure of syringic acid exhibits a benzene ring with a hydroxyl (-OH), one carboxylic acid (-COOH), and two methoxy (-OCH₃) groups linked to the ring. The medicinal qualities of syringic acid are attributed due to the presence of methoxy groups on the aromatic ring at positions 3 and 5. Wide variety of pharmacological characteristics are exhibited by syringic acid, such as anti-inflammatory, hepatoprotective, cardioprotective, neuroprotective, antibacterial, antidiabetic, and antiendotoxic effects [Shimsa et al., 2024] and colorectal cancer [Mihanfar et al., 2020].

Compound Label	Name	m/z	RT	Algorithm	Mass
Cpd 1: Syringic acid; C9 H10 O5	Syringic acid	257.0668	4.056	Find By Formula	198.0524

MS Zoomed Spectrum





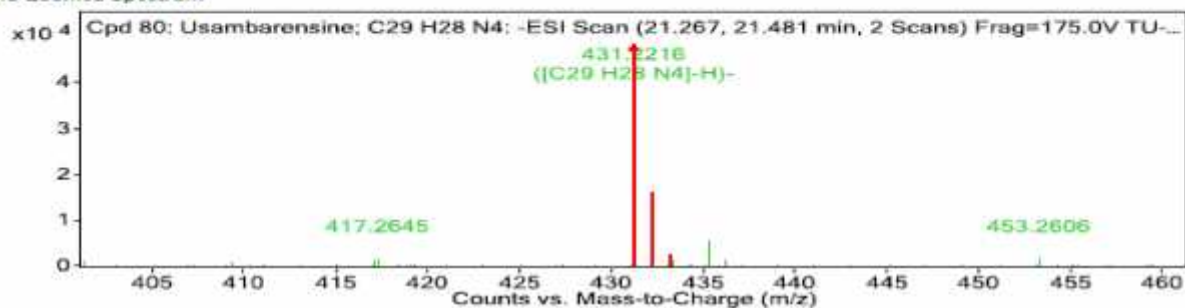
Syringic acid
 Chemical Formula: C₉H₁₀O₅
 Molecular Weight: 198.17

Usambarensine

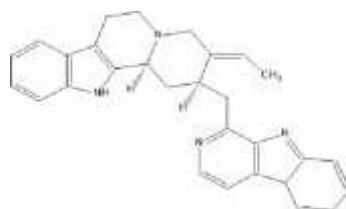
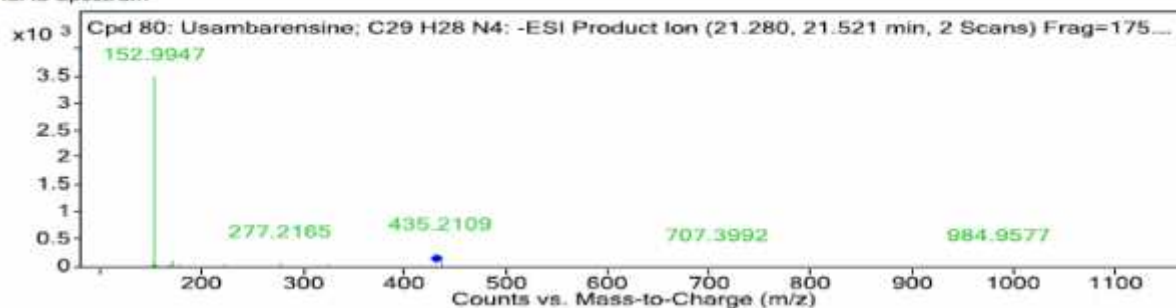
Usambarensine is a tertiary amine alkaloid. Usambarensine has atropine-like and spasmolytic properties, anti-amoebic and Anti-plasmodial Activities [Wright et al., 1991], as antineoplastic drugs for cancer chemotherapy. [Isah,2016] and antiplasmodial activity [Passemar et al., 2011].

Compound Label	Name	m/z	RT	Algorithm	Mass
Cpd 80: Usambarensine; C29 H28 N4	Usambarensine	431.2216	21.401	Auto MS/MS	432.2289

MS Zoomed Spectrum



MSMS Spectrum



Usambarensine
 Chemical Formula: C₂₉H₂₈N₄
 Molecular Weight: 432.50

HR-LCMS profiling of plant samples from Upperdangadhi, screened a large variety of compounds bearing anti-bacterial, anti-viral, anti-fungal, anti-tumor, antioxidant, anti-inflammatory, anti-malarial, anti-helminthic, anti-hypertensive, anti-vomiting, anti-depression and anxiety, anti-apoptosis, purgative properties. Some compounds are potential to treat Parkinson's disease and to enhance mitochondrial normal function. From designed tests and their results, it is obvious that cocoyam can be the best alternative of cereal crops in nutrient supplement and a good source of pharmaceutical products along with other known medicinally rich bios.

CHAPTER 5. CONCLUSIONS AND RECOMMENDATION

5.1 Conclusion

Cocoyam (*Colocasia esculenta*) is potential source of secondary metabolites and nutrient elements beneficial for human health. The chemical constituents present in the extracts prove that it may become the super food for the oppressed society having low economic status. The application of UV-VIS spectrophotometer, HR-LCMS and ICP-AES, a comprehensive analysis of tuber, petiole and leaves revealed the presence of varieties of phytochemicals. TPC and TFC were more in leaf than petiole and tuber (19.96±0.01 mg GAE/g DS, 89.1±0.006 mg QE/g DS). Highest antioxidant activity in Padampur leaf (27.82±0.01 mg AAE/g DS) and palung tuber (IC₅₀ = 240.45 µg/ml). Maximum TCC reported in Upperdangadhi tuber (112±0.0 mg GE/g DS). Significant amounts of essential elements (Ca, Cu, Fe, Mg, Zn) and trace levels of toxic metals like arsenic (As, Cr) were recorded via ICP-AES.

The HR-LCMS analysis of UPT in both positive and negative ESI mode revealed total 200 compounds, among which 60 are known, 140 are unknown. Among known, 11 alkaloids, 4 phenolics, 5 flavonoids, 3 glycosides, 5 steroids and 32 other compounds are present. In the similar way UPS sample contains total 200 compounds, out of which 115 known and 85 are unknown. Out of known compounds, 12 are alkaloids, 15 phenolic, 26 flavonoids, 9 glycosides and 55 other class of compounds. The UPL sample screened total 197 compounds, among them, 89 are known while 116 are unknown. In the core of 89 known compounds: 13 alkaloids, 17 flavonoids, 3 terpenoids, 1 steroid and 57 other compounds. These compounds have demonstrated Anti-tumor, anti-inflammatory, Antioxidant anti-obesity, antidiabetic, antibacterial, anti-amoebic, Anti-plasmodial, Anesthetic anti-hypertensive, skin whitening properties, Anti-nausea, anti-vomiting, Antidepressant properties and also effective in Parkinson's disease.

The findings of this research contribute to the existing body of knowledge in the chemical profile of cocoyam and it might become a valuable source of natural antioxidants. To sum up, Cocoyam is the rich source of varieties of chemical constituents and nutrient elements. Hence it may be potential for the use in processed food, pharmaceuticals, cosmetics etc.

5.2 Recommendations

Based on the findings presented in this thesis on the chemical composition of cocoyam (*Colocasia esculenta*), the following recommendations are put forth for future research and development:

1. Isolation of pure compounds and their characterization.
2. In-silico pharmacokinetic study, molecular docking and their characterization as to noble drug
3. Unidentified compound appeared in the HR-LCMS should be isolated and characterized.

REFERENCES

- Adefegha, S. A. (2018). Impact of pasting on starch composition, estimated glycemic index, phenolic constituents, antioxidant activities and antidiabetic properties of flour produced from cocoyam (*Colocasia esculenta*) corm. *Journal of Food Biochemistry*, 42(4), e12514. <https://doi.org/10.1111/jfbc.12514>
- Ahmad Khan, M. S., & Ahmad, I. (2019). Herbal Medicine. In *New Look to Phytomedicine* (pp. 3–13). Elsevier. <https://doi.org/10.1016/B978-0-12-814619-4.00001-X>
- Ainaz Mihanfar, Saber Ghazizadeh Darband, Shirin Sadighparvar b, Mojtaba Kaviani c, & Mohammad Mirza-Aghazadeh-Attari d,e, Bahman Yousefi f, Maryam Majidinia. (2020). *In vitro and in vivo anticancer effects of syringic acid on colorectal cancer: Possible mechanistic view*. <https://doi.org/10.1016/j.cbi.2020.109337>
- Al Kazman, B. S. M., Harnett, J. E., & Hanrahan, J. R. (2022). Traditional Uses, Phytochemistry and Pharmacological Activities of Annonaceae. *Molecules*, 27(11), 3462. <https://doi.org/10.3390/molecules27113462>
- Alanis, A. J. (2005). Resistance to Antibiotics: Are We in the Post-Antibiotic Era? *Archives of Medical Research*, 36(6), 697–705. <https://doi.org/10.1016/j.arcmed.2005.06.009>
- Al-Gubory, K. H., Fowler, P. A., & Garrel, C. (2010). The roles of cellular reactive oxygen species, oxidative stress and antioxidants in pregnancy outcomes. *The International Journal of Biochemistry & Cell Biology*, 42(10), 1634–1650. <https://doi.org/10.1016/j.biocel.2010.06.001>
- Armanini, D., Sabbadin, C., Donà, G., Clari, G., & Bordin, L. (2014). Aldosterone receptor blockers spironolactone and canrenone: Two multivalent drugs. *Expert Opinion on Pharmacotherapy*, 15(7), 909–912. <https://doi.org/10.1517/14656566.2014.896901>
- Arroyo, G., Bustos, J. A., Lescano, A. G., Gonzales, I., Saavedra, H., Rodriguez, S., Pretell, E. J., Bonato, P. S., Lanchote, V. L., Takayanagui, O. M., Horton, J., Gonzalez, A. E., Gilman, R. H., Garcia, H. H., Cysticercosis Working Group in Peru, Tsang, V. C. W., O’Neal, S., Martinez, M., Zimic, M., ... Friedland, J. (2019). Albendazole Sulfoxide Plasma Levels and Efficacy of Antiparasitic Treatment in Patients With Parenchymal

- Neurocysticercosis. *Clinical Infectious Diseases*, 69(11), 1996–2002.
<https://doi.org/10.1093/cid/ciz085>
- Arvieux, C., & Tribut, O. (2005). Amprenavir or Fosamprenavir plus Ritonavir in HIV Infection: Pharmacology, Efficacy and Tolerability Profile. *Drugs*, 65(5), 633–659.
<https://doi.org/10.2165/00003495-200565050-00005>
- Assi, M. A., Hezmee, M. N. M., Haron, A. W., Sabri, M. Y., & Rajion, M. A. (2016). The detrimental effects of lead on human and animal health. *Veterinary World*, 9(6), 660–671. <https://doi.org/10.14202/vetworld.2016.660-671>
- Awa, E., & Eleazu, C. (2015). Bioactive constituents and antioxidant activities of raw and processed cocoyam (*Colocasia esculenta*): Chemical composition of raw and processed cocoyam. *Nutrafoods*, 14(3), 133–140. <https://doi.org/10.1007/s13749-015-0033-x>
- Badmus, J. A., Ekpo, O. E., Sharma, J. R., Sibuyi, N. R. S., Meyer, M., Hussein, A. A., & Hiss, D. C. (2020). An Insight into the Mechanism of Holamine- and Funtumine-Induced Cell Death in Cancer Cells. *Molecules*, 25(23), 5716. <https://doi.org/10.3390/molecules25235716>
- Baek, J., & Lee, M.-G. (2016). Oxidative stress and antioxidant strategies in dermatology. *Redox Report*, 21(4), 164–169. <https://doi.org/10.1179/1351000215Y.0000000015>
- Balouiri, M., Sadiki, M., & Ibsouda, S. K. (2016). Methods for in vitro evaluating antimicrobial activity: A review. *Journal of Pharmaceutical Analysis*, 6(2), 71–79. <https://doi.org/10.1016/j.jpha.2015.11.005>
- Bansal, A. K., & Bilaspuri, G. S. (2011). Impacts of Oxidative Stress and Antioxidants on Semen Functions. *Veterinary Medicine International*, 2011, 1–7. <https://doi.org/10.4061/2011/686137>
- Barbara Holmes and Eugene M. Sorkin. (1986). Indoramin A Review of its Pharmacodynamic and Pharmacokinetic Properties, and Therapeutic Efficacy in Hypertension and Related Vascular, Cardiovascular and Airway Diseases. *ADIS Drug Information Services, Drugs 31 : 467-499 (1986)*.

- Bedekar, A., Shah, K., & Koffas, M. (2010). Natural Products for Type II Diabetes Treatment. In *Advances in Applied Microbiology* (Vol. 71, pp. 21–73). Elsevier. [https://doi.org/10.1016/S0065-2164\(10\)71002-9](https://doi.org/10.1016/S0065-2164(10)71002-9)
- Bernardini, S., Tiezzi, A., Laghezza Masci, V., & Ovidi, E. (2018). Natural products for human health: An historical overview of the drug discovery approaches. *Natural Product Research*, 32(16), 1926–1950. <https://doi.org/10.1080/14786419.2017.1356838>
- Beutler, J. A. (2009). Natural Products as a Foundation for Drug Discovery. *Current Protocols in Pharmacology*, 46(1). <https://doi.org/10.1002/0471141755.ph0911s46>
- Bjørklund, G., & Chirumbolo, S. (2017). Role of oxidative stress and antioxidants in daily nutrition and human health. *Nutrition*, 33, 311–321. <https://doi.org/10.1016/j.nut.2016.07.018>
- Boakye, A. A., Wireko Manu, F. D., Oduro, I., Ellis, W. O., Gudjónsdóttir, M., & Chronakis, I. S. (2018). Utilizing cocoyam (*Xanthosoma sagittifolium*) for food and nutrition security: A review. *Food Science & Nutrition*, 6(4), 703–713. <https://doi.org/10.1002/fsn3.602>
- Ce, O., Pc, A., & Ck, E. (2017). Studies on the Flowers and Stems of Two Cocoyam Varieties: *Xanthosoma sagittifolium* and *Colocasia esculenta*. *Natural Products Chemistry & Research*, 05(06). <https://doi.org/10.4172/2329-6836.1000263>
- Chang, Chia-Chi, Yang, Ming-Hua, Wen, Hwei-Mei, & Chern, Jiing-Chuan. (2002). Estimation of total flavonoid content in propolis by two complementary colorimetric methods. *Journal of Food and Drug Analysis*, 10(3), 178–182.
- Chee, C. W., Mohd Hashim, N., & Nor Rashid, N. (2024). Morindone as a potential therapeutic compound targeting TP53 and KRAS mutations in colorectal cancer cells. *Chemico-Biological Interactions*, 392, 110928. <https://doi.org/10.1016/j.cbi.2024.110928>
- Chen, C., Chen, Y., Wu, P., & Chen, B. (2014). Update on new medicinal applications of gentamicin: Evidence-based review. *Journal of the Formosan Medical Association*,

113(2), 72–82. <https://doi.org/10.1016/j.jfma.2013.10.002>

- Chen, X., Wang, Y., Dong, X., Hu, G., & Liu, L. (2017). Engineering rTCA pathway and C4-dicarboxylate transporter for l-malic acid production. *Applied Microbiology and Biotechnology*, 101(10), 4041–4052. <https://doi.org/10.1007/s00253-017-8141-8>
- Chen, Y., & Kirchmair, J. (2020). Cheminformatics in Natural Product based Drug Discovery. *Molecular Informatics*, 39(12), 2000171. <https://doi.org/10.1002/minf.202000171>
- Chen, Y. L., Law, P. Y., & Loh, H. H. (2004). Inhibition of Akt/Protein Kinase B Signaling by Naltrindole in Small Cell Lung Cancer Cells. *Cancer Research*, 64(23), 8723–8730. <https://doi.org/10.1158/0008-5472.CAN-03-3091>
- Cheuka, P., Mayoka, G., Mutai, P., & Chibale, K. (2016). The Role of Natural Products in Drug Discovery and Development against Neglected Tropical Diseases. *Molecules*, 22(1), 58. <https://doi.org/10.3390/molecules22010058>
- D browski, R., Syska, P., M czy ska, J., Farkowski, M., Sawicki, S., Kubaszek-Kornatowska, A., Michałek, P., Kowalik, I., Szwed, H., & Hryniewiecki, T. (2020). Clinical efficacy of potassium canreonate-canrenone in sinus rhythm restoration among patients with atrial fibrillation—A protocol of a pilot, randomized, double-blind, placebo-controlled study (CANREN-AF trial). *Trials*, 21(1), 397. <https://doi.org/10.1186/s13063-020-04277-3>
- Daum, R. S., Kar, S., & Kirkpatrick, P. (2007). Retapamulin. *Nature Reviews Drug Discovery*, 6(11), 865–866. <https://doi.org/10.1038/nrd2442>
- David-Chukwu, N., Amadi, A., Onuabuchi, I., & Chukwu, M. (2022). Influence of storage life and variety on the micro-nutrient compositions of stored cocoyam-based products. *Food Therapy and Health Care*, 4(2), 9. <https://doi.org/10.53388/FTHC20220501009>
- Dias, D. A., Urban, S., & Roessner, U. (2012). A Historical Overview of Natural Products in Drug Discovery. *Metabolites*, 2(2), 303–336. <https://doi.org/10.3390/metabo2020303>
- Domingo, J. L., & Marquès, M. (2021). The effects of some essential and toxic

- metals/metalloids in COVID-19: A review. *Food and Chemical Toxicology*, *152*, 112161. <https://doi.org/10.1016/j.fct.2021.112161>
- Du, Z., Zhang, S., Lin, Y., Zhou, L., Wang, Y., Yan, G., Zhang, M., Wang, M., Li, J., Tong, Q., Duan, Y., & Du, G. (2019). Momordicoside G Regulates Macrophage Phenotypes to Stimulate Efficient Repair of Lung Injury and Prevent Urethane-Induced Lung Carcinoma Lesions. *Frontiers in Pharmacology*, *10*, 321. <https://doi.org/10.3389/fphar.2019.00321>
- Dumlu, F. A., Aydin, T., Odabasoglu, F., Berktaş, O. A., Kutlu, Z., Erol, H. S., Halici, M. B., Cadirci, E., & Cakir, A. (2019). Anti-inflammatory and antioxidant properties of jervine, a steroidal alkaloid from rhizomes of *Veratrum album*. *Phytomedicine*, *55*, 191–199. <https://doi.org/10.1016/j.phymed.2018.06.035>
- Eleazu, C. O. (2016). Characterization of the natural products in cocoyam (*Colocasia esculenta*) using GC–MS. *Pharmaceutical Biology*, *54*(12), 2880–2885. <https://doi.org/10.1080/13880209.2016.1190383>
- Elif Acar Arslan, MD,* Erhan Arslan, MD,† Anil Kılınc, MD,‡ and Özkan Göksu, PhD§. (2017). Effect of Biperiden Treatment in Acute Orofacial and Extremity Dyskinesia With Methylphenidate Therapy. *Pediatric Emergency Care*, Volume 00(Number 00). www.pec-online.com
- Esra Birben PhD,¹ Umit Murat Sahiner MD,¹ Cansin Sackesen MD,¹ & Serpil Erzurum MD,² and Omer Kalayci, MD¹. (2012). Oxidative Stress and Antioxidant Defense. *WAO Journal*.
- Ezeonu, C. S., & Ejikeme, C. M. (2016). Qualitative and Quantitative Determination of Phytochemical Contents of Indigenous Nigerian Softwoods. *New Journal of Science*, *2016*, 1–9. <https://doi.org/10.1155/2016/5601327>
- Fair, R. J., & Tor, Y. (2014). Antibiotics and Bacterial Resistance in the 21st Century. *Perspectives in Medicinal Chemistry*, *6*, PMC.S14459. <https://doi.org/10.4137/PMC.S14459>
- Ferreres, F., Gonçalves, R. F., Gil-Izquierdo, A., Valentão, P., Silva, A. M. S., Silva, J. B.,

- Santos, D., & Andrade, P. B. (2012). Further Knowledge on the Phenolic Profile of *Colocasia esculenta* (L.) Shott. *Journal of Agricultural and Food Chemistry*, 60(28), 7005–7015. <https://doi.org/10.1021/jf301739q>
- Fikry, H., Saleh, L. A., Sadek, D. R., & Alkhalek, H. A. A. (2025). The possible protective effect of luteolin on cardiovascular and hepatic changes in metabolic syndrome rat model. *Cell and Tissue Research*, 399(1), 27–60. <https://doi.org/10.1007/s00441-024-03927-1>
- Foti, M. C. (2015). Use and Abuse of the DPPH[•] Radical. *Journal of Agricultural and Food Chemistry*, 63(40), 8765–8776. <https://doi.org/10.1021/acs.jafc.5b03839>
- Foyer, C. H., & Shigeoka, S. (2011). Understanding Oxidative Stress and Antioxidant Functions to Enhance Photosynthesis. *Plant Physiology*, 155(1), 93–100. <https://doi.org/10.1104/pp.110.166181>
- Frederick M. Kahan, Helmut Kropp, Jon G. Sundelof and Jerome Birnbaum. (1983). Thienamycin: Development of imipenem-cilastatin. *Journal of Antimicrobial Chemotherapy*, 12, 1–35.
- Friedel, H. A., Campoli-Richards, D. M., & Goa, K. L. (1989). Sultamicillin: A Review of its Antibacterial Activity, Pharmacokinetic Properties and Therapeutic Use. *Drugs*, 37(4), 491–522. <https://doi.org/10.2165/00003495-198937040-00005>
- Ganesan, A. (2008). The impact of natural products upon modern drug discovery. *Current Opinion in Chemical Biology*, 12(3), 306–317. <https://doi.org/10.1016/j.cbpa.2008.03.016>
- Granot, E., & Kohen, R. (2004). Oxidative stress in childhood—In health and disease states. *Clinical Nutrition*, 23(1), 3–11. [https://doi.org/10.1016/S0261-5614\(03\)00097-9](https://doi.org/10.1016/S0261-5614(03)00097-9)
- Gupta, R. K., Patel, A. K., Shah, N., Choudhary, A. K., Jha, U. K., Yadav, U. C., Gupta, P. K., & Pakuwal, U. (2014). Oxidative Stress and Antioxidants in Disease and Cancer: A Review. *Asian Pacific Journal of Cancer Prevention*, 15(11), 4405–4409. <https://doi.org/10.7314/APJCP.2014.15.11.4405>

- Ha, H.-L. (2010). Oxidative stress and antioxidants in hepatic pathogenesis. *World Journal of Gastroenterology*, *16*(48), 6035. <https://doi.org/10.3748/wjg.v16.i48.6035>
- Halliwell, B. (2014). Cell culture, oxidative stress, and antioxidants: Avoiding pitfalls. *Biomedical Journal*, *0*(0), 0. <https://doi.org/10.4103/2319-4170.128725>
- Harvey, A. (2008). Natural products in drug discovery. *Drug Discovery Today*, *13*(19–20), 894–901. <https://doi.org/10.1016/j.drudis.2008.07.004>
- Harvey, A. L., Edrada-Ebel, R., & Quinn, R. J. (2015). The re-emergence of natural products for drug discovery in the genomics era. *Nature Reviews Drug Discovery*, *14*(2), 111–129. <https://doi.org/10.1038/nrd4510>
- Hassan, A., Mekhael, Maged K. G., Hanna, A., Simon, A., Tóth, G., & Duddeck, H. (2019). Phytochemical investigation of *Corchorus olitorius* and *Corchorus capsularis* (Family Tiliaceae) that grow in Egypt. *Egyptian Pharmaceutical Journal*, *0*(0), 0. https://doi.org/10.4103/epj.epj_51_18
- Hauser, R. A., Isaacson, S., & Clinch, T. (2014). Randomized, placebo-controlled trial of trimethobenzamide to control nausea and vomiting during initiation and continued treatment with subcutaneous apomorphine injection. *Parkinsonism & Related Disorders*, *20*(11), 1171–1176. <https://doi.org/10.1016/j.parkreldis.2014.08.010>
- He, X., Sun, Y., Lu, X., Yang, F., Li, T., Deng, C., Song, J., & Huang, X. (2023). Assessment of the anti-inflammatory mechanism of quercetin 3,7 dirhamnoside using an integrated pharmacology strategy. *Chemical Biology & Drug Design*, *102*(6), 1534–1552. <https://doi.org/10.1111/cbdd.14346>
- Hébert, M., Bellavance, G., & Barriault, L. (2022). Total Synthesis of Ginkgolide C and Formal Syntheses of Ginkgolides A and B. *Journal of the American Chemical Society*, *144*(39), 17792–17796. <https://doi.org/10.1021/jacs.2c08351>
- Henning, U., Krieger, K., & Klimke, A. (1999). Specific binding of 3h-spiroperone to peripheral blood cells: Relevance for the interpretation of binding studies in psychiatric disorders. *Progress in Neuro-Psychopharmacology and Biological Psychiatry*, *23*(2), 225–241. [https://doi.org/10.1016/S0278-5846\(98\)00104-3](https://doi.org/10.1016/S0278-5846(98)00104-3)

- Igbinoba, S., Onyeji, C., & Akanmu, M. (2016). Modulation of cytochrome P450 3A4 mediated quinine metabolism in healthy volunteers by two honey samples from different floral and geographical sources. *International Journal of Basic and Clinical Pharmacology*, 823–828. <https://doi.org/10.18203/2319-2003.ijbcp20161528>
- Ijaz, M. U., Qamer, M., Hamza, A., Ahmed, H., Afsar, T., Abulmeaty, M., Ayub, A., & Razak, S. (2023). RETRACTED ARTICLE: Sciadopitysin mitigates spermatological and testicular damage instigated by paraquat administration in male albino rats. *Scientific Reports*, 13(1), 19753. <https://doi.org/10.1038/s41598-023-46898-z>
- Isah, T. (2016). Anticancer alkaloids from trees: Development into drugs. *Pharmacognosy Reviews*, 10(20), 90. <https://doi.org/10.4103/0973-7847.194047>
- James Q. Del Rosso, DO; Joseph Bikowski, MD. (2008). Trolamine-Containing Topical Emulsion: Clinical Applications in Dermatology. *Drug Therapy Topics*, 81, 209–214.
- Jasemizad, T., & Padhye, L. P. (2022). Photodegradation and adsorption of hexazinone in aqueous solutions: Removal efficiencies, kinetics, and mechanisms. *Environmental Science and Pollution Research*, 29(32), 48330–48339. <https://doi.org/10.1007/s11356-022-19205-y>
- Jenkins, R. R. (1993). Exercise, Oxidative Stress, and Antioxidants: A Review. *International Journal of Sport Nutrition*, 3(4), 356–375. <https://doi.org/10.1123/ijns.3.4.356>
- Jensen, S. R., Li, H.-Q., Albach, D. C., & Gotfredsen, C. H. (2008). Phytochemistry and molecular systematics of *Trienophora rupestris* and *Oreosolen wattii* (Scrophulariaceae). *Phytochemistry*, 69(11), 2162–2166. <https://doi.org/10.1016/j.phytochem.2008.05.010>
- Jiang, L., Liu, L., Chen, H., Zhang, W., He, L., & Zeng, X. (2022). Effects of autochthonous starter cultures on bacterial communities and metabolites during fermentation of Yu jiangsuan, a Chinese traditional fermented condiment. *LWT*, 168, 113874. <https://doi.org/10.1016/j.lwt.2022.113874>
- Jie Bi, Qingli Yang^{1*}, Jie Sun, Jian Chen and Juan Zhang. (2011). Study on Ultrasonic Extraction Technology and Oxidation Resistance of Total Flavonoids from Peanut

Hull. *Food Sci. Technol. Res.*, 17(3), 187–198.

JOS~E E. LEYSEN, W. GOMMEREN and PIERRE M. LADURON. (1978). SPIPERONE: A LIGAND OF CHOICE RECEPTORS 1. KINETICS AND CHARACTERISTICS OF FOR NEUROLEPTIC IN VITRO BINDING. *BmchemlcaPlh Armacology*, 21, 307~ 316.

Joseph Onyeka. (2014). Status of Cocoyam (*Colocasia esculenta* and *Xanthosoma* spp) in West and Central Africa: Production, Household Importance and the Threat from Leaf Blight. *CGIAR Research Program on Roots, Tubers and Bananas (RTB)*.

Joshi, A., Sharma, V., Singh, J., & Kaushik, V. (2022). Chemi-Informatic Approach to Investigate Putative Pharmacoactive Agents of Plant Origin to Eradicate COVID-19. *Coronaviruses*, 3(3), e170322188687. <https://doi.org/10.2174/2666796701999201203210036>

Julia A. Balfour and Karen L. Goa. (1997). Dolasetron A Review of its Pharmacology and Therapeutic Potential in the Management of Nausea and Vomiting Induced by Chemotherapy, Radiotherapy or Surgery. *ADIS DRUG EVALUATION, Drug s 1997 Aug*; 54 (2); 273-298.

Kasi Pandima Devia, Dicson Sheeja Malara, Seyed Fazel Nabavib, Antoni Suredac, & Jianbo Xiaod,e, Seyed Mohammad Nabavib,**,1, Maria Dagliaf*,1. (2015). Kaempferol and inflammation: From chemistry to medicine. *Pharmacological Research*.

Kaur, J., Gulati, M., Singh, S. K., Kuppusamy, G., Kapoor, B., Mishra, V., Gupta, S., Arshad, M. F., Porwal, O., Jha, N. K., Chaitanya, M. V. N. L., Chellappan, D. K., Gupta, G., Gupta, P. K., Dua, K., Khursheed, R., Awasthi, A., & Corrie, L. (2022). Discovering multifaceted role of vanillic acid beyond flavours: Nutraceutical and therapeutic potential. *Trends in Food Science & Technology*, 122, 187–200. <https://doi.org/10.1016/j.tifs.2022.02.023>

Kaur, S., Mendonca, P., & Soliman, K. F. A. (2024). The Anticancer Effects and Therapeutic Potential of Kaempferol in Triple-Negative Breast Cancer. *Nutrients*, 16(15), 2392. <https://doi.org/10.3390/nu16152392>

- Kellner, R. L. L., & Dettner, K. (1996). Differential efficacy of toxic pederin in deterring potential arthropod predators of *Paederus* (Coleoptera: Staphylinidae) offspring. *Oecologia*, *107*(3), 293–300. <https://doi.org/10.1007/BF00328445>
- Kim, B.-H., Yi, E. H., Li, Y.-C., Park, I.-C., Park, J. Y., & Ye, S.-K. (2019). Anticancer Activity of Tubulosine through Suppression of Interleukin-6-Induced Janus Kinase 2/Signal Transducer and Activation of Transcription 3 Signaling. *Journal of Breast Cancer*, *22*(3), 362. <https://doi.org/10.4048/jbc.2019.22.e34>
- Kim, S. Y., & Benowitz, N. L. (1990). Poisoning Due to Class IA Antiarrhythmic Drugs Quinidine, Procainamide and Disopyramide: *Drug Safety*, *5*(6), 393–420. <https://doi.org/10.2165/00002018-199005060-00002>
- Kingston, D. G. I. (2011). Modern Natural Products Drug Discovery and Its Relevance to Biodiversity Conservation. *Journal of Natural Products*, *74*(3), 496–511. <https://doi.org/10.1021/np100550t>
- Kirby, W. M. M., G. M. Yoshihara, K. S. Sundsted, and J. H. Warren. (1957). Clinical usefulness of a single disc method for antibiotic sensitivity testing. *Antibiotics Annu.*, *8*(2), 1956–1957.
- Knight, V., Sanglier, J.-J., DiTullio, D., Braccili, S., Bonner, P., Waters, J., Hughes, D., & Zhang, L. (2003). Diversifying microbial natural products for drug discovery. *Applied Microbiology and Biotechnology*, *62*(5–6), 446–458. <https://doi.org/10.1007/s00253-003-1381-9>
- Kobayashi, N., Katsumata, H., Katayama, H., Oiwa, H., Goto, J., & Takeuchi, Y. (2000). A monoclonal antibody-based enzyme-linked immunosorbent assay of ursodeoxycholic acid 3-sulfates in human urine. *The Journal of Steroid Biochemistry and Molecular Biology*, *72*(5), 265–272. [https://doi.org/10.1016/S0960-0760\(00\)00032-7](https://doi.org/10.1016/S0960-0760(00)00032-7)
- Koehn, F. E., & Carter, G. T. (2005). The evolving role of natural products in drug discovery. *Nature Reviews Drug Discovery*, *4*(3), 206–220. <https://doi.org/10.1038/nrd1657>
- Kohen, R., & Nyska, A. (2002). Invited Review: Oxidation of Biological Systems: Oxidative Stress Phenomena, Antioxidants, Redox Reactions, and Methods for Their

- Quantification. *Toxicologic Pathology*, 30(6), 620–650.
<https://doi.org/10.1080/01926230290166724>
- Kostelnik, A., Cegan, A., & Pohanka, M. (2017). Anti-Parkinson Drug Biperiden Inhibits Enzyme Acetylcholinesterase. *BioMed Research International*, 2017, 1–5.
<https://doi.org/10.1155/2017/2532764>
- Kuper, K. M., Boles, D. M., Mohr, J. F., & Wanger, A. (2009). Antimicrobial Susceptibility Testing: A Primer for Clinicians. *Pharmacotherapy: The Journal of Human Pharmacology and Drug Therapy*, 29(11), 1326–1343.
<https://doi.org/10.1592/phco.29.11.1326>
- Lambert, P. (2005). Bacterial resistance to antibiotics: Modified target sites. *Advanced Drug Delivery Reviews*, 57(10), 1471–1485. <https://doi.org/10.1016/j.addr.2005.04.003>
- Lee, S. T., Welch, K. D., Panter, K. E., Gardner, D. R., Garrossian, M., & Chang, C.-W. T. (2014). Cyclopamine: From Cyclops Lambs to Cancer Treatment. *Journal of Agricultural and Food Chemistry*, 62(30), 7355–7362.
<https://doi.org/10.1021/jf5005622>
- Leong, X. Y., Thanikachalam, P. V., Pandey, M., & Ramamurthy, S. (2016). A systematic review of the protective role of swertiamarin in cardiac and metabolic diseases. *Biomedicine & Pharmacotherapy*, 84, 1051–1060.
<https://doi.org/10.1016/j.biopha.2016.10.044>
- Li, J. W.-H., & Vederas, J. C. (2009). Drug Discovery and Natural Products: End of an Era or an Endless Frontier? *Science*, 325(5937), 161–165.
<https://doi.org/10.1126/science.1168243>
- Li, S., Tan, H.-Y., Wang, N., Zhang, Z.-J., Lao, L., Wong, C.-W., & Feng, Y. (2015). The Role of Oxidative Stress and Antioxidants in Liver Diseases. *International Journal of Molecular Sciences*, 16(11), 26087–26124. <https://doi.org/10.3390/ijms161125942>
- Ligon, B. L. (2004). Penicillin: Its discovery and early development. *Seminars in Pediatric Infectious Diseases*, 15(1), 52–57. <https://doi.org/10.1053/j.spid.2004.02.001>

- Lin, Y., Liu, P.-G., Liang, W.-Q., Hu, Y.-J., Xu, P., Zhou, J., Pu, J.-B., & Zhang, H.-J. (2018). Luteolin-4-O-glucoside and its aglycone, two major flavones of *Gnaphalium affine* D. Don, resist hyperuricemia and acute gouty arthritis activity in animal models. *Phytomedicine*, *41*, 54–61. <https://doi.org/10.1016/j.phymed.2018.02.002>
- MARYAM ZAHIN1, , FARRUKH AQIL2, & AND IQBAL AHMAD*. (2009). THE IN VITRO ANTIOXIDANT ACTIVITY AND TOTAL PHENOLIC CONTENT OF FOUR INDIAN MEDICINAL PLANTS. *International Journal of Pharmacy and Pharmaceutical Sciences, Vol. 1*.
- Massera, D., Sherrid, M. V., Adlestein, E., Bokhari, N., Alvarez, I. C., Wu, W. Y., Reuter, M. C., Maron, M. S., Maron, B. J., & Rowin, E. J. (2025). Disopyramide Revisited for Treatment of Symptomatic Obstructive Hypertrophic Cardiomyopathy: Efficacy and Safety in Patients Treated for at Least 5 Years. *Journal of the American Heart Association*, *14*(2), e037639. <https://doi.org/10.1161/JAHA.124.037639>
- Masuko, T., Minami, A., Iwasaki, N., Majima, T., Nishimura, S.-I., & Lee, Y. C. (2005). Carbohydrate analysis by a phenol–sulfuric acid method in microplate format. *Analytical Biochemistry*, *339*(1), 69–72. <https://doi.org/10.1016/j.ab.2004.12.001>
- Mathur, P. P. (1972). Cardiovascular effects of a newer antiarrhythmic agent, disopyramide phosphate. *American Heart Journal*, *84*(6), 764–770. [https://doi.org/10.1016/0002-8703\(72\)90069-5](https://doi.org/10.1016/0002-8703(72)90069-5)
- Meyer, T. A., Roberson, C. R., Rajab, M. H., Davis, J., & McLeskey, C. H. (2005). Dolasetron Versus Ondansetron for the Treatment of Postoperative Nausea and Vomiting: *Anesthesia & Analgesia*, *100*(2), 373–377. <https://doi.org/10.1213/01.ANE.0000144421.96275.D1>
- Mishra, B. B., & Tiwari, V. K. (2011). Natural products: An evolving role in future drug discovery. *European Journal of Medicinal Chemistry*, *46*(10), 4769–4807. <https://doi.org/10.1016/j.ejmech.2011.07.057>

- Morgan, C. J. A., Curran, H. V., & the Independent Scientific Committee on Drugs (ISCD). (2012). Ketamine use: A review. *Addiction*, *107*(1), 27–38. <https://doi.org/10.1111/j.1360-0443.2011.03576.x>
- Muhamad Fadzil, N. S., Sekar, M., Gan, S. H., Bonam, S. R., Wu, Y. S., Vaijanathappa, J., Ravi, S., Lum, P. T., & Dhadde, S. B. (2021). Chemistry, Pharmacology and Therapeutic Potential of Swertiamarin – A Promising Natural Lead for New Drug Discovery and Development. *Drug Design, Development and Therapy, Volume 15*, 2721–2746. <https://doi.org/10.2147/DDDT.S299753>
- Musial, C., Kuban-Jankowska, A., & Gorska-Ponikowska, M. (2020). Beneficial Properties of Green Tea Catechins. *International Journal of Molecular Sciences*, *21*(5), 1744. <https://doi.org/10.3390/ijms21051744>
- Mweta, D. E., Labuschagne, M. T., Bonnet, S., Swarts, J., & Saka, J. D. K. (2010). Isolation and physicochemical characterisation of starch from cocoyam (*Colocasia esculenta*) grown in Malawi. *Journal of the Science of Food and Agriculture*, n/a-n/a. <https://doi.org/10.1002/jsfa.4029>
- Nagata, H., Inagaki, Y., Yamamoto, Y., Maeda, K., Kataoka, K., Osawa, K., & Shizukuishi, S. (2006). Inhibitory effects of macrocarpals on the biological activity of *Porphyromonas gingivalis* and other periodontopathic bacteria. *Oral Microbiology and Immunology*, *21*(3), 159–163. <https://doi.org/10.1111/j.1399-302X.2006.00269.x>
- Newman, D. J. (2008). Natural Products as Leads to Potential Drugs: An Old Process or the New Hope for Drug Discovery? *Journal of Medicinal Chemistry*, *51*(9), 2589–2599. <https://doi.org/10.1021/jm0704090>
- Niki, E. (2016). Oxidative stress and antioxidants: Distress or eustress? *Archives of Biochemistry and Biophysics*, *595*, 19–24. <https://doi.org/10.1016/j.abb.2015.11.017>
- Ogut, E., Armagan, K., & Gül, Z. (2022). The role of syringic acid as a neuroprotective agent for neurodegenerative disorders and future expectations. *Metabolic Brain Disease*, *37*(4), 859–880. <https://doi.org/10.1007/s11011-022-00960-3>

- Oku, T., Murata-Takenoshita, Y., Yamazaki, Y., Shimura, F., & Nakamura, S. (2014). D-Sorbose inhibits disaccharidase activity and demonstrates suppressive action on postprandial blood levels of glucose and insulin in the rat. *Nutrition Research*, *34*(11), 961–967. <https://doi.org/10.1016/j.nutres.2014.09.009>
- Olatunde, K., Adebayo, K., Muhumuza, J., & Bada, B. (2018). Assessment of variability in proximate/anti-nutritive composition of cocoyam within Nigeria and Uganda. *Journal of Applied Sciences and Environmental Management*, *22*(5), 737. <https://doi.org/10.4314/jasem.v22i5.20>
- Olayiwola, I., Folaranmi, F., Adebowale, A. A., Oluseye, O., Ajoke, S., & Wasiu, A. (2013). Chemical, mineral composition, and sensory acceptability of cocoyam based recipes enriched with cowpea flour. *Food Science & Nutrition*, *1*(3), 228–234. <https://doi.org/10.1002/fsn3.30>
- Olufunmilayo, E. O., Gerke-Duncan, M. B., & Holsinger, R. M. D. (2023). Oxidative Stress and Antioxidants in Neurodegenerative Disorders. *Antioxidants*, *12*(2), 517. <https://doi.org/10.3390/antiox12020517>
- Otekunrin, O. A., Sawicka, B., Adeyonu, A. G., Otekunrin, O. A., & Racho, L. (2021). Cocoyam [*Colocasia esculenta* (L.) Schott]: Exploring the Production, Health and Trade Potentials in Sub-Saharan Africa. *Sustainability*, *13*(8), 4483. <https://doi.org/10.3390/su13084483>
- Owusu-Darko, P. G., Paterson, A., & Omenyo, E. L. (2014). Cocoyam (corms and cormels)—An underexploited food and feed resource. *Journal of Agricultural Chemistry and Environment*, *03*(01), 22–29. <https://doi.org/10.4236/jacen.2014.31004>
- Passemar, C., Saléry, M., Soh, P. N., Linas, M.-D., Ahond, A., Poupat, C., & Benoit-Vical, F. (2011). Indole and aminoimidazole moieties appear as key structural units in antiplasmodial molecules. *Phytomedicine*, *18*(13), 1118–1125. <https://doi.org/10.1016/j.phymed.2011.03.010>

- Patricia G. Owusu-Darko, Alistair Paterson, Emmanuel L. Omenyo. (2014). Cocoyam (corms and cormels) an underexploited food and feed resource. *Journal of Agricultural Chemistry and Environment*, 03. <http://dx.doi.org/10.4236/jacen.2014.31004>
- Paudel, P. K., Bhattarai, B. P., & Kindlmann, P. (2012). An Overview of the Biodiversity in Nepal. In P. Kindlmann (Ed.), *Himalayan Biodiversity in the Changing World* (pp. 1–40). Springer Netherlands. https://doi.org/10.1007/978-94-007-1802-9_1
- Paurakh Magar, Galaxy Pokhrel, Sachin Silwal, , Sabi Thapa, , Ajaya Mahato, , Bishisha Adhikara, , Shiva Bagale, , Pabitra shrestha, , & Bodhbabu Bhattaraia*, Ganga Raj Pokhrel a*. (2024). Antioxidant Activity and HR-LCMS Analysis of Phytochemicals Present in the Methanolic Extract of the Rhizomes of Paris Polyphylla (Satuwa). *Api Journal of Science*, 1(1), 8–17.
- Perrone, S., Negro, S., Tataranno, M. L., & Buonocore, G. (2010). Oxidative stress and antioxidant strategies in newborns. *The Journal of Maternal-Fetal & Neonatal Medicine*, 23(sup3), 63–65. <https://doi.org/10.3109/14767058.2010.509940>
- Pisoschi, A. M., & Pop, A. (2015). The role of antioxidants in the chemistry of oxidative stress: A review. *European Journal of Medicinal Chemistry*, 97, 55–74. <https://doi.org/10.1016/j.ejmech.2015.04.040>
- Pohl, F., & Kong Thoo Lin, P. (2018). The Potential Use of Plant Natural Products and Plant Extracts with Antioxidant Properties for the Prevention/Treatment of Neurodegenerative Diseases: In Vitro, In Vivo and Clinical Trials. *Molecules*, 23(12), 3283. <https://doi.org/10.3390/molecules23123283>
- Pokhrel, G., Silwal, S., Thapa, S., Giri, S., Mahato, A., Adhikari, B., Bagale, S., Bhattarai, B., & Pokhrel, G. R. (2024). Assessment of Micronutrients and Toxic Elements in Rice Grains: Implications for Food Safety and Nutrition. *BMC Journal of Scientific Research*, 7(1), 37–49. <https://doi.org/10.3126/bmcjsr.v7i1.72942>
- Poole, K., Russell, A., & Lambert, P. (2005). Mechanisms of antimicrobial resistance: Opportunities for new targeted therapies. *Advanced Drug Delivery Reviews*, 57(10), 1443–1445. <https://doi.org/10.1016/j.addr.2005.05.001>

- Prieto, P., Pineda, M., & Aguilar, M. (1999). Spectrophotometric Quantitation of Antioxidant Capacity through the Formation of a Phosphomolybdenum Complex: Specific Application to the Determination of Vitamin E. *Analytical Biochemistry*, 269(2), 337–341. <https://doi.org/10.1006/abio.1999.4019>
- Rabiu, I., Yusha'u, M., Isah Abdullahi, J., & Muhammed, A. (2024). Phytochemical analysis, Thin Layer Chromatography and Gas Chromatography Mass Spectroscopy profile of colocasia esculenta and manihot esculanta extracts and their potential for drug discovery. *Drug Discovery*, 18(41), 1–20. <https://doi.org/10.54905/disssi.v18i41.e10dd1975>
- Ramawat, K. G. (Ed.). (2009). *Herbal Drugs: Ethnomedicine to Modern Medicine*. Springer Berlin Heidelberg. <https://doi.org/10.1007/978-3-540-79116-4>
- Rao, A. V., & Balachandran, B. (2002). Role of Oxidative Stress and Antioxidants in Neurodegenerative Diseases. *Nutritional Neuroscience*, 5(5), 291–309. <https://doi.org/10.1080/1028415021000033767>
- Reyes, S., Abdelraouf, K., & Nicolau, D. P. (2020). In vivo activity of human-simulated regimens of imipenem alone and in combination with relebactam against *Pseudomonas aeruginosa* in the murine thigh infection model. *Journal of Antimicrobial Chemotherapy*, dkaa145. <https://doi.org/10.1093/jac/dkaa145>
- Rizos, I., Brachmann, J., Lengfelder, W., Schmitt, C., Olshausen, K. V., Kübler, W., & Senges, J. (1987). Effects of intravenous disopyramide and quinidine on normal myocardium and on the characteristics of arrhythmias: Intraindividual comparison in patients with sustained ventricular tachycardia. *European Heart Journal*, 8(2), 154–163. <https://doi.org/10.1093/oxfordjournals.eurheartj.a062243>
- Rodrigues, T., Reker, D., Schneider, P., & Schneider, G. (2016). Counting on natural products for drug design. *Nature Chemistry*, 8(6), 531–541. <https://doi.org/10.1038/nchem.2479>

- Sagar, N. A., Pareek, S., & Gonzalez-Aguilar, G. A. (2020). Quantification of flavonoids, total phenols and antioxidant properties of onion skin: A comparative study of fifteen Indian cultivars. *Journal of Food Science and Technology*, *57*(7), 2423–2432. <https://doi.org/10.1007/s13197-020-04277-w>
- Sahari Shimsa, Sukanta Mondal, Saraswathy Mini. (2024). Syringic acid: A promising phenolic phytochemical with extensive therapeutic applications. *R&D of Functional Food Products*, *1*(5), 1–14.
- Saugstad, O. D. (2003). Bronchopulmonary dysplasia—Oxidative stress and antioxidants. *Seminars in Neonatology*, *8*(1), 39–49. [https://doi.org/10.1016/S1084-2756\(02\)00194-X](https://doi.org/10.1016/S1084-2756(02)00194-X)
- Sefa-Dedeh, S., & Agyir-Sackey, E. K. (2004). Chemical composition and the effect of processing on oxalate content of cocoyam *Xanthosoma sagittifolium* and Colocasia esculenta cormels. *Food Chemistry*, *85*(4), 479–487. [https://doi.org/10.1016/S0308-8146\(02\)00244-3](https://doi.org/10.1016/S0308-8146(02)00244-3)
- Senyilmaz-Tiebe, D., Pfaff, D. H., Virtue, S., Schwarz, K. V., Fleming, T., Altamura, S., Muckenthaler, M. U., Okun, J. G., Vidal-Puig, A., Nawroth, P., & Teleanu, A. A. (2018). Dietary stearic acid regulates mitochondria in vivo in humans. *Nature Communications*, *9*(1), 3129. <https://doi.org/10.1038/s41467-018-05614-6>
- Serna-Loaiza, S., Carmona-Garcia, E., & Cardona, C. A. (2018). Potential raw materials for biorefineries to ensure food security: The Cocoyam case. *Industrial Crops and Products*, *126*, 92–102. <https://doi.org/10.1016/j.indcrop.2018.10.005>
- Shady, N. H., Mokhtar, F. A., Abdullah, H. S., Abdel-Aziz, S. A., Mohamad, S. A., Imam, M. S., El Afify, S. R., & Abdelmohsen, U. R. (2024). In Vitro and Randomized Controlled Clinical Study of Natural Constituents' Anti-HPV Potential for Treatment of Plantar Warts Supported with In Silico Studies and Network Analysis. *Pharmaceuticals*, *17*(6), 759. <https://doi.org/10.3390/ph17060759>
- Shaito, A., Thuan, D. T. B., Phu, H. T., Nguyen, T. H. D., Hasan, H., Halabi, S., Abdelhady, S., Nasrallah, G. K., Eid, A. H., & Pintus, G. (2020). Herbal Medicine for

- Cardiovascular Diseases: Efficacy, Mechanisms, and Safety. *Frontiers in Pharmacology*, *11*, 422. <https://doi.org/10.3389/fphar.2020.00422>
- Shakur, H., Andrews, P., Asser, T., Balica, L., Boeriu, C., Quintero, J. D. C., Dewan, Y., Druwé, P., Fletcher, O., Frost, C., Hartzenberg, B., Mantilla, J. M., Murillo-Cabezas, F., Pahl, J., Ravi, R. R., Rätsep, I., Sampaio, C., Singh, M., Svoboda, P., & Roberts, I. (2009). The BRAIN TRIAL: A randomised, placebo controlled trial of a Bradykinin B2 receptor antagonist (Anatibant) in patients with traumatic brain injury. *Trials*, *10*(1), 109. <https://doi.org/10.1186/1745-6215-10-109>
- Sharifi-Rad, M., Anil Kumar, N. V., Zucca, P., Varoni, E. M., Dini, L., Panzarini, E., Rajkovic, J., Tsouh Fokou, P. V., Azzini, E., Peluso, I., Prakash Mishra, A., Nigam, M., El Rayess, Y., Beyrouthy, M. E., Polito, L., Iriti, M., Martins, N., Martorell, M., Docea, A. O., ... Sharifi-Rad, J. (2020). Lifestyle, Oxidative Stress, and Antioxidants: Back and Forth in the Pathophysiology of Chronic Diseases. *Frontiers in Physiology*, *11*, 694. <https://doi.org/10.3389/fphys.2020.00694>
- Shree, P., Mishra, P., Selvaraj, C., Singh, S. K., Chaube, R., Garg, N., & Tripathi, Y. B. (2022). Targeting COVID-19 (SARS-CoV-2) main protease through active phytochemicals of ayurvedic medicinal plants – *Withania somnifera* (Ashwagandha), *Tinospora cordifolia* (Giloy) and *Ocimum sanctum* (Tulsi) – a molecular docking study. *Journal of Biomolecular Structure and Dynamics*, *40*(1), 190–203. <https://doi.org/10.1080/07391102.2020.1810778>
- Shrestha, N., Shrestha, S., Koju, L., Shrestha, K. K., & Wang, Z. (2016). Medicinal plant diversity and traditional healing practices in eastern Nepal. *Journal of Ethnopharmacology*, *192*, 292–301. <https://doi.org/10.1016/j.jep.2016.07.067>
- Silva, T., Reis, J., Teixeira, J., & Borges, F. (2014). Alzheimer's disease, enzyme targets and drug discovery struggles: From natural products to drug prototypes. *Ageing Research Reviews*, *15*, 116–145. <https://doi.org/10.1016/j.arr.2014.03.008>
- Singleton, V. L., Orthofer, R., & Lamuela-Raventós, R. M. (1999). [14] Analysis of total phenols and other oxidation substrates and antioxidants by means of folin-ciocalteu

- reagent. In *Methods in Enzymology* (Vol. 299, pp. 152–178). Elsevier.
[https://doi.org/10.1016/S0076-6879\(99\)99017-1](https://doi.org/10.1016/S0076-6879(99)99017-1)
- Sleda, M. A., Li, Z.-H., Behera, R., Baierna, B., Li, C., Jumpathong, J., Malwal, S. R., Kawamukai, M., Oldfield, E., & Moreno, S. N. J. (2022). The Heptaprenyl Diphosphate Synthase (Coq1) Is the Target of a Lipophilic Bisphosphonate That Protects Mice against *Toxoplasma gondii* Infection. *mBio*, *13*(5), e01966-22.
<https://doi.org/10.1128/mbio.01966-22>
- Stephen T Odonkor a*, Kennedy K Addo. (2011). Bacteria Resistance to Antibiotics: Recent Trends and Challenges. *International Journal of Biological & Medical Research*, (4): 1204-1210(2), 1204–1210.
- Stephen T Odonkor ,Kennedy K Addo. (2011). Bacteria Resistance to Antibiotics: Recent Trends and Challenges. *International Journal of Biological & Medical Research*, 2(4), 1204–1210.
- Stuart Noble and Karen L. Goa. (n.d.). Amprenavir A Review of its Clinical Potential in Patients with HIV Infection. *ADIS DRUG EVALUATION, Drugs 2000 Dec; 60 (6): 1383-1410.*
- Stuart R. Hameroff, MD, Juan Carlos Lerman, PhD and Casey D. Blitt, MD. (1984). Etidocaine Aerosol for Laryngotracheal Anesthesia Clinical Evaluation Using Exploratory Data Analysis. *BMJ Journals*, 9(4). <https://doi.org/10.1136/rapm-00115550-198409040-00006>
- Szuba, T. (2002). Was There Collective Intelligence Before Life on Earth? Considerations on the Formal Foundations of Intelligence, Life, and Evolution. *World Futures*, 58(1), 61–80. <https://doi.org/10.1080/02604020210404>
- Tajner-Czopek, A., Gertchen, M., Rytel, E., Kita, A., Kucharska, A. Z., & Sokół-Ł towska, A. (2020). Study of Antioxidant Activity of Some Medicinal Plants Having High Content of Caffeic Acid Derivatives. *Antioxidants*, 9(5), 412.
<https://doi.org/10.3390/antiox9050412>

- Takayanagi, I., Hisayama, T., Iwase, M., Sakuma, N., & Nagai, H. (1989). Pharmacological properties of tiropramide, an antispasmodic drug. *General Pharmacology: The Vascular System*, 20(3), 335–339. [https://doi.org/10.1016/0306-3623\(89\)90269-3](https://doi.org/10.1016/0306-3623(89)90269-3)
- Tanus, T., Scangarella-Oman, N. E., Dalessandro, M., Li, G., Breton, J. J., & Tomayko, J. F. (2014). A Randomized, Double-blind, Comparative Study to Assess the Safety and Efficacy of Topical Retapamulin Ointment 1% Versus Oral Linezolid in the Treatment of Secondarily Infected Traumatic Lesions and Impetigo Due to Methicillin-Resistant *Staphylococcus aureus*. *Advances in Skin & Wound Care*, 27(12), 548–559. <https://doi.org/10.1097/01.ASW.0000456631.20389.ae>
- Tay, D. (2013). Tropical and Subtropical Root and Tuber Crops. In M. N. Normah, H. F. Chin, & B. M. Reed (Eds.), *Conservation of Tropical Plant Species* (pp. 249–292). Springer New York. https://doi.org/10.1007/978-1-4614-3776-5_12
- Thomford, N. E., Senthebane, D. A., Rowe, A., Munro, D., Seele, P., Maroyi, A., & Dzobo, K. (2018). Natural Products for Drug Discovery in the 21st Century: Innovations for Novel Drug Discovery. *International Journal of Molecular Sciences*, 19(6), 1578. <https://doi.org/10.3390/ijms19061578>
- Thomson, A., Hemphill, D., & Jeejeebhoy, K. N. (1998). Oxidative Stress and Antioxidants in Intestinal Disease. *Digestive Diseases*, 16(3), 152–158. <https://doi.org/10.1159/000016859>
- Tohru Dairi, Kazuo Yamaguchi, and Mamoru Hasegawa. (1992). N-Formimidoyl fortimicin A synthase, a unique oxidase involved in fortimicin A biosynthesis: Purification, characterization and gene cloning. *Mol Gen Genet* (1992) 236:49-59.
- Tufikul Islam, Mohammad Raquibul Hasan, Aumit Roy, Md. Shafiqul Islam, Md. Afaz Uddin, Md. Ariful Islam, Md. Nuruzzaman Neon, Md. Sohel Rana. (2015). SCREENING OF IN-VITRO ANTIOXIDANT, BRINE SHRIMP LETHALITY BIOASSAY AND ANTIMICROBIAL ACTIVITIES OF EXTRACTS OF BRIDELIA RETUSA (L.) SPRENG. FRUIT. *International Journal of Pharmacy*, 5(4), 1058–1067.

- Vetter, I., Davis, J. L., Rash, L. D., Anangi, R., Mobli, M., Alewood, P. F., Lewis, R. J., & King, G. F. (2011). Venomics: A new paradigm for natural products-based drug discovery. *Amino Acids*, 40(1), 15–28. <https://doi.org/10.1007/s00726-010-0516-4>
- Wada, E., Feyissa, T., & Tesfaye, K. (2019). Proximate, Mineral and Antinutrient Contents of Cocoyam (*Xanthosoma sagittifolium* (L.) Schott) from Ethiopia. *International Journal of Food Science*, 2019, 1–7. <https://doi.org/10.1155/2019/8965476>
- Wakeling, A. (1983). Efficacy and side effects of mianserin, a tetracyclic antidepressant. *Postgraduate Medical Journal*, 59(690), 229–231. <https://doi.org/10.1136/pgmj.59.690.229>
- Wang, J.-K., & Higa, S. (Eds.). (1983). *Taro, a review of colocasia esculenta and its potentials*. University of Hawaii Press.
- Wang, L., Pan, X., Jiang, L., Chu, Y., Gao, S., Jiang, X., Zhang, Y., Chen, Y., Luo, S., & Peng, C. (2022). The Biological Activity Mechanism of Chlorogenic Acid and Its Applications in Food Industry: A Review. *Frontiers in Nutrition*, 9, 943911. <https://doi.org/10.3389/fnut.2022.943911>
- Wang, S., Gong, T., Lu, J., Kano, Y., & Yuan, D. (2013). Simultaneous determination of tectorigenin and its metabolites in rat plasma by ultra performance liquid chromatography/quadrupole time-of-flight mass spectrometry. *Journal of Chromatography B*, 933, 50–58. <https://doi.org/10.1016/j.jchromb.2013.06.009>
- Wei Lei , Zhenyun Huo, W. L. a, Zhenyun Huo. (2020). Jervine inhibits non-small cell lung cancer (NSCLC) progression by suppressing Hedgehog and AKT signaling via triggering autophagyregulated apoptosis. *Biochemical and Biophysical Research Communications*.
- Wilson, S. R., Strand, M. F., Krapp, A., Rise, F., Petersen, D., & Krauss, S. (2010). Hedgehog antagonist cyclopamine isomerizes to less potent forms when acidified. *Journal of Pharmaceutical and Biomedical Analysis*, 52(5), 707–713. <https://doi.org/10.1016/j.jpba.2010.02.017>

- Wirz-Justice, A., & Lichtsteiner, M. (1976). Sex specific differences in noradrenaline uptake and its inhibition by maprotiline. *The Journal of Pharmacy and Pharmacology*, 28(2), 172–175. <https://doi.org/10.1111/j.2042-7158.1976.tb04124.x>
- Wright, C., Bray, D., O'Neill, M., Warhurst, D., Phillipson, J., Quetin-Leclercq, J., & Angenot, L. (1991). Antiamœbic and Antiplasmodial Activities of Alkaloids Isolated from *Strychnos usambarensis*. *Planta Medica*, 57(04), 337–340. <https://doi.org/10.1055/s-2006-960112>
- Wu, J.-Y., Ding, H.-Y., Wang, T.-Y., Cai, C.-Z., & Chang, T.-S. (2022). Application of Biotransformation-Guided Purification in Chinese Medicine: An Example to Produce Butin from Licorice. *Catalysts*, 12(7), 718. <https://doi.org/10.3390/catal12070718>
- W.-W. MA, J.E. ANDERSON, A.T. MCKENZIE, S.R. BYRN, J.L. MCLAUGHLIN. (1990). TUBULOSINE: AN ANTITUMOR CONSTITUENT OF POGONOPUS SPECIOSUS. *Journal of Natural ProdwtS*, 53(4), 1009–1014.
- Xu, M., Zhang, L., Zeng, Y., Zhou, Z., & Han, Y. (2025). Preparation and characterization of Levan composite film incorporating vanillin for use as a potential edible coating for peony seed oil. *International Journal of Biological Macromolecules*, 288, 138732. <https://doi.org/10.1016/j.ijbiomac.2024.138732>
- Yakan, S., Aydin, T., Gulmez, C., Ozden, O., Eren Erdogan, K., Daglioglu, Y. K., Andic, F., Atakisi, O., & Cakir, A. (2019). The protective role of jervine against radiation-induced gastrointestinal toxicity. *Journal of Enzyme Inhibition and Medicinal Chemistry*, 34(1), 789–798. <https://doi.org/10.1080/14756366.2019.1586681>
- Yang, G., Xie, H., Wang, C., Zhang, C., Yu, L., Zhang, L., Liu, X., Xu, R., Song, Z., Liu, R., & Ueda, M. (2023). Design, synthesis, and discovery of Eudistomin Y derivatives as lysosome-targeted antiproliferation agents. *European Journal of Medicinal Chemistry*, 250, 115193. <https://doi.org/10.1016/j.ejmech.2023.115193>

- Yao, X., Ling, Y., Guo, S., He, S., Wang, J., Zhang, Q., Wu, W., Zou, M., Zhang, T., Nandakumar, K. S., Chen, X., & Liu, S. (2018). Inhibition of dengue viral infection by diasarone-I is associated with 2'O methyltransferase of NS5. *European Journal of Pharmacology*, *821*, 11–20. <https://doi.org/10.1016/j.ejphar.2017.12.029>
- Yoshikawa, M., Murakami, T., Shimada, H., Yoshizumi, S., Saka, M., Yamahara, J., & Matsuda, H. (1998). Medicinal Foodstuffs. XIV. On the Bioactive Constituents of Moroheiya. (2): New Fatty Acids, Corchorifatty Acids A, B, C, D, E, and F, from the Leaves of *Corchorus olitorius* L. (Tiliaceae): Structures and Inhibitory Effect on NO Production in Mouse Peritoneal Macrophages. *Chemical and Pharmaceutical Bulletin*, *46*(6), 1008–1014. <https://doi.org/10.1248/cpb.46.1008>
- Zhang, L., Song, J., Kong, L., Yuan, T., Li, W., Zhang, W., Hou, B., Lu, Y., & Du, G. (2020). The strategies and techniques of drug discovery from natural products. *Pharmacology & Therapeutics*, *216*, 107686. <https://doi.org/10.1016/j.pharmthera.2020.107686>
- Zhang, R., Chae, S., Kang, K. A., Piao, M. J., Ko, D. O., Wang, Z. H., Park, D. B., Park, J. W., You, H. J., & Hyun, J. W. (2008). Protective effect of butin against hydrogen peroxide-induced apoptosis by scavenging reactive oxygen species and activating antioxidant enzymes. *Molecular and Cellular Biochemistry*, *318*(1–2), 33–42. <https://doi.org/10.1007/s11010-008-9854-x>
- Zhao, M., Zhu, Y., Wang, H., Zhang, W., & Mu, W. (2023). Recent advances on N-acetylneuraminic acid: Physiological roles, applications, and biosynthesis. *Synthetic and Systems Biotechnology*, *8*(3), 509–519. <https://doi.org/10.1016/j.synbio.2023.06.009>

APPENDIX I

HR-LCMS METHODS OF ANALYSIS

Acquisition Method Report



Name: Binary Pump

Model: G4220B

Flow 0.300 mL/min
 Use Solvent Types Yes
 Stroke Mode Synchronized
 Low Pressure Limit 0.00 bar
 High Pressure Limit 1200.00 bar
 Max. Flow Ramp Up 100.000 mL/min²
 Max. Flow Ramp Down 100.000 mL/min²
 Expected Mixer No check
 Stroke A
 Automatic Stroke Calculation A Yes
 Stop Time
 Stoptime Mode Time set
 Stoptime 35.00 min
 Post Time
 Posttime Mode Off

Solvent Composition

	Channel	Ch. 1 Solv.	Name 1	Ch2 Solv.	Name 2	Selected	Used	Percent
1	A	100.0 % Water V.02	0.1% FA in water	100.0 % Water V.02	0.1% FA in water	Ch. 2	Yes	95.00 %
2	B	100.0 % Acetonitrile V.03		100.0 % Acetonitrile V.02		Ch. 2	Yes	5.00 %

Timetable

	Time	A	B	Flow	Pressure
1	1.00 min	95.00 %	5.00 %	0.300 mL/min	1200.00 bar
2	25.00 min	0.00 %	100.00 %	0.300 mL/min	1200.00 bar
3	30.00 min	0.00 %	100.00 %	0.300 mL/min	1200.00 bar
4	31.00 min	95.00 %	5.00 %	0.300 mL/min	1200.00 bar
5	35.00 min	95.00 %	5.00 %	0.300 mL/min	1200.00 bar

Name: Column Comp.

Model: G1316C

Ready when front door open Yes
 Left Temperature Control
 Temperature Control Mode Temperature Set
 Temperature 40.00 °C
 Enable Analysis Left Temperature
 Enable Analysis Left Temperature On Yes
 Enable Analysis Left Temperature Value 0.80 °C
 Right Temperature Control
 Right temperature Control Mode Temperature Set
 Right temperature 40.00 °C
 Enable Analysis Right Temperature
 Enable Analysis Right Temperature On Yes
 Enable Analysis Right Temperature Value 0.80 °C
 Stop Time
 Stoptime Mode As pump/injector
 Post Time
 Posttime Mode Off

Name: Binary Pump

Model: G4220B

Flow 0.300 mL/min
 Use Solvent Types Yes
 Stroke Mode Synchronized
 Low Pressure Limit 0.00 bar
 High Pressure Limit 1200.00 bar
 Max. Flow Ramp Up 100.000 mL/min²
 Max. Flow Ramp Down 100.000 mL/min²
 Expected Mixer No check
 Stroke A
 Automatic Stroke Calculation A Yes
 Stop Time
 Stoptime Mode Time set
 Stoptime 35.00 min
 Post Time
 Posttime Mode Off

Solvent Composition

	Channel	Ch. 1 Solv.	Name 1	Ch2 Solv.	Name 2	Selected	Used	Percent
1	A	100.0 % Water V.02	0.1% FA in water	100.0 % Water V.02	0.1% FA in water	Ch. 2	Yes	95.00 %
2	B	100.0 % Acetonitrile V.03		100.0 % Acetonitrile V.02		Ch. 2	Yes	5.00 %

Timetable

	Time	A	B	Flow	Pressure
1	1.00 min	95.00 %	5.00 %	0.300 mL/min	1200.00 bar
2	25.00 min	0.00 %	100.00 %	0.300 mL/min	1200.00 bar
3	30.00 min	0.00 %	100.00 %	0.300 mL/min	1200.00 bar
4	31.00 min	95.00 %	5.00 %	0.300 mL/min	1200.00 bar
5	35.00 min	95.00 %	5.00 %	0.300 mL/min	1200.00 bar

Name: Column Comp.

Model: G1316C

Ready when front door open Yes
 Left Temperature Control
 Temperature Control Mode Temperature Set
 Temperature 40.00 °C
 Enable Analysis Left Temperature
 Enable Analysis Left Temperature On Yes
 Enable Analysis Left Temperature Value 0.80 °C
 Right Temperature Control
 Right temperature Control Mode Temperature Set
 Right temperature 40.00 °C
 Enable Analysis Right Temperature
 Enable Analysis Right Temperature On Yes
 Enable Analysis Right Temperature Value 0.80 °C
 Stop Time
 Stoptime Mode As pump/injector
 Post Time
 Posttime Mode Off

Acquisition Method Report



Name: HiP Sampler

Model: G4226A

Auxiliary

Draw Speed	100.0 $\mu\text{L}/\text{min}$
Eject Speed	100.0 $\mu\text{L}/\text{min}$
Draw Position Offset	0.0 mm
Wait Time After Drawing	2.0 s
Sample Flush Out Factor	5.0
Vial/Well bottom sensing	Yes

Injection

Injection Mode	Injection with needle wash
Injection Volume	5.00 μL
Needle Wash	
Needle Wash Location	Flush Port
Wash Time	3.0 s

Highthroughput

Automatic Delay Volume Reduction	No
Overlapped Injection	
Enable Overlapped injection	No

Valve Switching

Valve Movements	0
Valve Switch Time 1	
Switch Time 1 Enabled	Yes
Switch Time 1	0.01 min
Valve Switch Time 2	
Switch Time 2 Enabled	No
Valve Switch Time 3	
Switch Time 3 Enabled	No
Valve Switch Time 4	
Switch Time 4 Enabled	No

Stop Time

Stoptime Mode	As pump/No limit
---------------	------------------

Post Time

Posttime Mode	Off
---------------	-----

Time Segment 1

Acquisition Method Summary

MS Scan Range (m/z)	100
MS Scan Range (min)	1.000
MS Scan Rate (scans/min)	1.000
MS/MS Scan Rate (scans/min)	1.000
Isolation Width (m/z)	Medium (10.000)

Acquisition Parameters

Charge	1	Offset	+10
Stage	1	Y Range	1000000
Mass	1000	Ion Mode	Dual AJS ESI
Storage	1	Ion Mode	Dual AJS ESI

Pressure Detectors

MS Scan Range (m/z)	100
MS Scan Range (min)	1.000
MS Scan Rate (scans/min)	1.000
MS/MS Scan Rate (scans/min)	1.000
Isolation Width (m/z)	Medium (10.000)
Charge	1
Stage	1
Mass	1000
Storage	1

Source Parameters

Parameter	Value
Gas Flow (l/min)	10.0
Gas Flow (psi)	10.0
Gas Flow (mL/min)	10.0
Gas Flow (mL/min)	10.0
Gas Flow (mL/min)	10.0
Gas Flow (mL/min)	10.0

Reference Masses

MS Scan Range (m/z)	100
MS Scan Range (min)	1.000
MS Scan Rate (scans/min)	1.000
MS/MS Scan Rate (scans/min)	1.000
Isolation Width (m/z)	Medium (10.000)

Chromatogram

Chrom Type	TIC
Chrom Type	TIC
Chrom Type	TIC

Acquisition Method Info

Method Name: metabolite_ESI_+VE_MSMS.m
 Method Path: D:\MassHunter\Methods\2024\metabolite_ESI_+VE_MSMS.m
 Method Description: Default Method

Device List

- HIP Sampler
- Binary Pump
- Column Comp.
- DAD
- Q-TOF

TOF/Q-TOF Mass Spectrometer

Component Name	MS Q-TOF	Component Model	G6550A
Ion Source	Dual AJS ESI	Stop Time (min)	30.00
Can wait for temp.	Enable	Fast Polarity	N/A
MS Abs. threshold	200	MS Rel. threshold(%)	0.010
MS/MS Abs. threshold	5	MS/MS Rel. threshold(%)	0.010
Tune File	Azotune.tun		

Time Segments

Time Segment #	Start Time (min)	Diverter Valve State	Storage Mode	Ion Mode
1	0	MS	Both	Dual AJS ESI

APPENDIX II

HR-LCMS profiling of compounds in UPT in positive ESI mode.

Acquisition Method Report



Name: DAD **Model:** G4212B
Peakwidth: $\pm 0.10 \text{ min}$ (2.0 : response time) (2.5 Hz)
UV Lamp Required: Yes
Analog Output 1:
Analog 1 Zero Offset: 5 %
Analog 1 Attenuation: 1000 mAU
Signals:
Prepare Mode:
Margin for negative Absorbance: 100 mAU
Autobalance:
Autobalance Pre-run: Yes
Autobalance Post-run: No
Spectrum:
Spectrum Range -WL from: 190.0 nm
Spectrum Range -WL to: 640.0 nm
Spectrum Step: 2.0 nm
Spectrum Store: All
Stoptime:
Stoptime Mode: At pump/injector
Posttime:
Posttime Mode: Off

Signals

Signal table

	Use Sig.	Signal	Wavelength	Bandwidth	Use Ref.	Ref Wavel.	Ref Bandw.
1	Yes	Signal A	260.0 nm	4.0 nm	Yes	360.0 nm	100.0 nm
2	Yes	Signal B	300.0 nm	4.0 nm	Yes	360.0 nm	100.0 nm
3	Yes	Signal C	345.0 nm	4.0 nm	Yes	360.0 nm	100.0 nm
4	Yes	Signal D	254.0 nm	4.0 nm	Yes	360.0 nm	100.0 nm
5	No	Signal E					
6	No	Signal F					
7	No	Signal G					
8	No	Signal H					

Qtd 40: 8,12-Dibenzoyl-12,12-epoxydodecanamide C18 H34 O3	12,870	222,240.0	17276	8,12-Dibenzoyl-12,12-epoxydodecanamide	C18 H34 O3			C18 H34 O3	C18 H34 O3	-0.96
Qtd 40: 8,12-Dibenzoyl-12,12-epoxydodecanamide C18 H34 O3	15,411	222,240.0		8,12-Dibenzoyl-12,12-epoxydodecanamide	C18 H34 O3			C18 H34 O3	C18 H34 O3	-0.82
Qtd 40: Triethylolacrylate C20 H28 O4	13,512	288,192.0		Triethylolacrylate	C21 H28 N2 O3			C21 H28 N2 O3	C21 H28 N2 O3	6.81
Qtd 31: 8,12-Dibenzoyl-12,12-epoxydodecanamide C18 H34 O3	12,782	222,240.0		8,12-Dibenzoyl-12,12-epoxydodecanamide	C18 H34 O3			C18 H34 O3	C18 H34 O3	-0.82
Qtd 33: 1,20-Dibenzoyl-20-oxadecanamide acid C28 H40 O4	13,674	288,228.0		1,20-Dibenzoyl-20-oxadecanamide acid	C18 H34 O3			C18 H34 O3	C18 H34 O3	-0.82
Qtd 33: Hexaglycolol A C20 H38 O4	16,074	375,288.0		Hexaglycolol A	C21 H38 O4			C21 H38 O4	C21 H38 O4	-0.2
Compound 14										
Qtd 33: Hexaglycolol B C21 H38 O4	17,222	375,288.0		Hexaglycolol B	C21 H38 O4			C21 H38 O4	C21 H38 O4	-0.63
Qtd 33: Hexaglycolol B C21 H38 O4	17,222	375,288.0		Hexaglycolol B	C21 H38 O4			C21 H38 O4	C21 H38 O4	-0.39
Qtd 8: C18 H35 N O	17.5	281,272.0	1128		C18 H35 N O	281,272.0	-0.22	C18 H35 N O	C18 H35 N O	
Qtd 37: 4,8-Dioxa-1,6-dioxepan-2-ylidene-2-oxo-1,3-dioxane-5-carboxylic acid C22 H32 O7	17,346	354,384.0	7628	4,8-Dioxa-1,6-dioxepan-2-ylidene-2-oxo-1,3-dioxane-5-carboxylic acid	C21 H32 O7			C21 H32 O7	C21 H32 O7	-0.57
Qtd 38: 1,20-Dibenzoyl-20-oxadecanamide C28 H40 O3	17,902	477,288.0	3147	1,20-Dibenzoyl-20-oxadecanamide	C21 H34 N O2			C21 H34 N O2	C21 H34 N O2	-0.51
Qtd 39: Adiponitrile C12 H18 N2	17,912	332,124.0	2120	Adiponitrile	C21 H42 N2 O4			C21 H42 N2 O4	C21 H42 N2 O4	-12.86
Qtd 41: 4-Hydroxy-2,2,6,6-tetramethyl-1,3-dioxane-5-carboxylic acid C16 H28 O4	18,098	282,288.0	2404	4-Hydroxy-2,2,6,6-tetramethyl-1,3-dioxane-5-carboxylic acid	C18 H28 O3			C18 H28 O3	C18 H28 O3	3.52
Qtd 45: 2,2,6,6-Tetramethyl-1,3-dioxane-5-carboxylic acid C16 H28 O4	18,020	316,624.0		2,2,6,6-Tetramethyl-1,3-dioxane-5-carboxylic acid	C18 H28 O3			C18 H28 O3	C18 H28 O3	-0.28
Compound 63										
Qtd 63: Hexaglycolol C C21 H38 O4	18,412	382,488.0		Hexaglycolol C	C21 H38 O4			C21 H38 O4	C21 H38 O4	-0.27
Qtd 63: Dipropylamine A C12 H22 N	18,412	344,288.0		Dipropylamine A	C21 H32 O4			C21 H32 O4	C21 H32 O4	-0.42
Qtd 64: 2,2,6,6-Tetramethyl-1,3-dioxane-5-carboxylic acid C16 H28 O4	18,322	386,228.0		2,2,6,6-Tetramethyl-1,3-dioxane-5-carboxylic acid	C18 H22 O3			C18 H22 O3	C18 H22 O3	-0.4
Qtd 65: Hexadecane C16 H34	18,474	322,124.0	3880	Hexadecane	C21 H28 N4 O2			C21 H28 N4 O2	C21 H28 N4 O2	14.22
Qtd 66: 2,2,6,6-Tetramethyl-1,3-dioxane-5-carboxylic acid C16 H28 O4	18,420	382,228.0	4620	2,2,6,6-Tetramethyl-1,3-dioxane-5-carboxylic acid	C21 H42 N2 O4			C21 H42 N2 O4	C21 H42 N2 O4	12.72
Qtd 67: 1,20-Dibenzoyl-20-oxadecanamide C28 H40 O3	18,698	432,288.0		1,20-Dibenzoyl-20-oxadecanamide	C21 H44 N O2			C21 H44 N O2	C21 H44 N O2	-0.42
Qtd 68: Hexadecane C16 H34	18,122	322,124.0	4280	Hexadecane	C21 H28 N4 O2			C21 H28 N4 O2	C21 H28 N4 O2	14.26
Qtd 7: 1-Hydroxy-2,2,6,6-tetramethyl-1,3-dioxane-5-carboxylic acid C16 H28 O4	18,222	294,228.0	2288	1-Hydroxy-2,2,6,6-tetramethyl-1,3-dioxane-5-carboxylic acid	C18 H28 O3			C18 H28 O3	C18 H28 O3	-0.21
Qtd 69: Adiponitrile C12 H18 N2	18,712	366,288.0		Adiponitrile	C21 H44 O4			C21 H44 O4	C21 H44 O4	-0.52
Qtd 70: 1,20-Dibenzoyl-20-oxadecanamide C28 H40 O3	18,920	382,228.0	3188	1,20-Dibenzoyl-20-oxadecanamide	C21 H46 N O2			C21 H46 N O2	C21 H46 N O2	-0.87
Qtd 71: 2-(2,4,6-Trimethyl-1,3,5-triazin-2-ylidene)-2-oxo-1,3-dioxane-5-carboxylic acid C22 H32 O7	18,922	389,288.0		2-(2,4,6-Trimethyl-1,3,5-triazin-2-ylidene)-2-oxo-1,3-dioxane-5-carboxylic acid	C21 H28 N4 O2			C21 H28 N4 O2	C21 H28 N4 O2	2.61
Qtd 72: 1,20-Dibenzoyl-20-oxadecanamide C28 H40 O3	18,928	382,228.0		1,20-Dibenzoyl-20-oxadecanamide	C21 H46 N O2			C21 H46 N O2	C21 H46 N O2	-0.86
Qtd 73: Adiponitrile C12 H18 N2	18,228	386,288.0		Adiponitrile	C21 H44 O4			C21 H44 O4	C21 H44 O4	0.24
Qtd 74: 1,3-Dioxane-5-carboxylic acid C16 H28 O4										
Qtd 75: 2,2,6,6-Tetramethyl-1,3-dioxane-5-carboxylic acid C16 H28 O4	18,222	422,228.0		2,2,6,6-Tetramethyl-1,3-dioxane-5-carboxylic acid	C21 H28 N4			C21 H28 N4	C21 H28 N4	3.28
Qtd 76: 2-(2,4,6-Trimethyl-1,3,5-triazin-2-ylidene)-2-oxo-1,3-dioxane-5-carboxylic acid C22 H32 O7	18,224	389,288.0		2-(2,4,6-Trimethyl-1,3,5-triazin-2-ylidene)-2-oxo-1,3-dioxane-5-carboxylic acid	C21 H28 N4 O2			C21 H28 N4 O2	C21 H28 N4 O2	1.28
Qtd 77: Isocyanic acid C17 H12 O2	18,228	328,228.0	2960	Isocyanic acid	C21 H28 O3			C21 H28 O3	C21 H28 O3	-12.96
Qtd 78: 2,2,6,6-Tetramethyl-1,3-dioxane-5-carboxylic acid C16 H28 O4	18,222	422,228.0		2,2,6,6-Tetramethyl-1,3-dioxane-5-carboxylic acid	C21 H28 N4			C21 H28 N4	C21 H28 N4	4.12
Qtd 79: Hexaglycolol B C21 H38 O4	18,228	472,228.0		Hexaglycolol B	C21 H40 O3			C21 H40 O3	C21 H40 O3	-0.92
Qtd 80: 2,2,6,6-Tetramethyl-1,3-dioxane-5-carboxylic acid C16 H28 O4	18,222	422,228.0	4820	2,2,6,6-Tetramethyl-1,3-dioxane-5-carboxylic acid	C21 H28 N4			C21 H28 N4	C21 H28 N4	3.58
Qtd 81: 1,3-Dioxane-5-carboxylic acid C16 H28 O4	18,782	442,228.0	4120	1,3-Dioxane-5-carboxylic acid	C21 H40 O3			C21 H40 O3	C21 H40 O3	-0.2
Qtd 82: 4-Ethoxy-2,2,6,6-tetramethyl-1,3-dioxane-5-carboxylic acid C20 H38 O5	18,928	428,228.0		4-Ethoxy-2,2,6,6-tetramethyl-1,3-dioxane-5-carboxylic acid	C21 H42 O3			C21 H42 O3	C21 H42 O3	-0.24
Qtd 83: 2-Hydroxy-1,3-dioxane-5-carboxylic acid C16 H28 O5	20,028	272,228.0		2-Hydroxy-1,3-dioxane-5-carboxylic acid	C21 H42 O3			C21 H42 O3	C21 H42 O3	-0.42
Qtd 8: C18 H35 N O	20,222	378,228.0	388		C21 H44 O4	378,228.0	0.15	C21 H44 O4	C21 H44 O4	
Compound 84										
Qtd 85: Carbazole C12 H8 N2	22,228	282,228.0		Carbazole	C21 H28 N2 O			C21 H28 N2 O	C21 H28 N2 O	-0.42
Compound 85										
Qtd 87: Dicyclopentadiene C10 H12	22,228	272,172.0		Dicyclopentadiene	C21 H22 N2 O			C21 H22 N2 O	C21 H22 N2 O	7.24
Qtd 13: C18 H35 N O	22,246	284,246.0	2388		C21 H22 O	284,246.0	-0.76	C21 H22 O	C21 H22 O	
Qtd 88: 4-Ethoxy-2,2,6,6-tetramethyl-1,3-dioxane-5-carboxylic acid C20 H38 O5	22,278	428,228.0		4-Ethoxy-2,2,6,6-tetramethyl-1,3-dioxane-5-carboxylic acid	C21 H42 O3			C21 H42 O3	C21 H42 O3	-0.22
Qtd 89: 2-Hydroxy-1,3-dioxane-5-carboxylic acid C16 H28 O5	22,272	428,228.0		2-Hydroxy-1,3-dioxane-5-carboxylic acid	C21 H40 O4			C21 H40 O4	C21 H40 O4	-0.22
Qtd 90: 2-Hydroxy-1,3-dioxane-5-carboxylic acid C16 H28 O5	22,28	272,228.0		2-Hydroxy-1,3-dioxane-5-carboxylic acid	C21 H42 O3			C21 H42 O3	C21 H42 O3	-0.22

APPENDIX V

HR-LCMS profiling of compounds in UPT in negative ESI mode

Compound 48	6.833		11594						
Compound 49	7.282		10584						
Compound 50	7.34		10248						
Compound 51	7.376		15233						
Compound 52	7.431		10792						
Cpd 53: 7,7-Dihydroxy-4-methoxy-4-propylflavone; C21 H24 O4	7.576	348.1389		7,7-Dihydroxy-4-methoxy-4-propylflavone	C21 H24 O4	C21 H24 O4	C21 H24 O4	-4.26	1
Compound 54	7.657		21405						
Compound 55	7.848		23994						
Compound 56	7.91								
Cpd 57: Genistein; C21 H20 O5	8.039	437.2188		Genistein	C21 H20 O5	C21 H20 O5	C21 H20 O5	-7.45	1
Compound 58	8.139		10064						
Compound 59	8.227		11560						
Compound 60	8.526		13724						
Cpd 61: 4-O-alpha-D-Galactopyranosylcatechin B; C22 H22 H O8	8.761	427.1361		4-O-alpha-D-Galactopyranosylcatechin B	C22 H22 H O8	C22 H22 H O8	C22 H22 H O8	3.44	8
Compound 62	8.852		11734						
Compound 63	8.908		10994						
Cpd 64: 2-Indoxy-4-(1H)-quinolone; C20 H19 N O	8.91	384.121		2-Indoxy-4-(1H)-quinolone	C20 H19 N O	C20 H19 N O	C20 H19 N O	13.02	1
Cpd 65: 6'-Acetylgenistein; C20 H18 H O5	8.975	446.1407		6'-Acetylgenistein	C20 H18 H O5	C20 H18 H O5	C20 H18 H O5	-7.34	2
Cpd 66: Istinone C; C19 H17 O5	9.173	465.366		Istinone C	C19 H17 O5	C19 H17 O5	C19 H17 O5	-12.4	3
Compound 67	9.808		14723						
Compound 68	10.072								
Cpd 69: Istinone A; C19 H17 O5	10.074	465.376		Istinone A	C19 H17 O5	C19 H17 O5	C19 H17 O5	-11.08	1
Compound 70	10.908								
Cpd 71: C18-18; C17 H17 N O14	11.138	778.3023		C18-18	C17 H17 N O14	C17 H17 N O14	C17 H17 N O14	-4.03	1
Cpd 72: Isoquiginate; C20 H21 H O3	11.161	323.1557	11982	Isoquiginate	C20 H21 H O3	C20 H21 H O3	C20 H21 H O3	-11.12	4
Compound 73	11.222		17664						
Compound 74	11.482		29902						
Compound 75	11.581		21918						
Cpd 76: Isoquiginate; C20 H21 H O3	11.694	323.1554	11762	Isoquiginate	C20 H21 H O3	C20 H21 H O3	C20 H21 H O3	-10.09	4
Compound 77	11.894		13425						
Compound 78	12.221								
Compound 79	12.276								
Compound 80	12.467								
Compound 81	12.844								
Compound 82	14.28								
Compound 83	14.379								
Compound 84	14.617								
Compound 85	15.128		11872						
Cpd 86: 5-Octadec-3-yl-pyrroline; C19 H34 O2	15.23	427.2139		5-Octadec-3-yl-pyrroline	C19 H34 O2	C19 H34 O2	C19 H34 O2	3.02	7
Compound 87	15.746								
Cpd 88: Hattoriol; C19 H26 O3	15.774	414.1953		Hattoriol	C19 H26 O3	C19 H26 O3	C19 H26 O3	-3.27	3
Cpd 89: 6'-O-259H; C21 H20 O5	15.783	437.2254		6'-O-259H	C21 H20 O5	C21 H20 O5	C21 H20 O5	-9.1	1
Cpd 90: Octanoylgucuronide; C14 H24 O8	18.238	323.1481	32518	Octanoylgucuronide	C14 H24 O8	C14 H24 O8	C14 H24 O8	6.18	3
Cpd 91: Lycopodol(242/523); C19 H26 H O7 P	18.33	365.4089		Lycopodol(242/523)	C19 H26 H O7 P	C19 H26 H O7 P	C19 H26 H O7 P	3.21	1
Cpd 92: 2,3-O-isobutyl-4-xylyloxy; C15 H16 O	18.58	234.2449		2,3-O-isobutyl-4-xylyloxy	C15 H16 O	C15 H16 O	C15 H16 O	14.08	1
Compound 93	20.198								
Cpd 94: 4-Dimethylbutyltinol; C15 H18 H O3	20.5	277.1474		4-Dimethylbutyltinol	C15 H18 H O3	C15 H18 H O3	C15 H18 H O3	1.28	3
Compound 95	21.708								
Compound 96	22.572								
Compound 97	24.148								
Cpd 98: Diphenol A; C14 H18 H O6	24.276	576.2752	20178	Diphenol A	C14 H18 H O6	C14 H18 H O6	C14 H18 H O6	-3.86	4
Compound 99	24.33								
Compound 100	24.817								

APPENDIX VI

HR-LCMS profiling of compounds in UPS in negative ESI mode

Compound 1	2.10	413.07	Compound 1	C ₁₀ H ₁₃ O ₂			C ₁₀ H ₁₃ O ₂	C ₁₀ H ₁₃ O ₂	-1.00
Compound 2	2.20	413.07	Compound 2	C ₁₀ H ₁₃ O ₂			C ₁₀ H ₁₃ O ₂	C ₁₀ H ₁₃ O ₂	0.00
Compound 3	2.30	413.07	Compound 3	C ₁₀ H ₁₃ O ₂			C ₁₀ H ₁₃ O ₂	C ₁₀ H ₁₃ O ₂	-0.00
Compound 4	2.40	413.07	Compound 4	C ₁₀ H ₁₃ O ₂			C ₁₀ H ₁₃ O ₂	C ₁₀ H ₁₃ O ₂	-0.00
Compound 5	2.50	413.07	Compound 5	C ₁₀ H ₁₃ O ₂			C ₁₀ H ₁₃ O ₂	C ₁₀ H ₁₃ O ₂	0.00
Compound 6	2.60	413.07	Compound 6	C ₁₀ H ₁₃ O ₂			C ₁₀ H ₁₃ O ₂	C ₁₀ H ₁₃ O ₂	0.00
Compound 7	2.70	413.07	Compound 7	C ₁₀ H ₁₃ O ₂			C ₁₀ H ₁₃ O ₂	C ₁₀ H ₁₃ O ₂	0.00
Compound 8	2.80	413.07	Compound 8	C ₁₀ H ₁₃ O ₂			C ₁₀ H ₁₃ O ₂	C ₁₀ H ₁₃ O ₂	0.00
Compound 9	2.90	413.07	Compound 9	C ₁₀ H ₁₃ O ₂			C ₁₀ H ₁₃ O ₂	C ₁₀ H ₁₃ O ₂	0.00
Compound 10	3.00	413.07	Compound 10	C ₁₀ H ₁₃ O ₂			C ₁₀ H ₁₃ O ₂	C ₁₀ H ₁₃ O ₂	0.00

Data File: T1-014-CD-001 Sample Name: T1-014-CD-001
 Sample Type: Sample Method: P1-01
 Instrument Name: QTOF User Name:
 Acq Method: run000001_001_000001.ms Acquired Time: 11/17/2014 1:30:00 AM
 SRM Calibration Status: Success DI Method: Default.ms
 Comments:

Sample Group:
 Acquisition ID: 00000001000000000000
 VialIndex: 0-7000.00 (0-1000.0)

Compound Table

Compound Label	RT	Mass	Abund	Name	Formula	MS Formula	DB Formula	DB RT (min)	DB MS
Compound 1	1.28								
Compound 2	1.40								
Compound 3	1.53								
Compound 4	1.75		8038						
Compound 5	1.88		3670						
Compound 6	3.10								
Q17: 1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	3.40	272.2611	4380	1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	C ₁₁ H ₁₀ N ₄ O ₃	C ₁₁ H ₁₀ N ₄ O ₃	C ₁₁ H ₁₀ N ₄ O ₃	6.50	1
Q18: 1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	3.70	272.2611	3408	1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	C ₁₁ H ₁₀ N ₄ O ₃	C ₁₁ H ₁₀ N ₄ O ₃	C ₁₁ H ₁₀ N ₄ O ₃	7.40	1
Q19: 1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	4.75	324.1470	2580	1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	C ₁₂ H ₁₂ N ₄ O ₃	C ₁₂ H ₁₂ N ₄ O ₃	C ₁₂ H ₁₂ N ₄ O ₃	-0.1	1
Q20: 1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	5.10	344.1671	1800	1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	C ₁₃ H ₁₄ N ₄ O ₃	C ₁₃ H ₁₄ N ₄ O ₃	C ₁₃ H ₁₄ N ₄ O ₃	-7.40	1
Q21: 1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	5.75	406.2153	1000	1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	C ₁₄ H ₁₆ N ₄ O ₃	C ₁₄ H ₁₆ N ₄ O ₃	C ₁₄ H ₁₆ N ₄ O ₃	-0.50	1
Q22: 1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	6.40	468.2635	500	1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	C ₁₅ H ₁₈ N ₄ O ₃	C ₁₅ H ₁₈ N ₄ O ₃	C ₁₅ H ₁₈ N ₄ O ₃	-0.10	1
Compound 23	8.70								
Q23: 1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	8.83	468.2635	1000	1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	C ₁₅ H ₁₈ N ₄ O ₃	C ₁₅ H ₁₈ N ₄ O ₃	C ₁₅ H ₁₈ N ₄ O ₃	-6.10	1
Q24: 1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	8.97	488.2836	1200	1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	C ₁₆ H ₂₀ N ₄ O ₃	C ₁₆ H ₂₀ N ₄ O ₃	C ₁₆ H ₂₀ N ₄ O ₃	-0.10	1
Q25: 1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	9.10	508.3037	1000	1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	C ₁₇ H ₂₂ N ₄ O ₃	C ₁₇ H ₂₂ N ₄ O ₃	C ₁₇ H ₂₂ N ₄ O ₃	8.00	1
Q26: 1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	9.23	528.3238	1000	1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	C ₁₈ H ₂₄ N ₄ O ₃	C ₁₈ H ₂₄ N ₄ O ₃	C ₁₈ H ₂₄ N ₄ O ₃	-0.10	1
Q27: 1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	9.4	548.3439	1000	1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	C ₁₉ H ₂₆ N ₄ O ₃	C ₁₉ H ₂₆ N ₄ O ₃	C ₁₉ H ₂₆ N ₄ O ₃	7.30	1
Q28: 1,7-bis(methylamino)-1,7-dimethylheptane-2,6-dione	9.56	342.1371		1,7-bis(methylamino)-1,7-dimethylheptane-2,6-dione	C ₁₂ H ₂₂ O ₄	C ₁₂ H ₂₂ O ₄	C ₁₂ H ₂₂ O ₄	-8.1	1
Compound 29	6.67								
Compound 30	6.94								
Q29: 1,7-bis(methylamino)-1,7-dimethylheptane-2,6-dione	7.00	342.1371		1,7-bis(methylamino)-1,7-dimethylheptane-2,6-dione	C ₁₂ H ₂₂ O ₄	C ₁₂ H ₂₂ O ₄	C ₁₂ H ₂₂ O ₄	-7.00	1
Compound 31	7.03								
Q30: 1,7-bis(methylamino)-1,7-dimethylheptane-2,6-dione	7.03	402.2071	5120	1,7-bis(methylamino)-1,7-dimethylheptane-2,6-dione	C ₁₃ H ₂₄ O ₄	C ₁₃ H ₂₄ O ₄	C ₁₃ H ₂₄ O ₄	-6.30	1
Q31: 1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	7.08	594.4611		1-oxo-1H-pyridin-2-ylidene-1H-imidazole-5-carboxamide	C ₂₅ H ₃₀ N ₄ O ₃	C ₂₅ H ₃₀ N ₄ O ₃	C ₂₅ H ₃₀ N ₄ O ₃	-0.1	1
Q32: Cyclopentane	7.66	411.1103	2405	Cyclopentane	C ₅ H ₁₀	C ₅ H ₁₀	C ₅ H ₁₀	8.6	1
Q33: Hexamethylenetetramine	7.70	345.1719		Hexamethylenetetramine	C ₆ H ₁₂ N ₄	C ₆ H ₁₂ N ₄	C ₆ H ₁₂ N ₄	0.70	1
Q34: 4-(methylamino)-2,4-dimethylheptane-2,6-dione	7.88	624.1101		4-(methylamino)-2,4-dimethylheptane-2,6-dione	C ₁₁ H ₂₂ O ₄	C ₁₁ H ₂₂ O ₄	C ₁₁ H ₂₂ O ₄	-14.80	1
Compound 32	7.94		4110						
Q35: 1,1,1,3,3,3-hexafluoro-4,4,4-trifluoro-2,2,2-trimethylbutane	8.00	318.1813	1341	1,1,1,3,3,3-hexafluoro-4,4,4-trifluoro-2,2,2-trimethylbutane	C ₈ H ₁₈ F ₉	C ₈ H ₁₈ F ₉	C ₈ H ₁₈ F ₉	-7.20	1
Q36: 1,1,1,3,3,3-hexafluoro-4,4,4-trifluoro-2,2,2-trimethylbutane	8.17	318.1813	1000	1,1,1,3,3,3-hexafluoro-4,4,4-trifluoro-2,2,2-trimethylbutane	C ₈ H ₁₈ F ₉	C ₈ H ₁₈ F ₉	C ₈ H ₁₈ F ₉	6.70	1
Q37: 1,1,1,3,3,3-hexafluoro-4,4,4-trifluoro-2,2,2-trimethylbutane	8.18	338.2014		1,1,1,3,3,3-hexafluoro-4,4,4-trifluoro-2,2,2-trimethylbutane	C ₉ H ₂₀ F ₉	C ₉ H ₂₀ F ₉	C ₉ H ₂₀ F ₉	0.00	1
Compound 33	8.18		1000						
Compound 34	8.46								
Compound 35	8.64		1000						
Q38: 1,1,1,3,3,3-hexafluoro-4,4,4-trifluoro-2,2,2-trimethylbutane	8.69	440.1108		1,1,1,3,3,3-hexafluoro-4,4,4-trifluoro-2,2,2-trimethylbutane	C ₁₀ H ₂₂ F ₉	C ₁₀ H ₂₂ F ₉	C ₁₀ H ₂₂ F ₉	0.00	1
Q39: 1-Carbonyl-1,1,1,3,3,3-hexafluoro-4,4,4-trifluoro-2,2,2-trimethylbutane	8.88	318.1813	1000	1-Carbonyl-1,1,1,3,3,3-hexafluoro-4,4,4-trifluoro-2,2,2-trimethylbutane	C ₉ H ₁₈ F ₉ O	C ₉ H ₁₈ F ₉ O	C ₉ H ₁₈ F ₉ O	-6.00	1
Q40: 1-Carbonyl-1,1,1,3,3,3-hexafluoro-4,4,4-trifluoro-2,2,2-trimethylbutane	8.90	340.2014	1000	1-Carbonyl-1,1,1,3,3,3-hexafluoro-4,4,4-trifluoro-2,2,2-trimethylbutane	C ₁₀ H ₂₀ F ₉ O	C ₁₀ H ₂₀ F ₉ O	C ₁₀ H ₂₀ F ₉ O	-6.10	1
Q41: 1-Carbonyl-1,1,1,3,3,3-hexafluoro-4,4,4-trifluoro-2,2,2-trimethylbutane	9.09	340.2014	1000	1-Carbonyl-1,1,1,3,3,3-hexafluoro-4,4,4-trifluoro-2,2,2-trimethylbutane	C ₁₀ H ₂₀ F ₉ O	C ₁₀ H ₂₀ F ₉ O	C ₁₀ H ₂₀ F ₉ O	-5.00	1
Compound 40	9.18		1000						
Q42: 1-Carbonyl-1,1,1,3,3,3-hexafluoro-4,4,4-trifluoro-2,2,2-trimethylbutane	9.23	461.2103		1-Carbonyl-1,1,1,3,3,3-hexafluoro-4,4,4-trifluoro-2,2,2-trimethylbutane	C ₁₁ H ₂₂ F ₉ O	C ₁₁ H ₂₂ F ₉ O	C ₁₁ H ₂₂ F ₉ O	-0.00	1
Compound 44	9.44								
Compound 45	9.56		1000						

APPENDIX VII

HR-LCMS profiling of compounds in UPL in negative ESI mode

Peak ID	Retention Time (min)	Abundance	Compound Name	MS/MS Scan 1	MS/MS Scan 2	MS/MS Scan 3	MS/MS Scan 4	MS/MS Scan 5
Qp1 01	1.26	154,179	Isobutanol, 2-(3-oxopropyl)-	C8 112 018	C8 112 011	C8 112 018		
Qp1 04	1.76	154,187	2-Phenylacetic acid	C8 101 041AP	C8 101 041P	C8 101 041P		
Qp1 05	2.62	118,188	Acetic acid	C4 108 02 06 0 5	C4 108 02 06 0 5	C4 108 02 06 0 5		
Qp1 06	3.24	172,194	2-Hydroxybutanoic acid	C4 140 02 5	C4 140 02 5	C4 140 02 5		
Qp1 07	4.26	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 08	4.76	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 09	5.26	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 10	5.76	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 11	6.26	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 12	6.76	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 13	7.26	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 14	7.76	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 15	8.26	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 16	8.76	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 17	9.26	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 18	9.76	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 19	10.26	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 20	10.76	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 21	11.26	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 22	11.76	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 23	12.26	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 24	12.76	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 25	13.26	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 26	13.76	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 27	14.26	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 28	14.76	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 29	15.26	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 30	15.76	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 31	16.26	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 32	16.76	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 33	17.26	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 34	17.76	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 35	18.26	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 36	18.76	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 37	19.26	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 38	19.76	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 39	20.26	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 40	20.76	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 41	21.26	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 42	21.76	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 43	22.26	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 44	22.76	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 45	23.26	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 46	23.76	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 47	24.26	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 48	24.76	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 49	25.26	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		
Qp1 50	25.76	164,195	Acetic acid	C4 108 02 5	C4 108 02 5	C4 108 02 5		

APPENDIX VIII PREPARATIONS

Total Phenolic content

Preparation of 10% Sodium Carbonate

10% Sodium carbonate was prepared by dissolving the 10 gram of sodium carbonate in

100 ml of distilled water.

Preparation of 10% Folin-Ciocalteu Reagent

10ml FCR was diluted in 80ml of distilled water.

Preparation of Standard Gallic Acid Solution for TPC determination

1000 $\mu\text{g/ml}$ of stock solution was prepared by dissolving like 25mg of gallic acid in 25ml

of methanol (i.e., 1mg of gallic acid in 1 ml of solvent). The stock solution was diluted to

archive final concentration of (5, 10, 20, 40, 80, 100) $\mu\text{g/mL}$ (i.e.then prepared using a two-fold

dilution process). It is worth noting that the solution utilize for testing were freshly prepared.

Total Flavonoids content

Preparation of Standard Quercetin Solution

Stock solution of Quercetin was made in methanol by dissolving 10 mg of rutin in 10 ml

methanol (i.e., 1mg/ml). The final concentrations were adjusted in

(10, 20, 40,80) $\mu\text{g/mL}$ by doing two-fold dilution from the stock solution.

Preparation of plant extracts

The working solution of plant extract i.e., 250 µg/mL were prepared by diluting from the 2.5mg/ml of stock solution.

Preparation of Aluminum 10% Trichloride Hexa-Hydrate & 1M potassium acetate

10% Aluminum trichloride hexahydrate ($\text{AlCl}_3 \cdot 6 \text{H}_2\text{O}$) was prepared by dissolving 10 gram of solid mass into the 100 ml of distilled water. Similarly, 1M potassium acetate was prepared by dissolving 9.815g pot. Acetate into 100ml distilled water.

Phosphomolybdenum assay

Preparation of 0.6 M sulfuric acid

330.30 µl of concentrated sulphuric acid was poured into 100 ml of volumetric flask in distilled water to get 0.6 M of dilute Sulphuric acid solution.

Preparation of 28 mM sodium phosphate

0.498 mg of sodium phosphate dry weight was dissolved in distilled water to obtain 28 mM sodium phosphate solution.

Preparation of 4 mM ammonium molybdate

4 mM ammonium molybdate solution was prepared by dissolving 0.494 mg of ammonium molybdate solution into the 100 ml of volumetric flask using distilled water as solvent.

At last mix them (0.6 M sulfuric acid, 28 mM sodium phosphate and 4 mM ammonium molybdate) in equal portion to obtained Phosphomolybdenum reagents.

Preparation of Ascorbic acid

Stock solution of Ascorbic acid was made in methanol by dissolving 10 mg of ascorbic acid in 10ml methanol (i.e., 1mg/ml). The final concentrations were adjusted in (5, 10, 20, 40, 80, 100) µg/mL by doing two-fold dilution from the stock solution

DPPH free radical scavenging assay

Preparation of 0.004% DPPH solution

0.004% DPPH solution was mutinously prepared by dissolving 20mg of DPPH in 500 ml of volumetric flask, which was properly shielded with aluminum foil (i.e., DPPH is light sensitive).

Preparation of Butylated Hydroxy Anisole (BHA) solution

Weight out 10 mg of BHA in the 10 ml of methanol solution to make 1mg/ml of stock solution. Which was further diluted to prepare (1, 2, 3, 4, 5, 10, 20, 40, 80) µg/mL working solutions.

Preparation of plant extracts solution

Dissolve 25mg in 25ml to make 1mg/ml plant extract using vortex mixer. Load them to 10-time higher concentration than desire concentration for 10 ml volumetric flask.

Preparation of Media

45.6g of Mueller Hinton Agar was dissolved completely in 1200ml Distilled water and stirred with autoclaved glass rod. Then the mixture was transferred equally into three 500ml conical flasks. The agar mixture was then autoclaved at 121⁰C for 20 minutes. The pressure of autoclave was 15atm. Thus, prepared agar solution was left for few minutes under cooling.



वन तथा वनसंरक्षण विभाग
वनस्पति विभाग
राष्ट्रिय हर्वेरियम तथा वनस्पति प्रयोगशाला
सोदावरी, काठमाडौं



पत्र नम्बरा: ०८०/०८९
च.नं. १२५

सोदावरी, काठमाडौं

मिति २०८०/०६/२६

विषय: नमूना पहिचान सम्बन्धमा।

श्री वीरन्द्र बहुगुठी क्याम्पस,
भरतपुर, चितवन।

प्रस्तुत विषयमा तहाँ क्याम्पसको ए.सं. ०८०/०८९, च.नं. ६४४, मिति २०८०/०६/२९ को पत्र साथ वनस्पतिको नमूना प्राप्त भई व्यहोरा अबगत भयो। पत्र मार्फत ल्याइएको नमूनाको पहिचान गरी अतिरिक्त विशेषज्ञको प्रतिवेदन (पाना १) सहै पत्रसाथ संलग्न गरी पठाइएको व्यहोरा अनुरोध छ।


००१३/२६
सुभाष खत्री

वरिष्ठ अनुसन्धान अधिकृत
(१६३६८९)

कार्यालय प्रमुख



प्राविधिक विशेषज्ञको प्रतिवेदन

१. नमूना परिक्षण गर्ने पठाउने व्यक्ति/विकाय:- श्री अजय महता (वि.पि. इतां नं. ५-२-१९-५/७७-२०११)
- १(क) विद्यार्थीहरूको नाम:- बनस्पतिको नमूना धान-१
२. प्राप्त नमूनाको विवरण:- २०८०/०६/२५
३. यस कार्यालयमा प्राप्त मिति:-
४. परिक्षणका आधारहरू:- (क) हर्बेरियममा भएका नमूनाहरू संगको तुलनात्मक अध्ययन
(ख) सुन्दर्भ सान्नीहरूको अध्ययन ।
५. पहिचान प्रतिवेदन:- प्राप्त नमूनाको Morphological अध्ययन तथा यस राष्ट्रिय हर्बेरियम तथा बनस्पति प्रयोगशालाको हर्बेरियममा राखिएका नमूनाहरू संगको तुलनात्मक अध्ययन गर्दा उक्त नमूनाहरू निम्नागुसार भएको पहिचान हुन गएको ।

S.N	Scientific Name	Family	Remarks
1	<i>Colocasia esculenta</i> (L.) Schott	Araceae	

६. परिक्षण गर्ने अधिकारी:-


रिता बत्ती
अनुसन्धान अधिकृत
(१९८२००)

PLAGARISM REPORT



त्रिभुवन विश्वविद्यालय
TRIBHUVAN UNIVERSITY

☎ : 056 { 493253
493689
493159

वीरेन्द्र बहुमुखी क्याम्पस

BIRENDRA MULTIPLE CAMPUS

भरतपुर, चितवन
Bharatpur, Chitwan

पत्रसंख्या :
च. नं. (Ref.) :

मिति : २०८१।१०।२०
Date :

जो जस सँग सम्बन्धित छ ।

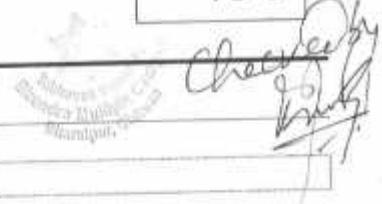
तपाईं **AJAYAMAHA TO** ले यस पुस्तकालयमा प्लेजारिजम परीक्षण गर्नका लागि हार्डकपी र सफ्टकपीको विषय वस्तुमा कुनै फरक छैन भनी स्वघोषणा गरी पेनड्राइभ/ईमेल मार्फत विभागमा पेश गर्नुभएकोले विभागबाट इमेल मार्फत प्राप्त सफ्टकपी **NUTRIENT ELEMENTS AND HR-LCMS ANALYSIS OF SECONDARY METABOLITES PRESENT IN METHANOLIC EXTRACTS OF COCOYAM *Colocasia esculenta (L.) schott*** शिर्षकको **M.Sc. in CHEMISTRY** तहको उपाधिका लागि तयार गरिएको Dissertation / Thesis मा प्लेजारिजम परीक्षण पछिको समानता सूची (Similarity Index) १०(दश) प्रतिशत रहेको व्यहोरा प्रमाणित गरिन्छ ।


महेन्द्रप्रसाद अधिकारी
पुस्तकालय प्रमुख

NUTRIENT ELEMENTS AND HR-LCMS ANALYSIS OF SECON...

By: AJAYA MAHATO
As of: Feb 2, 2025 4:17:58 PM
9,033 words - 70 matches - 55 sources

Similarity Index
10%



Mode: Summary Report

sources:

58 words / 1% - Internet from 05-Jan-2023 12:00AM
www.researchgate.net

52 words / 1% - Internet from 08-Feb-2023 12:00AM
www.researchgate.net

48 words / 1% - Internet from 21-May-2022 12:00AM
www.researchgate.net

73 words / 1% - Internet from 15-Mar-2023 12:00AM
physc.sgluniversity.ac.in

14 words / < 1% match - Internet from 03-Feb-2023 12:00AM
www.researchgate.net

11 words / < 1% match - Internet from 30-Jan-2023 12:00AM
www.researchgate.net

9 words / < 1% match - Internet from 28-Jan-2023 12:00AM
www.researchgate.net

38 words / < 1% match - Internet from 18-Feb-2024 12:00AM
elibrary.tucl.edu.np

33 words / < 1% match - Internet from 18-Jan-2024 12:00AM
elibrary.tucl.edu.np

9 words / < 1% match - Internet from 02-Feb-2024 12:00AM
elibrary.tucl.edu.np

30 words / < 1% match - Internet from 03-Dec-2021 12:00AM
www.pubfacts.com